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Thin Solid Films Volume 216, Issue 1 , 28 August 1992, Pages 123-125	
doi:10.1016/0040-6090(92)90880-K	This Document ▶ Abstract
The characteristics of cyclic phosphazene and its applications for hard coatings	ActionsCited BySave as Citation AlertE-mail Article
A. Yaguchi, S.Mori, M. Kitayama, T. Onda, A. Kurahashi and H. Ando	Export Citation Add to my Quick Links

Planning and Development Department, Idemitsu Petrochemical Co. Ltd., 1-1 Marunouchi 3-Chome, Chiyoda-ku, Tokyo 100, Japan

Available online 18 September 2002.

Abstract

A new type of curable material has been developed through the synthesis of hexachlorocyclotriphosphazene and 2-hydroxyethylmethacrylate. This is a colourless transparent viscous liquid, which is cured by irradiation with UV light, electron beams and heat. This cured material has extreme hardness as well as heat, chemical and stain resistance.

Thin Solid Films

Volume 216, Issue 1, 28 August 1992, Pages 123-125

This Document

▶ Abstract

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Ref #	Hits	Search Query	DBs	Default Operator	Plurals	Time Stamp
L1	11	\$phosphorine\$.clm.	US-PGPUB	OR	OFF	2006/09/04 13:55

Interference Seurch 10/536,510

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L4	33	430/283.1,302.CCLS. AND \$BIURET\$	US-PGPUB; USPAT	OR	OFF	2006/09/04 15:29
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L8	2	7 NOT 3	US-PGPUB; USPAT	OR	OFF	2006/09/04 15:29
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S2	40	\$triazatriphosphorine\$	US-PGPUB; USPAT	OR	OFF	2006/09/04 15:21
S3	58	\$triazatriphosphorine\$	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR	OFF	2006/09/04 08:18
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S6	238	\$PHOSPHAZEN\$ AND (photoa\$ photob\$ photoc\$ photod\$ photoe\$ photof\$ photog\$ photoh\$ photoi\$ photoj\$ photok\$ photol\$ photom\$ photon\$ photoo\$ photop\$ photov\$ photor\$ photox\$ photoX\$ photoY\$ photoX\$ photoX\$ photoZ\$) AND ("430"/\$ "522"/\$).CCLS.	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR	OFF	2006/09/04 08:22

<u>Ş</u> 7	11	\$PHOSPHAZEN\$ AND (photoa\$ photob\$ photoc\$ photod\$ photoe\$ photof\$ photog\$ photoh\$ photoi\$ photoi\$ photok\$ photol\$ photom\$ photon\$ photoo\$ photop\$ photov\$ photos\$ photox\$ photox\$ photoX\$ photoX\$ photoZ\$) AND ("430"/\$ "522"/\$).CCLS. AND BIURET\$	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR	OFF	2006/09/04 10:30
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S16	1	1999-592904.NRAN.	DERWENT	OR	OFF	2006/09/04 09:18
S17	9	\$PHOSPHAZEN\$ AND (photoa\$ photob\$ photoc\$ photod\$ photoe\$ photof\$ photog\$ photoh\$ photoi\$ photoi\$ photoo\$ photop\$ photom\$ photon\$ photoo\$ photop\$ photov\$ photov\$ photox\$ photoX\$ photoZ\$) AND ("430"/\$ "522"/\$).CCLS. AND (allophan\$)	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR	OFF	2006/09/04 11:19
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S19	1	("5082717").PN.	US-PGPUB; USPAT	OR	OFF	2006/09/04 10:47
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S23	2	ep-1104790-\$.did.	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR	OFF	2006/09/04 11:02
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S26	0	S17 not S7	US-PGPUB; USPAT	OR	OFF	2006/09/04 11:19
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S28	0	\$PHOSPHAZEN\$ AND (photoa\$ photob\$ photoc\$ photod\$ photoe\$ photof\$ photog\$ photoh\$ photoi\$ photoo\$ photoh\$ photon\$ photoo\$ photop\$ photor\$ photos\$ photot\$ photov\$ photov\$ photov\$ photox\$ photoX\$ photoX\$ photoX\$ photoZ\$) AND ("430"/\$ "522"/\$).CCLS. AND (\$imidiodicarbonic\$)	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR	OFF	2006/09/04 11:20
S29	40	ua adj 306h	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR	OFF	2006/09/04 12:24
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•	S31	. 1	DE-2361041-\$.DID.	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR	OFF	2006/09/04 12:40
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9/4/06 3:33:27 PM
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L16	1	1999-592904.NRAN.	DERWENT	OR	OFF	2006/09/04 09:18
L17	9.	\$PHOSPHAZEN\$ AND (photoa\$ photob\$ photoc\$ photod\$ photoe\$ photof\$ photog\$ photoh\$ photoi\$ photoi\$ photok\$ photol\$ photom\$ photon\$ photoo\$ photop\$ photov\$ photos\$ photot\$ photox\$ phot	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR ·	OFF	2006/09/04 11:19
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L23	2	ep-1104790-\$.did.	US-PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT; IBM_TDB	OR	OFF	2006/09/04 11:02
L24	1	("6045975").PN.	US-PGPUB; USPAT	OR	OFF	2006/09/04 11:02

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                 ADISCTI Reloaded and Enhanced
NEWS 17 AUG 30 CA(SM)/CAplus(SM) Austrian patent law changes
NEWS EXPRESS
              JUNE 30 CURRENT WINDOWS VERSION IS V8.01b, CURRENT
              MACINTOSH VERSION IS V6.0c(ENG) AND V6.0Jc(JP),
              AND CURRENT DISCOVER FILE IS DATED 26 JUNE 2006.
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=> S PHOSPHAZENE

L1 492 PHOSPHAZENE

=> S L1 AND ACRYL? 121575 ACRYL?

L2 31 L1 AND ACRYL?

=> S L2 AND ENE 6304255 ENE

L3 6 L2 AND ENE

=> D ALL 1-6

L3 ANSWER 1 OF 6 REGISTRY COPYRIGHT 2006 ACS on STN

RN 727704-44-1 REGISTRY

ED Entered STN: 17 Aug 2004

CN 1,4-Benzenedicarboxylic acid, polymer with 1,2-ethanediol, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine, 1,6-hexanediol, 2-hydroxypropyl 2-propenoate, 1,1'-methylenebis[isocyanatocyclohexane] and methyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

OTHER NAMES:

CN Ethylene glycol-1,1,3,3,5,5-hexakis(acryloylethylenedioxy)cyclotripho sphazene-1,6-hexanediol-2-hydroxypropyl acrylate-methylenebis(cyclohexyl isocyanate)-methyl methacrylate-terephthalic acid copolymer

MF (C30 H42 N3 O18 P3 . C15 H22 N2 O2 . C8 H6 O4 . C6 H14 O2 . C6 H10 O3 . C5 H8 O2 . C2 H6 O2) \times

CI PMS

PCT Polyacrylic, Polyamide, Polyamide formed, Polyester, Polyester formed, Polyphosphazene, Polyphosphazene formed

SR CA

LC STN Files: 'CA, CAPLUS

DT.CA CAplus document type: Patent

RL.P Roles from patents: PREP (Preparation); PRP (Properties); USES (Uses)

Ring System Data

Elemental|Elemental| Size of |Ring System| Ring | RID
Analysis |Sequence |the Rings| Formula |Identifier|Occurrence
EA | ES | SZ | RF | RID | Count

| C6 | 1C6 | 16 | 1C6 | 46.150.18 | 1 in CM |
|------|--------|----|-------|-----------|---------|
| | 1 | | 1 | | 16 |
| C6 | 1C6 | 16 | 1C6 | 46.150.1 | 1 in CM |
| | 1 | | 1 | | 2 |
| N3P3 | NPNPNP | 16 | [N3P3 | 46.716.5 | 1 in CM |
| | 1 | 1 | 1 | | 1 |

CM 1

CRN 137741-82-3

CMF C30 H42 N3 O18 P3

CM 2

CRN 28605-81-4 CMF C15 H22 N2 O2 CCI IDS



D1-NCO

CM 3

CRN 999-61-1 CMF C6 H10 O3

CM · 4

CRN 629-11-8 CMF C6 H14 O2 $HO-(CH_2)_6-OH$

CM 5

CRN 107-21-1 CMF C2 H6 O2

. HO- CH2- CH2- OH

CM 6

CRN 100-21-0 CMF C8 H6 O4

CM 7

CRN 80-62-6 CMF C5 H8 O2

1 REFERENCES IN FILE CA (1907 TO DATE)
1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

REFERENCE 1

AN 141:141213 CA

TI Three-dimensionally crosslinked resins and transparent molded products thereof

IN Kawasaki, Noboru; Kogo, Osamu; Enya, Masahiro

PA Mitsui Chemicals Inc., Japan

SO Jpn. Kokai Tokkyo Koho, 20 pp. CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C08F220-14

ICS C08F220-18; C08F220-28; C08F220-36; C08F230-02; C08F290-06

CC 37-3 (Plastics Manufacture and Processing)

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE
PI JP 2004217700 A2 20040805 JP 2003-3682 20030109

PRAI JP 2003-3682 20030109

AB Compns. comprising Me methacrylate, R1(OCONHQNHCO2-Pol)nOCONHQNHCO2R2 [R1, R2 = hydroxy (meth)acrylate residue; Q = residue of aliphatic, aromatic, or

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alicyclic diisocyanate; Pol = polyether diol or polyester diol residue; n
= 1-5], and radical initiators are polymerized to give the resins with no
runaway reaction. Thus, a mixture of 100 parts Me methacrylate and 40 parts
of polyether-polyurethane acrylate (3:4:2 triethylene glycol-IPDI-2-
hydroxyethyl acrylate condensate) was mixed with 0.2 part tert-Bu
peroxy-2-ethylhexanoate and 0.1 part tert-Bu peroxy-3,5,5-
trimethylhexanoate, poured in a space between glass sheets, and heated to
70-130° to give a transparent resin sheet showing Tg 115°,
flexural modulus 3.2 GPa, H2O absorption 0.45%, no cracks at mold
releasing, and good chemical resistance.
polyether polyurethane acrylate methyl methacrylate polymer prepn
Polyurethanes, preparation
RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or
engineered material use); PREP (Preparation); USES (Uses)
   (acrylic-polyether-; three-dimensionally crosslinked Me
   methacrylate-polyurethane (meth)acrylate resins for transparent molded
   products)
Polyesters, preparation
Polyethers, preparation
RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or
engineered material use); PREP (Preparation); USES (Uses)
   (acrylic-polyurethane-; three-dimensionally crosslinked Me
   methacrylate-polyurethane (meth)acrylate resins for transparent molded
   products)
Transparent materials
   (three-dimensionally crosslinked Me methacrylate-polyurethane
   (meth)acrylate resins for transparent molded products)
2094-99-7, 3-Isopropenyl-α,α-dimethylbenzyl isocyanate
RL: RCT (Reactant); RACT (Reactant or reagent)
   (reaction with hydroxy-containing methacrylate)
727704-32-7P, 2-Hydroxyethyl acrylate-isophorone diisocyanate-methyl
                                            727704-34-9P,
methacrylate-triethylene glycol copolymer
1,1,3,3,5,5-Hexakis(acryloylethylenedioxy)cyclotriphosphazene-2-
hydroxyethyl methacrylate-methyl methacrylate-2,4-TDI-tripropylene glycol
            727704-36-1P, Ethylene oxide-glycidol methacrylate-N-(3-
isopropenyl-\alpha, \alpha-dimethylbenzyl)-1-methacryloyloxypropan-2-yl
carbamate-methyl methacrylate-propylene oxide-2,4-TDI copolymer
                                                                   727704-3
8-3P, Dimethyloltricyclodecane diacrylate-ethoxylated bisphenol
A-hexamethylene diisocyanate-2-hydroxyethyl methacrylate-methyl
methacrylate copolymer
                         727704-40-7P, Dibutylene glycol-dicyclopentanyl
acrylate-2-hydroxyethyl acrylate-isophorone diisocyanate-N-(3-isopropenyl-
\alpha, \alpha-dimethylbenzyl)-2-methacryloyloxyethyl carbamate-methyl
methacrylate-tris[N-(2-acryloyloxyethyl)] isocyanurate copolymer
                                                                    727704-
42-9P, Adipic acid-ethylene glycol-1,6-hexanediol-isophorone
diisocyanate-methyl methacrylate-pentaerythritol triacrylate copolymer
727704-44-1P, Ethylene glycol-1,1,3,3,5,5-hexakis(acryloylethylenedioxy)cy
clotriphosphazene-1,6-hexanediol-2-hydroxypropyl acrylate-
methylenebis(cyclohexyl isocyanate)-methyl methacrylate-terephthalic acid
            727704-47-4P, Adipic acid-1,4-butanediol-ethylene
glycol-2-hydroxypropyl acrylate-N-(3-isopropenyl-\alpha, \alpha-
dimethylbenzyl)-1-methacryloyloxypropan-2-yl carbamate-methyl
methacrylate-trimethylhexamethylene diisocyanate copolymer
                                                              727704-49-6P,
Cyclohexyl methacrylate-ethylene glycol-1,6-hexanediol-2-hydroxyethyl
methacrylate-isophthalic acid-methylenebis(cyclohexyl isocyanate)-methyl
                         727704-51-0P, Adipic acid-ethylene
methacrylate copolymer
qlycol-1,6-hexanediol-2-hydroxyethyl acrylate-isobornyl
methacrylate-isophorone diisocyanate-N-(3-isopropenyl-\alpha, \alpha-
dimethylbenzyl)-2-methacryloyloxyethyl carbamate-methyl
methacrylate-tris[N-(2-acryloyloxyethyl)] isocyanurate copolymer
RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or
engineered material use); PREP (Preparation); USES (Uses)
   (three-dimensionally crosslinked Me methacrylate-polyurethane
   (meth)acrylate resins for transparent molded products)
126710-00-7P
               126710-08-5P
RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT
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(Reactant or reagent)

(three-dimensionally crosslinked Me methacrylate-polyurethane (meth)acrylate resins for transparent molded products)

- ANSWER 2 OF 6 REGISTRY COPYRIGHT 2006 ACS on STN L3
- 727704-34-9 REGISTRY RN
- ED Entered STN: 17 Aug 2004
- 2-Propenoic acid, 2-methyl-, 2-hydroxyethyl ester, polymer with CN 2,4-diisocyanato-1-methylbenzene, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6hexakis[2-[(1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine, [(1-methyl-1,2-ethanediyl)bis(oxy)]bis[propanol] and methyl

2-methyl-2-propenoate (9CI) (CA INDEX NAME)

OTHER NAMES:

- CN 1,1,3,3,5,5-Hexakis(acryloylethylenedioxy)cyclotriphosphazene-2hydroxyethyl methacrylate-methyl methacrylate-2,4-TDI-tripropylene glycol copolymer
- MF (C30 H42 N3 O18 P3 . C9 H20 O4 . C9 H6 N2 O2 . C6 H10 O3 . C5 H8 O2)x
- CI PMS
- PCT Polyacrylic, Polyether, Polyphosphazene, Polyphosphazene formed, Polyurethane, Polyurethane formed
- SR
- LCSTN Files: CA, CAPLUS
- DT.CA CAplus document type: Patent
- Roles from patents: PREP (Preparation); PRP (Properties); USES (Uses)

Ring System Data

| Elemental | Elemental | Size of | Ring System | Ring | ŖID |
|-----------|-----------|------------|-------------|------------|------------|
| Analysis | Sequence | Ithe Rings | Formula | Identifier | Occurrence |
| EA · | l ES | † SZ | RF | RID | Count |
| ======= | +======= | +======= | +======== | +======= | +======= |
| C6 | IC6 | 16 | IC6 | 46.150.18 | 1 in CM |
| | 1 | 1 | 1 | 1 | 4 |
| N3P3 | INPNPNP | 16 | N3P3 | 46.716.5 | 1 in CM |
| | 1 | 1 | 1 | 1 | 1 |

CM 1

CRN 137741-82-3

CMF C30 H42 N3 O18 P3

$$\begin{array}{c} O \\ H_2C = CH - C - O - CH_2 - CH_2 - O \\ H_2C = CH - C - O - CH_2 - CH_2 - O - CH_2 - CH_2 - O - C - CH = CH_2 \\ \hline \\ H_2C = CH - C - O - CH_2 - CH_2 - O - CH_2 - CH_2 - O - C - CH = CH_2 \\ \hline \\ H_2C = CH - C - O - CH_2 - CH_2 - O - CH_2 - CH_2 - O - C - CH = CH_2 \\ \hline \\ \end{array}$$

2 CM

24800-44-0 CRN CMF C9 H20 O4

CCI IDS 3 (D1-Me)

CM 3

CRN 868-77-9 CMF C6 H10 O3

$$^{\rm H_2C}$$
 O $^{\rm H_2}$ $^{\rm H_2}$ $^{\rm Me-}$ C-C-O-CH₂-CH₂-OH

CM 4

CRN 584-84-9 CMF C9 H6 N2 O2

CM 5

CRN 80-62-6 CMF C5 H8 O2

1 REFERENCES IN FILE CA (1907 TO DATE)
1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

REFERENCE 1

AN 141:141213 CA

TI Three-dimensionally crosslinked resins and transparent molded products thereof

IN Kawasaki, Noboru; Kogo, Osamu; Enya, Masahiro

PA Mitsui Chemicals Inc., Japan

SO Jpn. Kokai Tokkyo Koho, 20 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C08F220-14 ICS C08F220-18; C08F220-28; C08F220-36; C08F230-02; C08F290-06

CC 37-3 (Plastics Manufacture and Processing)

FAN.CNT 1

```
PATENT NO.
                    KIND DATE
                                      APPLICATION NO. DATE
     ______
PI JP 2004217700 A2 200
PRAI JP 2003-3682 20030109
                                       JP 2003-3682
                            20040805
                                                          20030109
    Compns. comprising Me methacrylate, R1(OCONHQNHCO2-Pol)nOCONHQNHCO2R2 [R1,
     R2 = hydroxy (meth)acrylate residue; Q = residue of aliphatic, aromatic, or
     alicyclic diisocyanate; Pol = polyether diol or polyester diol residue; n
     = 1-5], and radical initiators are polymerized to give the resins with no
     runaway reaction. Thus, a mixture of 100 parts Me methacrylate and 40 parts
     of polyether-polyurethane acrylate (3:4:2 triethylene glycol-IPDI-2-
     hydroxyethyl acrylate condensate) was mixed with 0.2 part tert-Bu
     peroxy-2-ethylhexanoate and 0.1 part tert-Bu peroxy-3,5,5-
     trimethylhexanoate, poured in a space between glass sheets, and heated to
     70-130° to give a transparent resin sheet showing Tg 115°,
     flexural modulus 3.2 GPa, H2O absorption 0.45%, no cracks at mold
     releasing, and good chemical resistance.
     polyether polyurethane acrylate methyl methacrylate polymer prepn
     Polyurethanes, preparation
     RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or
     engineered material use); PREP (Preparation); USES (Uses)
        (acrylic-polyether-; three-dimensionally crosslinked Me
       methacrylate-polyurethane (meth)acrylate resins for transparent molded
       products)
     Polyesters, preparation
ΙT
     Polyethers, preparation
     RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or
     engineered material use); PREP (Preparation); USES (Uses)
        (acrylic-polyurethane-; three-dimensionally crosslinked Me
       methacrylate-polyurethane (meth)acrylate resins for transparent molded
       products)
IT
     Transparent materials
        (three-dimensionally crosslinked Me methacrylate-polyurethane
        (meth) acrylate resins for transparent molded products)
     2094-99-7, 3-Isopropenyl-\alpha, \alpha-dimethylbenzyl isocyanate
ΙT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction with hydroxy-containing methacrylate)
     727704-32-7P, 2-Hydroxyethyl acrylate-isophorone diisocyanate-methyl
ΙT
     methacrylate-triethylene glycol copolymer 727704-34-9P,
     1,1,3,3,5,5-Hexakis(acryloylethylenedioxy)cyclotriphosphazene-2-
     hydroxyethyl methacrylate-methyl methacrylate-2,4-TDI-tripropylene glycol
               727704-36-1P, Ethylene oxide-glycidol methacrylate-N-(3-
     isopropenyl-\alpha, \alpha-dimethylbenzyl)-1-methacryloyloxypropan-2-yl
     carbamate-methyl methacrylate-propylene oxide-2,4-TDI copolymer
                                                                       727704-3
     8-3P, Dimethyloltricyclodecane diacrylate-ethoxylated bisphenol
     A-hexamethylene diisocyanate-2-hydroxyethyl methacrylate-methyl
     methacrylate copolymer 727704-40-7P, Dibutylene glycol-dicyclopentanyl
     acrylate-2-hydroxyethyl acrylate-isophorone diisocyanate-N-(3-isopropenyl-
     \alpha, \alpha-dimethylbenzyl)-2-methacryloyloxyethyl carbamate-methyl
     methacrylate-tris[N-(2-acryloyloxyethyl)] isocyanurate copolymer
                                                                        727704-
     42-9P, Adipic acid-ethylene glycol-1,6-hexanediol-isophorone
     diisocyanate-methyl methacrylate-pentaerythritol triacrylate copolymer
     727704-44-1P, Ethylene glycol-1,1,3,3,5,5-hexakis(acryloylethylenedioxy)cy
     clotriphosphazene-1,6-hexanediol-2-hydroxypropyl acrylate-
     methylenebis(cyclohexyl isocyanate)-methyl methacrylate-terephthalic acid
                 727704-47-4P, Adipic acid-1,4-butanediol-ethylene
     copolymer
     glycol-2-hydroxypropyl acrylate-N-(3-isopropenyl-\alpha, \alpha-
     dimethylbenzyl)-1-methacryloyloxypropan-2-yl carbamate-methyl
     methacrylate-trimethylhexamethylene diisocyanate copolymer
     Cyclohexyl methacrylate-ethylene glycol-1,6-hexanediol-2-hydroxyethyl
     methacrylate-isophthalic acid-methylenebis(cyclohexyl isocyanate)-methyl
                              727704-51-0P, Adipic acid-ethylene
     methacrylate copolymer
     glycol-1,6-hexanediol-2-hydroxyethyl acrylate-isobornyl
     methacrylate-isophorone diisocyanate-N-(3-isopropenyl-\alpha, \alpha-
     dimethylbenzyl)-2-methacryloyloxyethyl carbamate-methyl
     methacrylate-tris[N-(2-acryloyloxyethyl)] isocyanurate copolymer
```

```
RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or
     engineered material use); PREP (Preparation); USES (Uses)
        (three-dimensionally crosslinked Me methacrylate-polyurethane
        (meth)acrylate resins for transparent molded products)
ΙT
     126710-00-7P
                    126710-08-5P
     RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT
     (Reactant or reagent)
        (three-dimensionally crosslinked Me methacrylate-polyurethane
        (meth)acrylate resins for transparent molded products)
     ANSWER 3 OF 6 REGISTRY COPYRIGHT 2006 ACS on STN
L3
     255384-20-4 REGISTRY
RN
ED
     Entered STN: 09 Feb 2000
     2-Propenoic acid, 2-methyl-, 2-(hydroxymethyl)-2-[[(2-methyl-1-oxo-2-
CN
     propenyl)oxy]methyl]-1,3-propanediyl ester, polymer with
     2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-
     propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine and
     (1-methylethylidene)bis[4,1-phenyleneoxy(2-hydroxy-3,1-propanediyl)]
     di-2-propenoate (9CI) (CA INDEX NAME)
OTHER CA INDEX NAMES:
     1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-
     hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-, polymer with
     2-(hydroxymethyl)-2-[((2-methyl-1-oxo-2-propenyl)oxy]methyl]-1,3-
     propanediyl bis(2-methyl-2-propenoate) and (1-methylethylidene)bis[4,1-
     phenyleneoxy(2-hydroxy-3,1-propanediyl)] di-2-propenoate (9CI)
CN
     2-Propenoic acid, (1-methylethylidene)bis[4,1-phenyleneoxy(2-hydroxy-
     3,1-propanediyl)] ester, polymer with 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-
     hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-
     triazatriphosphorine and 2-(hydroxymethyl)-2-[[(2-methyl-1-oxo-2-
     propenyl)oxy]methyl]-1,3-propanediyl bis(2-methyl-2-propenoate) (9CI)
OTHER NAMES:
     Hexa(2-methacryloyloxyethoxy)cyclotriphosphazene-pentaerythritol
     trimethacrylate-Ripoxy SP 1509 copolymer
     (C36 H54 N3 O18 P3 . C27 H32 O8 . C17 H24 O7)x
CI
PCT
     Polyacrylic, Polyether, Polyphosphazene, Polyphosphazene formed
SR
                  CA, CAPLUS, USPATFULL
LC
     STN Files:
DT.CA CAplus document type: Patent
       Roles from patents: PREP (Preparation); PRP (Properties); USES (Uses)
Ring System Data
Elemental|Elemental| Size of |Ring System|
                                             Ring
Analysis | Sequence | the Rings | Formula | Identifier | Occurrence
           ES
                       SZ
                                  RF
                                         1
                                            RID | Count
        -1
                   1
                            =+======
                                       ==+======+===========
                                         |46.150.18 | 2 in CM
                   16
                             |C6
         |C6
                                                   12
                   1
                             1
                                         1
                                         |46.716.5 | 1 in CM
N3P3
         INPNPNP
                   16
                             |N3P3
                                                    |1
```

CM 1

CRN 92832-53-6

CMF C36 H54 N3 O18 P3

CM 2

CRN 4687-94-9 CMF C27 H32 O8

PAGE 1-B

CM 3

CRN 3524-66-1 CMF C17 H24 O7

2 REFERENCES IN FILE CA (1907 TO DATE)
2 REFERENCES IN FILE CAPLUS (1907 TO DATE)

REFERENCE 1

AN 132:109482 CA

TI Weather-resistant compositions and coating materials and molded articles

```
Ogawa, Takashi; Mori, Koji; Akada, Mitsuo
PA
     Otsuka Chemical Co., Ltd., Japan
SO
     PCT Int. Appl., 58 pp.
     CODEN: PIXXD2
DΤ
     Patent
     Japanese
LA
IC
     ICM C09D004-02
     ICS C08K005-3475; C08L033-04; C09D007-12; C07D249-20; C09K003-00
     42-10 (Coatings, Inks, and Related Products)
CC
     Section cross-reference(s): 55
FAN.CNT 1
     PATENT NO.
                     KIND DATE
                                          APPLICATION NO. DATE.
                           -----
     -----
                     ____
                                          _____
    WO 2000002964
                     A1 20000120
                                         WO 1999-JP3715
                                                          19990708
PΙ
        W: AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ,
            DE, DK, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS,
            JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK,
            MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ,
            TM, TR, TT, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ,
            MD, RU, TJ, TM
        RW: GH, GM, KE, LS, MW, SD, SL, SZ, UG, ZW, AT, BE, CH, CY, DE, DK,
            ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG,
            CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
     AU 9946504
                                                            19990708
                      A1
                            20000201
                                          AU 1999-46504
                            20010606
                                           EP 1999-929769
                                                            19990708
     EP 1104790
                      A1
            AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, SI, LT, LV, FI, RO
     US 6703139
                            20040309
                                           US 2001-720974
                                                            20010103
                      В1
                     19980710
PRAI JP 1998-196099
                     19990708
     WO 1999-JP3715
     The title compns. contain the reaction products of
AB
     bisbenzotriazolylphenols with lactones, (meth)acrylates, and curing
     agents. Thus, a coating composition contained Aronix M 8030 40,
     tetrahydrofurfuryl acrylate 20, trimethylolpropane triacrylate 40,
     e-caprolactone-2,2'-methylenebis[6-(2H-1,2,3-benzotriazole-2-yl)-4-
     (2-hydroxyethyl)phenol] copolymer 3, and Darocur 1173 3 parts.
ST
     weather resistant coating UV absorber; benzotriazolylhydroxyethylphenol
     lactone copolymer UV absorber; polyester benzotriazolylhydroxyethylphenol
     caprolactone UV absorber
ΙT
     Coating materials
        (abrasion-resistant; weather-resistant coating materials and moldings
        containing bisbenzotriazolylhydroxyethylphenol lactone copolymers)
ΙT
     Polymerization catalysts
        (photopolymn.; weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
     Vinyl compounds, uses
ΙT
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (polymers; weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
IΤ
     Polymerization catalysts
        (radical; weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
ΙT
     Polymerization
        (ring-opening; weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
IT
     Crosslinking agents
     Lacquers
     UV stabilizers
    Yellowing •
        (weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
IT
     Lactones
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (weather-resistant coating materials and moldings containing
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IN

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bisbenzotriazolylhydroxyethylphenol lactone copolymers)
ፐጥ
     Coating materials
        (weather-resistant; weather-resistant coating materials and moldings
        containing bisbenzotriazolylhydroxyethylphenol lactone copolymers)
IT
     7473-98-5, Darocur 1173
     RL: CAT (Catalyst use); USES (Uses)
        (Darocur 1173; weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
ΙT
     947-19-3, Irgacure 184
     RL: CAT (Catalyst use); USES (Uses)
        (Irgacure 184; weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
IT
     71868-10-5, Irgacure 907
     RL: CAT (Catalyst use); USES (Uses)
        (Irgacure 907; weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
ΙT
     78-67-1, AIBN
                    94-36-0, Benzoyl peroxide, uses
     RL: CAT (Catalyst use); USES (Uses)
        (weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
     255384-16-8P
                                                                  255384-21-5P
ΙT
                    255384-17-9P
                                  255384-18-0P
                                                  255384-20-4P
     255827-82-8P, ε-Caprolactone-2,2'-methylenebis[6-(2H-1,2,3-
     benzotriazole-2-yl)-4-(2-hydroxyethyl)phenol] copolymer
     RL: IMF (Industrial manufacture); MOA (Modifier or additive use); PREP
     (Preparation); USES (Uses)
        (weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
ΙT
     38808-38-7P, Butyl acrylate-methacrylic acid-methyl methacrylate-vinyl
     acetate copolymer
                       77492-22-9P, Acrylic acid-butyl acrylate-2-ethylhexyl
    methacrylate-hydroxyethyl methacrylate-methyl methacrylate-styrene
                 172156-26-2P, Acrylic acid-butyl acrylate-2-ethylhexyl
    methacrylate-hydroxyethyl methacrylate-styrene copolymer
    Acrylic acid-butyl acrylate-2-ethylhexyl methacrylate-formaldehyde-
    hydroxyethyl methacrylate-melamine-methyl methacrylate-styrene copolymer
     255820-11-2P, Acrylic acid-butyl acrylate-2-ethylhexyl
    methacrylate-formaldehyde-hydroxyethyl methacrylate-melamine-styrene
     copolymer
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
RE.CNT
              THERE ARE 9 CITED REFERENCES AVAILABLE FOR THIS RECORD
(1) Daicel Chemical Industries Ltd; JP 10265557 A 1998 CAPLUS
(2) Nippon Paint Co Ltd; US 5760137 A CAPLUS
(3) Nippon Paint Co Ltd; EP 756614 A1 CAPLUS
(4) Nippon Paint Co Ltd; WO 9522250 A1
(5) Nippon Paint Co Ltd; JP 10503787 A 1998
(6) Otsuka Chemical Co Ltd; US 5922882 A CAPLUS
(7) Otsuka Chemical Co Ltd; EP 855393 A1 CAPLUS
(8) Otsuka Chemical Co Ltd; WO 9735847 A1 CAPLUS
(9) Otsuka Chemical Co Ltd; JP 09316060 A 1997 CAPLUS
REFERENCE 2
ΑN
     132:109425 CA
     Coating compositions with good resistance to metal ion-induced
TI
     discoloration and weather and UV absorbents for use in the compositions
IN
     Ogawa, Takashi; Akada, Mitsuo; Mori, Hiroshi
PΑ
     Ohtsuka Chemical Co., Ltd., Japan
SO
     Jpn. Kokai Tokkyo Koho, 22 pp.
     CODEN: JKXXAF
DT
     Patent
LA
     Japanese
IC
     ICM C09D005-32
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ICS C09D004-02; C09D007-12; C09D163-10; C09D167-07; C09D171-00;

C09D175-16 C08F290-06; C09K003-00 42-5 (Coatings, Inks, and Related Products) FAN.CNT 1 PATENT NO. KIND DATE APPLICATION NO. DATE --------- ----------JP 2000017204 A2 20000118 JP 1998-186543 19980701 ΡI JP 2918543 B2 19990712 PRAI JP 1998-186543 19980701 The coating compns. contain (A) radically polymerizable monomers or/and curable resins, and bisbenzotriazolylphenol compds. Thus, mixing Magicron TC 16U Clear (aminoacrylic clear coating) with 2 phr RUVA-100 $\{2, 2'-methylenebis[6-(2H-1, 2, 3-benzotriazol-2-yl)-4-(2-yl)]$ hydroxyethyl)phenol]} gave a clear top coating composition for protecting prefinished metal sheet. ST multilayer coating clear coating UV absorbent bisbenzotriazolylphenol; light stabilizer bisbenzotriazolylphenol clear coating; metal ion discoloration resistance coating ΙT Epoxy resins, uses RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (acrylic; coating compns. with good resistance to metal ion-induced discoloration and weather and UV absorbents for use in compns.) ΙT RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (clear coating; coating compns. with good resistance to metal ion-induced discoloration and weather and UV absorbents for use in compns.) ΙT UV stabilizers (coating compns. with good resistance to metal ion-induced discoloration and weather and UV absorbents for use in compns.) ΙT Coating materials (transparent; coating compns. with good resistance to metal ion-induced discoloration and weather and UV absorbents for use in compns.) ΙT Epoxy resins, uses RL: TEM (Technical or engineered material use); USES (Uses) (under coating; coating compns. with good resistance to metal ion-induced discoloration and weather and UV absorbents for use in compns.) ΙT Coating materials (weather-resistant; coating compns. with good resistance to metal ion-induced discoloration and weather and UV absorbents for use in IT 9003-08-1P, Formaldehyde-melamine copolymer 38808-38-7P 51097-42-8P, Acrylic acid-butyl methacrylate-2-ethylhexyl acrylate-hydroxyethyl methacrylate-styrene copolymer 63747-55-7P, Acrylic acid-butyl methacrylate-2-ethylhexyl acrylate-hydroxyethyl methacrylate-methyl methacrylate-styrene copolymer 255384-16-8P, Aronix M-8030-tetrahydrofurfuryl acrylate-trimethylolpropane triacrylate 255384-17-9P, 2-(2-Ethoxyethoxy)ethyl acrylate-pentaerythritol triacrylate-Viscoat 3700 copolymer 255384-18-0P 255384-19-1P, Art Resin UN 3320HA-pentaerythritol triacrylate-pentaerythritol trimethacrylate-tetrahydrofurfuryl acrylate copolymer 255384-20-4P, Hexa (2-methacryloyloxyethoxy) cyclotriphosphazene-pentaerythritol trimethacrylate-Ripoxy SP 1509 copolymer 255384-21-5P RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (clear coating; coating compns. with good resistance to metal ion-induced discoloration and weather and UV absorbents for use in

196516-61-7, RUVA-100 196516-62-8 196516-63-9 196516-64-0

compns.)

ΙT

```
(light stabilizer; coating compns. with good resistance to metal
        ion-induced discoloration and weather and UV absorbents for use in
        compns.)
ΙΤ
     255390-94-4, Magicron TC 16U Clear
                                          255390-95-5, Magicron TA 16
     RL: POF (Polymer in formulation); PRP (Properties); TEM (Technical or
     engineered material use); USES (Uses)
        (top clear coating; coating compns. with good resistance to metal
        ion-induced discoloration and weather and UV absorbents for use in
        compns.)
L3
     ANSWER 4 OF 6 REGISTRY COPYRIGHT 2006 ACS on STN
RN
     212397-70-1 REGISTRY
     Entered STN: 08 Oct 1998
ED
     Benzenecarboperoxoic acid, 4,4'-carbonylbis[[[2-[(2-methyl-1-oxo-2-
CN
     propenyl)oxy]ethoxy]carbonyl]-, bis(1,1-dimethylethyl) ester, polymer with
     2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-
     propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine and
     \alpha-(1-oxo-2-propenyl)-\omega-[(1-oxo-2-propenyl)oxy]poly(oxy-1,2-
     ethanediyl), graft (9CI)
                              (CA INDEX NAME)
OTHER CA INDEX NAMES:
     1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-
     hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-, polymer with
     bis(1,1-dimethylethyl) 4,4'-carbonylbis[[[2-[(2-methyl-1-oxo-2-
     propenyl)oxy]ethoxy]carbonyl]benzenecarboperoxoate] and
     \alpha-(1-oxo-2-propenyl)-\omega-[(1-oxo-2-propenyl)oxy]poly(oxy-1,2-
     ethanediyl), graft (9CI)
CN
     Poly(oxy-1,2-ethanediyl), \alpha-(1-oxo-2-propenyl)-\omega-[(1-oxo-2-
     propenyl)oxy]-, polymer with bis(1,1-dimethylethyl) 4,4'-carbonylbis[[[2-
     [(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]carbonyl]benzenecarboperoxoate] and
     2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-
     propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine, graft (9CI)
OTHER NAMES:
     3,3'-Bis(methacryloyloxyethoxycarbonyl)-4,4'-bis(tert-
     butylperoxycarbonyl)benzophenone-polyethylene glycol diacrylate-
     hexakis(methacryloyloxyethoxy)cyclotriphosphazene graft copolymer
     (C37 H42 O15 . C36 H54 N3 O18 P3 . (C2 H4 O)n C6 H6 O3)x
MF
CI
PCT
     Polyacrylic, Polyether, Polyphosphazene, Polyphosphazene formed
SR
     STN Files:
                 CA, CAPLUS
DT.CA CAplus document type: Patent
RLD.P
       Roles for non-specific derivatives from patents: PREP (Preparation);
       USES (Uses)
Ring System Data
Elemental|Elemental| Size of |Ring System|
                                             Ring
Analysis |Sequence | the Rings | Formula | Identifier | Occurrence
                               RF
                                         | RID | Count
   EΑ
            ES
                   l SZ
                            - 1
         1
                                       --+-----
                                         |46.716.5 | 1 in CM
N3P3
         INPNPNP
                   16
                             | N3P3
                                                   12
                   1
                             1
                                         1
```

RL: MOA (Modifier or additive use); USES (Uses)

CM 1

1C6

C6

CRN 204009-97-2 CMF C37 H42 O15

16

1C6

|46.150.18 | 2 in CM

CM 2

CRN 92832-53-6 CMF C36 H54 N3 O18 P3

CM 3

CRN 26570-48-9

CMF (C2 H4 O)n C6 H6 O3

CCI PMS

$$H_2C = CH - C - CH_2 - CH_2$$

1 REFERENCES IN FILE CA (1907 TO DATE)

1 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA

1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

REFERENCE 1

AN 129:224371 CA

TI Ionic conductor and its preparation

IN Okuo, Masaki; Hikuchi, Keiki; Omura, Hiroshi; Suyama, Shuji

PA Nippon Oil and Fats Co., Ltd., Japan

```
Jpn. Kokai Tokkyo Koho, 19 pp.
SO
    CODEN: JKXXAF
DT
     Patent
LA
     Japanese
IC
     ICM H01B001-12
     ICS C08L051-00; C08F002-46
CC
     76-2 (Electric Phenomena)
     Section cross-reference(s): 35, 38, 72
FAN.CNT 1
     PATENT NO.
                     KIND DATE
                                          APPLICATION NO. DATE
     -----
                     _ ---
                           _____
                                           -----
     JP 10208542
                     A2
                           19980807
PΙ
                                          JP 1997-8975
                                                            19970121
PRAI JP 1997-8975
                     19970121
     The conductor comprises (A) a support, (B) a polymer layer containing
     photopolymn. initiating group, and (C) a ion conducting layer formed from
     a composition containing (a) ion conducting monomer and (b) soluble
electrolyte salt
     and/or electrolyte salt monomer, which is bonded with B via photopolymn.
     initiating group. B is formed on A, a composition containing ion conducting
    monomer and soluble electrolyte salt is contacted with B with irradiation of
     active energy to give the title ionic conductor. Ionic conductor layers
    having excellent peeling resistance and high strength are obtained.
ST
     ionic conductor peeling resistant; photopolymn ionic conductor coating
IT
     Polyoxyalkylenes, uses
     Polyoxyalkylenes, uses
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (acrylic-polyester-; graft copolymer ionic conductor layer and its
        preparation)
TΤ
     Polyesters, uses
     Polyesters, uses
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (acrylic-polyoxyalkylene-; graft copolymer ionic conductor layer and
        its preparation)
IT
     Polyoxyalkylenes, uses
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (acrylic; graft copolymer ionic conductor layer and its preparation)
TT
    Coating materials
        (elec. conductive; graft copolymer ionic conductor layer and its
       preparation)
TΤ
    Conducting polymers
     Ionic conductors
        (graft copolymer ionic conductor layer and its preparation)
IT
     Polymers, uses
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (graft; graft copolymer ionic conductor layer and its preparation)
IT
     Solid electrolytes
        (ionic graft copolymer conductor layer and its preparation)
ΙT
     Polyurethanes, uses
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (polyoxyalkylene-, acrylic; graft copolymer ionic conductor layer and
        its preparation)
ΙT
     7791-03-9, Lithium perchlorate
     RL: MOA (Modifier or additive use); USES (Uses)
        (4; graft copolymer ionic conductor layer and its preparation)
     13641-97-9DP, Acryloyl isocyanate, reaction products with acryloyl
     isocyanate, polymer with polyethylene oxides and (meth)acrylates,
     potassium complex
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (MAI; graft copolymer ionic conductor layer and its preparation)
```

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ΙT
     56-81-5DP, 1,2,3-Propanetriol, reaction product with ethylene glycol and
     acrylic acid, polymer with polyethylene oxides and (meth)acrylates, uses
    79-10-7DP, 2-Propenoic acid, reaction product with ethylene glycol and
     glycerin, polymer with polyethylene oxides and (meth)acrylates, uses
     79-41-4DP, polymer with polyethylene oxides and (meth)acrylates
     80-62-6DP, polymer with polyethylene oxides and (meth)acrylates
     106-91-2DP, reaction products with partially saponified poly(vinyl acetate),
     polymer with polyethylene oxides and (meth)acrylates
                                                          107-21-1DP,
     1,2-Ethanediol, reaction product with glycerin and acrylic acid, polymer
     with polyethylene oxides and (meth)acrylates, uses
                                                          2680-03-7DP, polymer
     with polyethylene oxides and (meth)acrylates, potassium complex
     3524-68-3DP, Kayarad PET-30, polymer with polyethylene oxides and
     (meth)acrylates, potassium complex
                                          6900-35-2DP, Potassium methacrylate,
     polymer with polyethylene oxides and (meth)acrylates, potassium complex
     7439-93-2DP, Lithium, complexes with ion-conducting polymers, uses
     7440-09-7DP, Potassium, complexes with ion-conducting polymers, uses
     7440-50-8DP, Copper, complexes with ion-conducting polymers, uses
     14798-03-9DP, Ammonium, complexes with ion-conducting polymers, uses
     16325-47-6DP, Ammonium methacrylate, polymer with polyethylene oxides and
     (meth)acrylates
                      25805-17-8DP, Polyethyloxazoline, reaction products with
     acryloyl isocyanate, polymer with polyethylene oxides and (meth)acrylates,
    potassium complex
                         25852-47-5DP, polymer with polyethylene oxides and
     (meth)acrylates, potassium complex 26570-48-9DP, polymer with
    polyethylene oxides and (meth)acrylates
                                               69488-61-5DP,
     Poly[(propionylimino)ethylene], reaction products with acryloyl
     isocyanate, polymer with polyethylene oxides and (meth)acrylates,
                         101027-21-8DP, Gohseran L 302, reaction product with
     potassium complex
     glycidyl methacrylate, polymer with polyethylene oxides and
                     120487-52-7DP, Eleminol RS 30, polymer with polyethylene
     (meth)acrylates
     oxides and (meth)acrylates
                                 203309-79-9DP, 1-[4-[2-[2-
     (Methacryloyloxy)ethoxycarbonyloxy]ethoxy]phenyl]-2-hydroxy-2-methylpropan-
     1-one, polymer with polyethylene oxides and (meth)acrylates
     212397-55-2DP, Butyl methacrylate-lauryl methacrylate-1-[4-[2-[2-
     (methacryloyloxy)ethoxycarbonyloxy]ethoxy]phenyl]-2-hydroxy-2-methylpropan-
     1-one-methyl methacrylate-polyethylene glycol diacrylate graft copolymer,
     lithium complex
                      212397-63-2DP, polymer with polyethylene oxides and
                      212397-64-3P, Cerium methacrylate-1-[4-[2-[2-
     (meth)acrylates
     (methacryloyloxy)ethoxycarbonyloxy]ethoxy]phenyl]-2-hydroxy-2-methylpropan-
     1-onemethyl methacrylate-polyethylene glycol diacrylate-potassium
     methacrylate-vinylpyrrolidone graft copolymer
                                                     212397-65-4P,
     1-[4-[2-[2-(Methacryloyloxy)ethoxycarbonyloxy]ethoxy]phenyl]-2-hydroxy-2-
     methylpropan-1-one-methyl methacrylate-polypropylene glycol
     dimethacrylate-Eleminol RS 30 graft copolymer
                                                    212397-66-5DP,
     1-[4-[2-[2-(Methacryloyloxy)ethoxycarbonyloxy]ethoxy]phenyl]-2-hydroxy-2-
     methylpropan-1-one-methyl methacrylate-vinylpyridine graft copolymer,
                      212397-67-6DP, 1-[4-[2-[2-(Methacryloyloxy)ethoxycarbony]]
     lithium complex
     loxy]ethoxy]phenyl]-2-hydroxy-2-methylpropan-1-one-methyl
     methacrylate-polyethylene glycol diacrylate-polyethylene glycol
     dimethacrylate graft copolymer, complex with lithium and potassium
     212397-67-6DP, 1-[4-[2-(2-(Methacryloyloxy)ethoxycarbonyloxy)ethoxy]phenyl
     ]-2-hydroxy-2-methylpropan-1-one-methyl methacrylate-polyethylene glycol
     diacrylate-polyethylene glycol dimethacrylate graft copolymer, potassium
               212397-68-7DP, 1-[4-[2-(Methacryloyloxy)ethoxycarbonyloxy]eth
     oxy]phenyl]-2-hydroxy-2-methylpropan-1-one-KBM 503-methyl
     methacrylate-polyethylene glycol diacrylate-polyethylene glycol
     dimethacrylate graft copolymer, potassium complex
                                                        212397-69-8DP,
     N, N-Dimethylacrylamide-1-[4-[2-(2-(methacryloyloxy)ethoxycarbonyloxy]ethox
     y]phenyl]-2-hydroxy-2-methylpropan-1-one-methyl methacrylate-
     pentaerythritol triacrylate-polyethylene glycol diacrylate-polyethylene
     glycol dimethacrylate graft copolymer, potassium complex
                                                                212397-70-1DP,
     3,3'-Bis(methacryloyloxyethoxycarbonyl)-4,4'-bis(tert-
     butylperoxycarbonyl)benzophenone-polyethylene glycol diacrylate-
     Hexakis (methacryloyloxyethoxy) cyclotriphosphazene graft copolymer, lithium
               212515-88-3DP, Glycidyl methacrylate-1-[4-[2-[2-
     (methacryloyloxy) ethoxycarbonyloxy] ethoxy] phenyl] -2-hydroxy-2-methylpropan-
```

1-one-methyl methacrylate-polyethylene glycol diacrylate-propylene glycol monomethyl ether-triethylene tetramine graft copolymer, lithium complex 212520-79-1DP, ammonium complex 212520-81-5DP, Hexamethylene diisocyanate polyethylene glycol copolymer diacrylate-1-[4-[2-[2-(methacryloyloxy)ethoxycarbonyloxy]ethoxy]phenyl]-2-hydroxy-2-methylpropan-1-one-methyl methacrylate-polyethylene glycol diacrylate graft copolymer, 212571-38-5DP, GE 3A, polymer with polyethylene oxides and (meth)acrylates 212571-90-9DP, Acrylonitrile-GE 3A-1-[4-[2-[2-(methacryloyloxy)ethoxycarbonyloxy]ethoxy]phenyl]-2-hydroxy-2-methylpropan-1-one-methyl methacrylate-polyethylene glycol diacrylate-polyethylene glycol dimethacrylate graft copolymer, potassium complex 212572-38-8P RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (graft copolymer ionic conductor layer and its preparation) 7447-39-4, Copper chloride (CuCl2), uses Potassium iodide (KI), uses 7778-74-7, Potassium perchlorate RL: MOA (Modifier or additive use); USES (Uses) (graft copolymer ionic conductor layer and its preparation) 212397-54-1P, 1-[4-[2-[2-(Methacryloyloxy)ethoxycarbonyloxy]ethoxy]phenyl]-2-hydroxy-2-methylpropan-1-one-methyl methacrylate-polyethylene glycol diacrylate graft copolymer 212397-55-2P, Butyl methacrylate-lauryl methacrylate-1-[4-[2-[2-(methacryloyloxy)ethoxycarbonyloxy]ethoxy]phenyl]-2-hydroxy-2-methylpropan-1-one-methyl methacrylate-polyethylene glycol diacrylate graft copolymer 212397-56-3P, Butyl methacrylate-1-[4-[2-[2-(methacryloyloxy) ethoxycarbonyloxy] ethoxy}phenyl]-2-hydroxy-2-methylpropan-1-one-methyl methacrylate-polyethylene glycol diacrylate graft copolymer 212397-57-4P, Dibutyl fumarate-methyl [2-[4-(2-hydroxy-2-methyl-1oxopropyl)phenoxy]ethyl] fumarate-polyethylene glycol diacrylate graft 212397-58-5P, Bis[2-[4-(2-hydroxy-2-methyl-1oxopropyl)phenoxy]ethyl] itaconate-dibutyl fumarate-polyethylene glycol diacrylate graft copolymer 212397-59-6P, Lauryl methacrylate-1-[4-(2methacryloylethoxy)phenyl]-2,2-dimethoxy-2-phenyl ethan-1-one-methyl methacrylate-polyethylene glycol diacrylate graft copolymer 212397-60-9P , 1,2-Diphenyl-1,2-ethanedione-2-O-acryloyloxime-1-[4-[2-[2-(methacryloyloxy) ethoxycarbonyloxy] ethoxy[phenyl]-2-hydroxy-2-methylpropan-1-one-methyl methacrylate-polyethylene glycol diacrylate graft copolymer 212397-61-0P, Hydroxyethyl methacrylate-methacrylic acid-1-[4-[2-[2-(methacryloyloxy) ethoxycarbonyloxy] ethoxy[phenyl]-2-hydroxy-2-methylpropan-1-one-methyl methacrylate-polyethylene glycol diacrylate graft copolymer 212397-62-1P, Bu methacrylate-N, N-dimethylaminoethyl methacrylate-1-[4-[2-[2-(methacryloyloxy)ethoxycarbonyloxy]ethoxy]phenyl]-2-hydroxy-2methylpropan-1-one-Me methacrylate-polyethylene glycol diacrylate-ST graft copolymer RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (ionic graft copolymer conductor layer and its preparation) ANSWER 5 OF 6 REGISTRY COPYRIGHT 2006 ACS on STN 155880-83-4 REGISTRY Entered STN: 22 Jun 1994 11,15-Dioxa-2,9-diazaoctadec-17-enoic acid, 10,16-dioxo-13,13-bis[{(1-oxo-2-propenyl)oxy]methyl]-, 3-[(1-oxo-2-propenyl)oxy]-2, 2-bis[[(1-oxo-2-propenyl)oxy]-2, 2-bis[](1-oxo-2-propenyl)oxy]-2, 2-bis[](1propenyl)oxy]methyl]propyl ester, polymer with 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6triazatriphosphorine and Sartomer C 9505 (9CI) (CA INDEX NAME) OTHER CA INDEX NAMES: 1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-, polymer with 3-[(1-oxo-2-propenyl)oxy]-2,2-bis[[(1-oxo-2-propenyl)oxy]methyl]propyl10,16-dioxo-13,13-bis[[(1-oxo-2-propenyl)oxy]methyl]-11,15-dioxa-2,9diazaoctadec-17-enoate and Sartomer C 9505 (9CI) Sartomer C 9505, polymer with 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine

and 3-[(1-oxo-2-propenyl)oxy]-2,2-bis[[(1-oxo-2-propenyl)oxy]methyl]propyl 10,16-dioxo-13,13-bis[[(1-oxo-2-propenyl)oxy]methyl]-11,15-dioxa-2,9-

ΙT

L3

RN

ED

diazaoctadec-17-enoate (9CI)

OTHER NAMES:

CN 1,1,3,3,5,5-hexa (methacryloylethylenedioxy) cyclotriphosphazene-

Sartomer 9505-UA 306H copolymer

MF (C36 H54 N3 O18 P3 . C36 H48 N2 O16 . Unspecified)x

CI PMS

PCT Manual component, Polyacrylic, Polyother, Polyphosphazene, Polyphosphazene formed

SR CA

LC STN Files: CA, CAPLUS

DT.CA CAplus document type: Patent

RL.P Roles from patents: PREP (Preparation)

Ring System Data

| Elemental | Elemental | . Size | of Ring Syst | em Ring | RID |
|-----------|-----------|---------|---------------|-----------|--------------|
| Analysis | Sequence | the Ri | ngs Formula | Identifie | r Occurrence |
| EA | ES | l SZ | RF | RID | Count |
| | +======= | -+===== | ===+====== | ==+====== | =+======= |
| N3P3 | INPNPNP | 16 | N3P3 | 46.716.5 | 1 in CM |
| | | i | ĺ | i | 12 |

CM 1

CRN 126904-04-9

CMF Unspecified

CCI PMS, MAN

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***

CM 2

CRN 92832-53-6

CMF C36 H54 N3 O18 P3

CM 3

CRN 77001-81-1 CMF C36 H48 N2 O16

PAGE 1-B

1 REFERENCES IN FILE CA (1907 TO DATE)

1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

REFERENCE 1

AN 121:85900 CA

TI Phosphazene-type polymer-coated parts with weatherability and abrasion resistance

IN Mori, Shigeo; Ando, Hiroyuki

PA Idemitsu Petrochemical Co, Japan

SO Jpn. Kokai Tokkyo Koho, 22 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C08J007-04

ICS B32B027-00; C08L085-02

ICA C08F299-02

ICI C08L085-02

CC 42-10 (Coatings, Inks, and Related Products)

Section cross-reference(s): 38

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE
PI JP 06001869 A2 19940111 JP 1992-162381 19920622

PRAI JP 1992-162381 19920622

Title parts comprise substrates, interlayers comprising (1) curable resins having thermal expansion coefficient (E) ≥5 + 10-5 cm-cm-1-°C-1 at 40° or (2) thermoplastic resins having modulus of tensile elasticity ≤2000 MPa, and surface layers comprising curable polyphosphazenes. Thus, 55.0 g hexachlorocyclotriphosphazene was treated with 143 g 2-hydroxyethyl methacrylate in toluene in the presence of pyridine at 80° for 8 h to give 1,1,3,3,5,5-hexa(methacryloyloxyethoxy)cyclotriphosphazene, 40 parts of which was mixed with UA 306H 30, Sartomer 9505 (bifunctional urethane acrylate) 30, colloidal silica 100, Irgacure 184 (I) 1, Tinuvin

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144 1, butanol 50, and Me iso-Bu ketone 80 parts, applied onto a
     transparent polycarbonate substrate with UV-cured precoating containing
     Sartomer 9640 (bifunctional urethane acrylate) 80, Kayarad PEG4000DA 12,
     Kayarad DPHA 3, I 3, a UV absorber 10, and solvents 200 parts showing E
     6.0 + 10-5 cm-cm-1-°C-1, treated at room temperature for 1 min,
     dried at 80° for 10 min, and UV-irradiated to give a 8-\mum
     surface layer showing retention of appearance, adhesion, and color after
     2000-h exposure to weatherometer.
    polyphosphazene coating primer thermal expansion; polycarbonate substrate
    abrasion resistant coating; acrylic polyphosphazene coating; photocurable
    phosphazene polymer coating; tensile elasticity primer polyphosphazene
     coating; weatherability polyphosphazene coating; wear resistance
    polyphosphazene coating; primer acrylic resin polyphosphazene topcoating
    Coating materials
        (abrasion- and heat- and weather-resistant, acrylic polyphosphazenes,
       primers with controlled thermal expansion and tensile elasticity for)
    Urethane polymers, uses
    RL: USES (Uses)
        (acrylic, primers, for polyphosphazene top coatings, with
       weatherability and wear resistance)
     155880-83-4P, 1,1,3,3,5,5-Hexa(methacryloylethylenedioxy)cyclotriphosphaze
    ne-Sartomer 9505-UA 306H copolymer
    RL: PREP (Preparation)
        (preparation of, coatings, with weatherability and wear resistance, primers
        for)
     92832-53-6P
     RL: PREP (Preparation)
        (preparation of, for curable coatings, with wear resistance and
       weatherability)
     9011-87-4P
                 155880-84-5P
                               156738-01-1P
     RL: PREP (Preparation)
        (preparation of, for interlayers for polyphosphazene top coatings, with
       weatherability and wear resistance)
     868-77-9
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with cyclotriphosphazenes, for curable coatings)
     940-71-6, Hexachlorocyclotriphosphazene
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with hydroxy-containing acrylates, for curable phosphazenes)
    ANSWER 6 OF 6 REGISTRY COPYRIGHT 2006 ACS on STN
     92832-54-7 REGISTRY
    Entered STN: 17 Dec 1984
     2-Propenoic acid, 2-methyl-, 1,2-ethanediylbis(oxy-2,1-ethanediyl) ester,
    polymer with 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-
     2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine (9CI) (CA INDEX
     NAME)
OTHER CA INDEX NAMES:
     1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-
     hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-, polymer with
     1,2-ethanediylbis(oxy-2,1-ethanediyl) bis(2-methyl-2-propenoate) (9CI)
OTHER NAMES:
     Hexakis (methacryloyloxyethoxy) cyclotriphosphazene-triethylene glycol
     dimethacrylate copolymer
     (C36 H54 N3 O18 P3 . C14 H22 O6)x
     PMS
     Polyacrylic, Polyphosphazene, Polyphosphazene formed
     STN Files: CA, CAPLUS, USPATFULL
DT.CA CAplus document type: Journal; Patent
       Roles from patents: PREP (Preparation)
RL.P
      Roles from non-patents: BIOL (Biological study); PREP (Preparation);
       PRP (Properties); USES (Uses)
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ST

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ΙT

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ΙT

IT

L3

RN ED

CN

MF

CI

| Elemental | Elemental | l Size of | Ring System | Ring | RID |
|------------------|----------------------|------------|--|--|------------|
| Analysis | Sequence | the Rings | Formula | Identifier | Occurrence |
| EA | ES | l SZ | RF | RID | Count |
| ========
N3P3 | +=======
 NPNPNP | · | +===================================== | +===================================== | • |
| NJEJ | | 1 | | 40.710.5 | 1 111 CM |

CM 1

CRN 92832-53-6 CMF C36 H54 N3 O18 P3

CM 2

CRN 109-16-0 CMF C14 H22 O6

- 4 REFERENCES IN FILE CA (1907 TO DATE)
- 4 REFERENCES IN FILE CAPLUS (1907 TO DATE)

REFERENCE 1

- AN 107:12858 CA
- TI A study on crown and bridge resin using PNC-EMA monomer
- AU Anzai, Misaki; Hoya, Ken; Toriyama, Fumito; Ide, Kozo; Kikuchi, Hisaji; Hirose, Hideharu; Yuda, Masashi; Ohashi, Masayoshi
- CS Sch. Dent., Nihon Univ., Japan
- SO Journal of Nihon University School of Dentistry (1986), 28(4), 240-8 CODEN: JNUDAT; ISSN: 0029-0432
- DT Journal
- LA English
- CC 63-7 (Pharmaceuticals)
- AB The properties of polymer resins for crowns and bridges using a new PNC-EMA monomer [P3N3(OCH2CH2O2CCCH3:CH2)6], mixed with monomers currently available on the market (14-70% weight) and poly(Me methacrylate) (PMMA) as the polymer base, were studied. Both the compressive strength and hardness increased with an increasing amts. of PNC-EMA monomer. Water sorption reached 1.86 mg/cm2 when the amount of PNC-EMA mixed was 28%. The degree of abrasion decreased in accordance with an increase in the amount of PNC-EMA showing a value one-third that of PMMA when mixed in a proportion of 70%.

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ST
     PNC EMA dental bridge crown; methacryloxyethylenedioxycyclotriphosphazene
    dental polymer; cyclotriphosphazene methacryloxyethylenedioxy dental
ΙT
    Dental materials and appliances
        (bridges, methacryloxyethylenedioxycyclotriphosphazene-containing polymers
        for, properties of)
ΙT
     Dental materials and appliances
        (composites, methacryloxyethylenedioxycyclotriphosphazene
        monomer-containing, properties of, for bridges and crowns)
ΙT
     Dental materials and appliances
        (crowns, methacryloxyethylenedioxycyclotriphosphazene-containing polymers
        for, properties of)
ΙT
     80-62-6
                          3253-39-2, 2,2-Bis(4-methacryloxy)phenylpropane
              109-16-0
     RL: BIOL (Biological study)
        (dental resin for crowns and bridges containing acrylic cyclotriphosphazene
        and, properties of)
     92832-53-6
ΙT
     RL: BIOL (Biological study)
        (dental resin for crowns and bridges containing, properties of)
IT
     36936-74-0P
                   92832-54-7P
                                108704-94-5P
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (preparation and properties of, for dental bridges and crowns)
ΙT
    7732-18-5, properties
     RL: PEP (Physical, engineering or chemical process); PROC (Process)
        (sorption of, by acrylic cyclotriphosphazene monomer-containing dental
        polymers)
REFERENCE 2
    102:191109 CA
AN
      Correction of: 101:198144
ΤI
     Synthesis of hexakis (methacryloyloxyethoxy) cyclotriphosphazene and its
    properties and the use as a composite resin
    Anzai, Misaki; Ohashi, Masayoshi
ΑU
     Sch. Dent., Nihon Univ., Tokyo, 101, Japan
CS
     Shika Zairyo, Kikai (1984), 3(3), 401-8
    CODEN: SZKIDA; ISSN: 0286-5858
DT
    Journal
     Japanese
LA
CC
     63-7 (Pharmaceuticals)
     Section cross-reference(s): 29
GΙ
         R = R = CH_2 = CMeCO_2CH_2CH_2O
         `R II, R=Cl
     The title compound (I) [92832-53-6] was prepared as new dental resin monomer.
AB
     Phosphonitrile chloride (II) [1832-07-1] and 2-hydroxyethyl methacrylate
     (HEMA) [868-77-9] were allowed to react at 40° to give I. A mixture
     of approx. 21% (weight) of I, 8.5% triethyleneglycol dimethacrylate and 70%
     Si3N4 were polymerized, and the polymer had a compressive strength, tensile
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strength and transverse strength of 473.6, 33.8 and 80.4 MPa, resp., and hardness HK 96, coefficient of thermal expansion $24.2 \times 106/^{\circ}$ and its

water sorption was 1.2 mg/cm2 at 1 wk.

Dental materials and fillings

12033-89-5, biological studies RL: BIOL (Biological study)

ST

ΙT

methacryloyloxyethoxy cyclotriphosphazene dental

(hexakis (methacryloyloxyethoxy) cyclotriphosphazene for)

```
(dental composite resin containing acrylic cyclotriphosphazene-triethylene
       glycol dimethacrylate copolymer and)
TΤ
    92832-54-7P
     RL: SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological
     study); PREP (Preparation); USES (Uses)
        (preparation of, as dental material)
     92832-53-6P
IT
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (preparation of, for dental resins)
     868-77-9
IT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with hexachlorotriphosphazene)
IT
     1832-07-1
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with hydroxyethyl methacrylate)
REFERENCE 3
AN
    102:32326 CA
     Dental sealants containing phosphazenes
ΤI
PA
    Nihon University, Japan
SO
    Belg., 12 pp.
    CODEN: BEXXAL
DΤ
    Patent
LA
    French
ICI
   A61
CC
    63-7 (Pharmaceuticals)
     Section cross-reference(s): 35
FAN.CNT 1
    PATENT NO.
                     KIND DATE
                                        APPLICATION NO. DATE
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                     ----
                                         _____
                                                         _____
PI
    BE 899654
                    A1 19840831
                                         BE 1984-212932
                                                          19840514
    JP 60038307
                    A2 19850227
                                         JP 1983-147690
                                                          19830811
    JP 03050726
                    B4 19910802
    US 4579880
                    A 19860401
                                         US 1984-603648
                                                          19840424
    CA 1219403
                    A1 19870324
                                         CA 1984-452920
                                                          19840426
                                         FR 1984-7420
    FR 2555440
                    A1 19850531
                                                          19840514
    FR 2555440
                    B1 19880916
                    A1 19850313
                                         GB 1984-12441
                                                          19840516
    GB 2144754
    GB 2144754
                     B2 19861217
                                         DE 1984-3421060 19840606
    DE 3421060
                    A1 19850228
    DE 3421060
                    C2
                           19920820
                                         NL 1984-1961
                                                          19840620
    NL 8401961
                     Α
                           19850301
PRAI JP 1983-147690 19830811
    Dental sealant compns. contain 10-99% polymers prepared from phosphazenes
     containing acrylic groups and alkylene glycol dimethacrylates and silica
     fillers. Thus, 1,1,3,3,5,5-hexakis (methacryloyloxyethoxy) cyclotriphosphaz
     ene (I) [92832-53-6] was prepared by the reaction of hydroxyethyl
     methacrylate [868-77-9] with hexachlorocyclotriphosphazene [940-71-6] in
     anhydrous benzene followed by the addition of pyridine. This compound was
polymerized
     in the presence of Bz202 for 6 h at 60° and for a further 2 h at
     120°. The properties of the resulting polyphosphazene
     [93891-06-6] are tabulated. A polymer for dental sealants was prepared from
     70 parts I, 30 parts triethylene glycol dimethacrylate and 0.3 parts Bz202
     and the properties of the polymer determined The properties of the phosphazene
     polymer such as resistance to compression stability, coefficient of thermal
     dilation, etc., were better than those of the polymers not containing
     phosphazene groups.
    cyclotriphosphazene acrylic dental sealant
ST
IT
     Phosphazene polymers
     RL: BIOL (Biological study)
        (cyclo-, methacrylic, dental sealants containing)
IT
     Dental materials and fillings
        (sealants, methacryloylcyclotriphosphazenes)
```

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ΙT
      7631-86-9, biological studies
      RL: BIOL (Biological study)
         (dental sealant compns. containing methacryloylcyclotriphosphazenes and)
·IT
      92832-53-6P
      RL: RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)
         (preparation and polymerization of, for dental sealant compns.)
      92832-54-7P
 ΙT
                    93891-06-6P
      RL: PREP (Preparation)
         (preparation of, for dental sealant compns.)
      868-77-9
 ΙT
      RL: RCT (Reactant); RACT (Reactant or reagent)
         (reaction of, with hexachlorocyclotriphosphazene)
 TT
      940-71-6
      RL: RCT (Reactant); RACT (Reactant or reagent)
         (reaction of, with hydroxyethyl methacrylate)
 REFERENCE 4
 ΑN
      101:198144 CA
 TΙ
      Synthesis of hexakis (methacryloyloxyethoxy) cyclotriphosphazene and its
      properties and the use as a composite resin
 ΑU
      Anzai, Misaki; Ohashi, Masayoshi
 CS
      Sch. Dent., Nihon Univ., Tokyo, 101, Japan
 SO
      Shika Zairyo, Kikai (1984), 59(5), 401-8
      CODEN: SZKIDA; ISSN: 0286-5858
 \mathsf{DT}
      Journal
      Japanese
 LΑ
 CC
      63-7 (Pharmaceuticals)
      Section cross-reference(s): 29
 GΙ
          R I, R=CH2 == CMeCO2CH2CH2O
          R II, R=Cl
      The title compound (I) [92832-53-6] was prepared as new dental resin monomer.
 AB
      Phosphonitrile chloride (II) [1832-07-1] and 2-hydroxyethyl methacrylate
             [868-77-9] were allowed to react at 40° to give I. A mixture
      of approx. 21% (weight) of I, 8.5% triethyleneglycol dimethacrylate and 70%
      Si3N4 were polymerized, and the polymer had a compressive strength, tensile
      strength and transverse strength of 473.6, 33.8 and 80.4 MPa, resp., and
      hardness HK 96, coefficient of thermal expansion 24.2 + 106/° and
      its water sorption was 1.2 mg/cm2 at 1 wk.
      methacryloyloxyethoxy cyclotriphosphazene dental
 ST
      Dental materials and fillings
 ΙT
         (hexakis (methacryloyloxyethoxy) cyclotriphosphazene for)
 IT
      12033-89-5, biological studies
      RL: BIOL (Biological study)
         (dental composite resin containing acrylic cyclotriphosphazene-triethylene
         glycol dimethacrylate copolymer and)
 IT
      92832-54-7P
      RL: SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological
      study); PREP (Preparation); USES (Uses)
         (preparation of, as dental material)
 ΙT
      92832-53-6P
      RL: SPN (Synthetic preparation); PREP (Preparation)
         (preparation of, for dental resins)
 IT
      RL: RCT (Reactant); RACT (Reactant or reagent)
```

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IΤ
     1832-07-1
     RL: RCT (Reactant); RACT (Reactant or reagent)
         (reaction of, with hydroxyethyl methacrylate)
=> D HIS
     (FILE 'HOME' ENTERED AT 08:07:37 ON 04 SEP 2006)
     FILE 'REGISTRY' ENTERED AT 08:07:48 ON 04 SEP 2006
L1
             492 S PHOSPHAZENE
L2
              31 S L1 AND ACRYL?
               6 S L2 AND ENE
L3
=> S L2 OR L3
             31 L2 OR L3
=> D 1-31
     ANSWER 1 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN
L4
     827320-65-0 REGISTRY
RN
     Entered STN: 08 Feb 2005
ED
     2-Propenoic acid, (1-methyl-1,2-ethanediyl)bis[oxy(methyl-2,1-ethanediyl)]
CN
     ester, polymer with 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-
     1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine (9CI) (CA
     INDEX NAME)
OTHER NAMES:
     Aronix M 220-hexakis (methacryloyloxyethoxy) cyclotriphosphazene
     copolymer
     (C36 H54 N3 O18 P3 . C15 H24 O6)x
MF
CI
PCT
     Polyacrylic, Polyphosphazene, Polyphosphazene formed
SR
LC
     STN Files:
                   CA, CAPLUS
           1 .
     CM
     CRN
          92832-53-6
     CMF C36 H54 N3 O18 P3
       CH<sub>2</sub>
                                                        CH<sub>2</sub>
                    -CH2
                                                       - C-- Me
                                                          CH<sub>2</sub>
    CH<sub>2</sub> O
                                                       0
             CH2-CH2-
                                                     CH<sub>2</sub>
       H<sub>2</sub>C
                 -CH2-CH2-
                                      CH2-CH2-
           2
     СM
```

CRN

CMF

CCI

42978-66-5

C15 H24 O6

IDS

(reaction of, with hexachlorocyclotriphosphazene)

3 (D1-Me)

1 REFERENCES IN FILE CA (1907 TO DATE)
1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 2 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN

RN 827320-64-9 REGISTRY

ED Entered STN: 08 Feb 2005

CN 2-Propenoic acid, 2-hydroxyethyl ester, polymer with 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine (9CI) (CA INDEX NAME)

OTHER NAMES:

CN Hexakis (methacryloyloxyethoxy) cyclotriphosphazene-2-hydroxyethyl acrylate copolymer

MF (C36 H54 N3 O18 P3 . C5 H8 O3) \times

CI PMS

PCT Polyacrylic, Polyphosphazene, Polyphosphazene formed

SR CA

LC STN Files: CA, CAPLUS

CM 1

CRN 92832-53-6 CMF C36 H54 N3 O18 P3

CM 2

CRN 818-61-1 CMF C5 H8 O3

1 REFERENCES IN FILE CA (1907 TO DATE)

1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 3 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN RN 794574-57-5 REGISTRY

ED Entered STN: 08 Dec 2004

CN 2-Propenamide, N-(1-methylethyl)-, polymer with 2,2,4,4,6,6-hexachloro-2,2,4,4,6,6-hexahydro-1,3,5,2,4,6-triazatriphosphorine, graft (9CI) (CA INDEX NAME)

OTHER NAMES:

CN Hexachlorocyclotriphosphazene-isopropylacrylamide graft copolymer

MF (C6 H11 N O . C16 N3 P3)x

CI PMS

PCT Polyacrylic, Polyphosphazene, Polyphosphazene formed

SR CA

LC STN Files: CA, CAPLUS

CM 1

CRN 2210-25-5 CMF C6 H11 N O

CM 2

CRN 940-71-6 CMF C16 N3 P3

4 REFERENCES IN FILE CA (1907 TO DATE)

4 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA

4 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 4 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN

RN 727704-44-1 REGISTRY

ED Entered STN: 17 Aug 2004

CN 1,4-Benzenedicarboxylic acid, polymer with 1,2-ethanediol, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine, 1,6-hexanediol, 2-hydroxypropyl 2-propenoate, 1,1'-methylenebis[isocyanatocyclohexane] and methyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

OTHER NAMES:

CN Ethylene glycol-1,1,3,3,5,5-hexakis(acryloylethylenedioxy)cyclotripho sphazene-1,6-hexanediol-2-hydroxypropyl acrylate-methylenebis(cyclohexyl isocyanate)-methyl methacrylate-terephthalic acid copolymer

MF (C30 H42 N3 O18 P3 . C15 H22 N2 O2 . C8 H6 O4 . C6 H14 O2 . C6 H10 O3 . C5 H8 O2 . C2 H6 O2) \times

CI PMS

PCT Polyacrylic, Polyamide, Polyamide formed, Polyester, Polyester formed, Polyphosphazene, Polyphosphazene formed

SR CA

LC STN Files: CA, CAPLUS

CM 1

CRN 137741-82-3 CMF C30 H42 N3 O18 P3

CM 2

CRN 28605-81-4 CMF C15 H22 N2 O2 CCI IDS



CM 3

CRN 999-61-1 CMF C6 H10 O3

$$\begin{array}{c|c} \text{OH} & \text{O} \\ | & || \\ \text{Me-CH-CH}_2\text{-O-C-CH} \end{array}$$

CM 4

CRN 629-11-8 CMF C6 H14 O2

$$HO-(CH_2)_6-OH$$

CM 5

CRN 107-21-1

 ${\rm HO-CH_2-CH_2-OH}$

CM 6

CRN 100-21-0 CMF C8 H6 O4

CM 7

CRN 80-62-6 CMF C5 H8 O2

1 REFERENCES IN FILE CA (1907 TO DATE)
1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 5 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN

RN 727704-34-9 REGISTRY

ED Entered STN: 17 Aug 2004

CN 2-Propenoic acid, 2-methyl-, 2-hydroxyethyl ester, polymer with 2,4-diisocyanato-1-methylbenzene, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine, [(1-methyl-1,2-ethanediyl)bis(oxy)]bis[propanol] and methyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

OTHER NAMES:

CN 1,1,3,3,5,5-Hexakis(acryloylethylenedioxy)cyclotriphosphazene-2-hydroxyethyl methacrylate-methyl methacrylate-2,4-TDI-tripropylene glycol copolymer

MF (C30 H42 N3 O18 P3 . C9 H20 Q4 . C9 H6 N2 O2 . C6 H10 O3 . C5 H8 O2)x

CI PMS

PCT Polyacrylic, Polyether, Polyphosphazene, Polyphosphazene formed, Polyurethane, Polyurethane formed

SR CA

LC STN Files: CA, CAPLUS

CM 1

CRN 137741-82-3

CMF C30 H42 N3 O18 P3

CRN 24800-44-0 CMF C9 H20 O4

CCI IDS

 ${\tt HO-CH_2-CH_2-O-CH_2-CH_2-O-CH_2-CH_2-OH}$

CM 3

CRN 868-77-9 CMF C6 H10 O3

CM 4

CRN 584-84-9 CMF C9 H6 N2 O2

CM 5

CRN 80-62-6 CMF C5 H8 O2

1 REFERENCES IN FILE CA (1907 TO DATE) 1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 6 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN

RN .552890-90-1 REGISTRY

ED Entered STN: 23 Jul 2003

CN 2-Propenamide, N-(1-methylethyl)-, polymer with 2-(4-ethenylphenoxy)-2,4,4,6,6-pentakis[4-(ethoxycarbonyl)phenoxy]-2,2,4,4,6,6-hexahydro-1,3,5,2,4,6-triazatriphosphorine (9CI) (CA INDEX NAME)

OTHER NAMES:

CN 2,4,4,6,6-Pentakis(4-ethoxycarbonylphenoxy)-2-(4-vinylphenoxy)cyclotriphosphazene-N-isopropylacrylamide copolymer (C53 H52 N3 O16 P3 . C6 H11 N O)x

CI PMS

PCT Polyacrylic, Polyphosphazene, Polyphosphazene formed, Polystyrene

SR CA

LC STN Files: CA, CAPLUS

CM 1

CRN 552890-89-8 CMF C53 H52 N3 O16 P3

PAGE 1-A

CRN 2210-25-5 CMF C6 H11 N O

- 1 REFERENCES IN FILE CA (1907 TO DATE)
- 1 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA
- 1 REFERENCES IN FILE CAPLUS (1907 TO DATE)
- L4 ANSWER 7 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN
- RN 273221-38-8 REGISTRY
- ED Entered STN: 27 Jun 2000
- CN Phosphonium, [(4-ethenylphenyl)methyl]trioctyl-, chloride, polymer with 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES:

CN 1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-, polymer with [(4-ethenylphenyl)methyl]trioctylphosphonium chloride (9CI)

OTHER NAMES:

- CN Hexakis(methacryloyloxyethoxy)cyclotriphosphazene-trioctyl(4-vinylbenzyl)phosphonium chloride) copolymer
- MF (C36 H54 N3 O18 P3 . C33 H60 P . C1)x
- CI PMS
- PCT Polyacrylic, Polyphosphazene, Polyphosphazene formed, Polystyrene
- SR CA
- LC STN Files: CA, CAPLUS

CM 1

CRN 92832-53-6

CMF C36 H54 N3 O18 P3

CRN 74443-79-1 (762190-78-3) CMF C33 H60 P . C1

• c1-

1 REFERENCES IN FILE CA (1907 TO DATE)
1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 8 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN

RN 272119-38-7 REGISTRY

ED Entered STN: 22 Jun 2000

CN 2-Propenoic acid, 2-ethyl-2-[[(1-oxo-2-propenyl)oxy]methyl]-1,3-propanediyl ester, polymer with N,N-dimethyl-2-propenamide, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine and α -(1-oxo-2-propenyl)- ω -[(1-oxo-2-propenyl)oxy]poly(oxy-1,2-ethanediyl) (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES:

CN 1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-, polymer with N,N-dimethyl-2-propenamide, 2-ethyl-2-[[(1-oxo-2-propenyl)oxy]methyl]-1,3-propanediyl di-2-propenoate and α -(1-oxo-2-propenyl)- ω -[(1-oxo-2-propenyl)oxy]poly(oxy-1,2-ethanediyl) (9CI)

CN 2-Propenamide, N,N-dimethyl-, polymer with 2-ethyl-2-[[(1-oxo-2-propenyl)oxy]methyl]-1,3-propanediyl di-2-propenoate, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine and α -(1-oxo-2-propenyl)- ω -[(1-oxo-2-propenyl)oxy]poly(oxy-1,2-ethanediyl) (9CI)

CN Poly(oxy-1,2-ethanediyl), α-(1-oxo-2-propenyl)-ω-[(1-oxo-2-propenyl)oxy]-, polymer with N,N-dimethyl-2-propenamide,
 2-ethyl-2-[[(1-oxo-2-propenyl)oxy]methyl]-1,3-propanediyl di-2-propenoate
 and 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine (9CI)
OTHER NAMES:

CN N, N-Dimethylacrylamide-hexakis (methacryloyloxyethoxy) cyclotriphosphaz ene-NK Ester A 600-trimethylolpropane triacrylate copolymer

MF (C36 H54 N3 O18 P3 . C15 H20 O6 . C5 H9 N O . (C2 H4 O) \hat{n} C6 H6 O3) x

CI PMS

PCT Polyacrylic, Polyether, Polyphosphazene, Polyphosphazene formed

SR CA

LC STN Files: CA, CAPLUS

CM 1

CRN 92832-53-6 CMF C36 H54 N3 O18 P3

CRN 26570-48-9

CMF (C2 H4 O)n C6 H6 O3

CCI PMS

$$H_2C = CH - C - CH_2 - CH_2$$

CM 3

CRN 15625-89-5 CMF C15 H20 O6

CM 4

CRN 2680-03-7 CMF C5 H9 N O

1 REFERENCES IN FILE CA (1907 TO DATE)
1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 9 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN

RN · 272119-37-6 REGISTRY

ED Entered STN: 22 Jun 2000

CN 2-Propenoic acid, 2-methyl-, 3-(trimethoxysilyl)propyl ester, polymer with

2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine and 2-methyl-2-[(1-oxo-2-propenyl)amino]-1-propanesulfonic acid monosodium salt (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES:

CN 1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-, polymer with 2-methyl-2-[(1-oxo-2-propenyl)amino]-1-propanesulfonic acid monosodium salt and 3-(trimethoxysilyl)propyl 2-methyl-2-propenoate (9CI)

CN 1-Propanesulfonic acid, 2-methyl-2-[(1-oxo-2-propenyl)amino]-, monosodium salt, polymer with 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine and 3-(trimethoxysilyl)propyl 2-methyl-2-propenoate (9CI)

OTHER NAMES:

CN Hexakis (methacryloyloxyethoxy) cyclotriphosphazene-KBM 503-sodium 2-acrylamido-2-methylpropanesulfonate copolymer

MF (C36 H54 N3 O18 P3 . C10 H20 O5 Si . C7 H13 N O4 S . Na)x

CI PMS

PCT Polyacrylic, Polyphosphazene, Polyphosphazene formed

SR CA

LC STN Files: CA, CAPLUS

CM 1

CRN 92832-53-6 CMF C36 H54 N3 O18 P3

CM 2

CRN 5165-97-9 (15214-89-8) CMF C7 H13 N O4 S . Na

● Na

1 REFERENCES IN FILE CA (1907 TO DATE)
1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 10 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN

RN 272119-36-5 REGISTRY

ED Entered STN: 22 Jun 2000

CN 2-Propenoic acid, 2-methyl-, 3-(trimethoxysilyl)propyl ester, polymer with 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine and 3,3,4,4,5,5,6,6,7,7,8,8,8-tridecafluorooctyl 2-propenoate (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES:

CN 1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-, polymer with 3,3,4,4,5,5,6,6,7,7,8,8,8-tridecafluoroctyl 2-propenoate and 3-(trimethoxysilyl)propyl 2-methyl-2-propenoate (9CI)

CN 2-Propenoic acid, 3,3,4,4,5,5,6,6,7,7,8,8,8-tridecafluorooctyl ester, polymer with 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine and 3-(trimethoxysilyl)propyl 2-methyl-2-propenoate (9CI)

OTHER NAMES:

CN Hexakis (methacryloyloxyethoxy) cyclotriphosphazene-KBM 503-2-(perfluorohexyl) ethyl acrylate copolymer

MF (C36 H54 N3 O18 P3 . C11 H7 F13 O2 . C10 H20 O5 Si)x

CI PMS

PCT Polyacrylic, Polyphosphazene, Polyphosphazene formed

SR CA

LC STN Files: CA, CAPLUS

CM 1

CRN 92832-53-6 CMF C36 H54 N3 O18 P3

propenyl)oxy]poly(oxy-1,2-ethanediyl), sodium 2-methyl-2-propenoate and

ethanediyl) and 3-(trimethoxysilyl)propyl 2-methyl-2-propenoate (9CI)

Hexakis (methacryloyloxyethoxy) cyclotriphosphazene-KBM 503-NK Ester A 600-sodium methacrylate-trimethylolpropane triacrylate copolymer

2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-

Poly(oxy-1,2-ethanediy1), α -(1-oxo-2-propeny1)- ω -[(1-oxo-2-

sodium 2-methyl-2-propenoate and 3-(trimethoxysilyl)propyl

2-methyl-2-propenoate (9CI)

2-ethyl-2-[[(1-oxo-2-propenyl)oxy]methyl]-1,3-propanediyl di-2-propenoate,

propenyl)oxy]-, polymer with 2-ethyl-2-[[(1-oxo-2-propenyl)oxy]methyl]-1,3propanediyl di-2-propenoate, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine,

(C36 H54 N3 O18 P3 . C15 H20 O6 . C10 H20 O5 Si . C4 H6 O2 . (C2 H4 O)n C6

3-(trimethoxysily1)propyl 2-methyl-2-propenoate (9CI) 2-Propenoic acid, 2-methyl-, sodium salt, polymer with

propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine, α -(1-oxo-2-propenyl)- ω -[(1-oxo-2-propenyl)oxy]poly(oxy-1,2-

CN

MF

OTHER NAMES:

H6 O3 . Na)x

(CA INDEX NAME)

CI PMS

PCT Polyacrylic, Polyether, Polyphosphazene, Polyphosphazene formed

SR CA

LC STN Files: CA, CAPLUS

CM 1

CRN 92832-53-6

CMF C36 H54 N3 O18 P3

CM 2

CRN 26570-48-9

CMF (C2 H4 O)n C6 H6 O3

CCI PMS

$$H_2C = CH - C - CH_2 - CH_2$$

CM 3

CRN 15625-89-5 CMF C15 H20 O6

CM 4

CRN 5536-61-8 (79-41-4)

CMF C4 H6 O2 . Na

🕨 Na

CM 5

CRN 2530-85-0 CMF C10 H20 O5 Si

1 REFERENCES IN FILE CA (1907 TO DATE)
1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

- L4 ANSWER 12 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN
- RN 272119-33-2 REGISTRY
- ED Entered STN: 22 Jun 2000
- CN 1-Propanesulfonic acid, 2-methyl-2-[(1-oxo-2-propenyl)amino]-, monosodium salt, polymer with 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES:

CN 1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-, polymer with 2-methyl-2-[(1-oxo-2-propenyl)amino]-1-propanesulfonic acid monosodium salt (9CI)

OTHER NAMES:

- CN Hexakis(methacryloyloxyethoxy)cyclotriphosphazene-sodium 2-acrylamido-2-methylpropanesulfonate copolymer
- MF (C36 H54 N3 O18 P3 . C7 H13 N O4 S . Na)x
- CI PMS
- PCT Polyacrylic, Polyphosphazene, Polyphosphazene formed
- SR CA
- LC STN Files: CA, CAPLUS

CM 1

CRN 92832-53-6

CMF C36 H54 N3 O18 P3

CRN 5165-97-9 (15214-89-8) CMF C7 H13 N O4 S . Na

Na

1 REFERENCES IN FILE CA (1907 TO DATE)
1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 13 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN

RN 272119-32-1 REGISTRY

ED Entered STN: 22 Jun 2000

CN 2-Propenamide, N,N-dimethyl-, polymer with 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES:

CN 1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-, polymer with N,N-dimethyl-2-propenamide (9CI)

OTHER NAMES:

CN N, N-Dimethylacrylamide-hexakis (methacryloyloxyethoxy) cyclotriphosphaz ene copolymer

MF (C36 H54 N3 O18 P3 . C5 H9 N O)x

CI PMS

PCT Polyacrylic, Polyphosphazene, Polyphosphazene formed

SR CA

LC STN Files: CA, CAPLUS

CM 1

CRN 92832-53-6

CMF C36 H54 N3 O18 P3

```
CH<sub>2</sub>
                                                    CH2
            O- CH2- CH2-
                                     CH2-CH2-
                                                      CH<sub>2</sub>
   CH<sub>2</sub>
       0
                 CH2-CH2
                                       -CH2-
     CM
          2
     CRN
          2680-03-7
     CMF
          C5 H9 N O
     0
Me2N-C-CH-CH2
               1 REFERENCES IN FILE CA (1907 TO DATE)
               1 REFERENCES IN FILE CAPLUS (1907 TO DATE)
     ANSWER 14 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN
L4
     255384-20-4 REGISTRY
RN
     Entered STN: 09 Feb 2000
ED
CN
     2-Propenoic acid, 2-methyl-, 2-(hydroxymethyl)-2-[[(2-methyl-1-oxo-2-
     propenyl)oxy]methyl]-1,3-propanediyl ester, polymer with
     2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-
     propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine and
     (1-methylethylidene)bis[4,1-phenyleneoxy(2-hydroxy-3,1-propanediyl)]
     di-2-propenoate (9CI) (CA INDEX NAME)
OTHER CA INDEX NAMES:
     1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-
     hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-, polymer with
     2-(hydroxymethyl)-2-[((2-methyl-1-oxo-2-propenyl)oxy]methyl]-1,3-
     propanediyl bis(2-methyl-2-propenoate) and (1-methylethylidene)bis[4,1-
     phenyleneoxy(2-hydroxy-3,1-propanediyl)] di-2-propenoate (9CI)
     2-Propenoic acid, (1-methylethylidene)bis[4,1-phenyleneoxy(2-hydroxy-
CN
     3,1-propanediyl)] ester, polymer with 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-
     hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-
     triazatriphosphorine and 2-(hydroxymethyl)-2-[[(2-methyl-1-oxo-2-
     propenyl)oxy]methyl]-1,3-propanediyl bis(2-methyl-2-propenoate) (9CI)
OTHER NAMES:
    Hexa(2-methacryloyloxyethoxy)cyclotriphosphazene-pentaerythritol
     trimethacrylate-Ripoxy SP 1509 copolymer
MF
     (C36 H54 N3 O18 P3 . C27 H32 O8 . C17 H24 O7)x
CI
PCT
     Polyacrylic, Polyether, Polyphosphazene, Polyphosphazene formed
SR
LC
                  CA, CAPLUS, USPATFULL
     STN Files:
```

CRN 92832-53-6 CMF C36 H54 N3 O18 P3

CM

1

CRN 4687-94-9 CMF C27 H32 O8

PAGE 1-B

CM 3

CRN 3524-66-1 CMF C17 H24 O7

2 REFERENCES IN FILE CA (1907 TO DATE)
2 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 15 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN

RN 212397-70-1 REGISTRY

ED Entered STN: 08 Oct 1998

CN Benzenecarboperoxoic acid, 4,4'-carbonylbis[[[2-[(2-methyl-1-oxo-2-

propenyl)oxy]ethoxy]carbonyl]-, bis(1,1-dimethylethyl) ester, polymer with 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine and α -(1-oxo-2-propenyl)- ω -[(1-oxo-2-propenyl)oxy]poly(oxy-1,2ethanediyl), graft (9CI) (CA INDEX NAME) OTHER CA INDEX NAMES: 1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-, polymer with bis(1,1-dimethylethyl) 4,4'-carbonylbis[[[2-[(2-methyl-1-oxo-2propenyl)oxy]ethoxy]carbonyl]benzenecarboperoxoate] and α -(1-oxo-2-propenyl)- ω -[(1-oxo-2-propenyl)oxy]poly(oxy-1,2ethanediyl), graft (9CI) Poly(oxy-1,2-ethanediyl), α -(1-oxo-2-propenyl)- ω -[(1-oxo-2-CN propenyl)oxy]-, polymer with bis(1,1-dimethylethyl) 4,4'-carbonylbis[[[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]carbonyl]benzenecarboperoxoate] and 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine, graft (9CI) OTHER NAMES: 3,3'-Bis(methacryloyloxyethoxycarbonyl)-4,4'-bis(tert-CN butylperoxycarbonyl)benzophenone-polyethylene glycol diacrylatehexakis(methacryloyloxyethoxy)cyclotriphosphazene graft copolymer (C37 H42 O15 . C36 H54 N3 O18 P3 . (C2 H4 O)n C6 H6 O3) \times MF CI Polyacrylic, Polyether, Polyphosphazene, Polyphosphazene formed PCT SR CA, CAPLUS LC STN Files: CM 1 CRN 204009-97-2

CM 2

CRN 92832-53-6

CMF C37 H42 O15

CMF C36 H54 N3 O18 P3

CRN 26570-48-9

CMF (C2 H4 O)n C6 H6 O3

CCI PMS

$$H_2C = CH - C - CH_2 - CH_2$$

1 REFERENCES IN FILE CA (1907 TO DATE)

1 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA

1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 16 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN

RN 211051-32-0 REGISTRY

ED Entered STN: 10 Sep 1998

CN 2-Propenoic acid, 2-methyl-, methyl ester, polymer with 2,2,4,4-tetrachloro-6-[(ferrocenylmethyl)methylamino]-2,2,4,4,6,6-hexahydro-6-[4-[(2-methyl-1-oxo-2-propenyl)oxy]butoxy]-1,3,5,2,4,6-triazatriphosphorine (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES:

CN 1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4-tetrachloro-6[(ferrocenylmethyl)methylamino]-2,2,4,4,6,6-hexahydro-6-[4-[(2-methyl-1-oxo-2-propenyl)oxy]butoxy]-, polymer with methyl 2-methyl-2-propenoate (9CI)

OTHER NAMES:

CN Methyl methacrylate-2,2,4,4-tetrachloro-2-(ferrocenylmethyl)(methyl)amino-6-(4-(2-methyl-2propenoyloxy)butoxy)cyclotriphosphazene copolymer

MF (C20 H27 C14 Fe N4 O3 P3 . C5 H8 O2)x

CI PMS

PCT Polyacrylic, Polyphosphazene, Polyphosphazene formed

SR CA

LC STN Files: CA, CAPLUS

CM 1

CRN 211051-09-1

CMF C20 H27 C14 Fe N4 O3 P3

CCI CCS

HC
$$H = H = C = CH_2 = N = P = C1$$
 $CH_2 = CH_2 = N = P = C1$
 $CH_2 = CH_2 = CH_2 = N = CH_2 = CH$

CRN 80-62-6 CMF C5 H8 O2

1 REFERENCES IN FILE CA (1907 TO DATE)
1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 17 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN

RN 200334-91-4 REGISTRY

ED Entered STN: 27 Jan 1998

CN Ethenol, polymer with 2,2,4,4,6-pentakis(4-ethylphenoxy)-2,2,4,4,6,6-hexahydro-6-[4-[(2-methyl-1-oxo-2-propenyl)oxy]phenoxy]-1,3,5,2,4,6-triazatriphosphorine, graft (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES:

CN 1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6-pentakis(4-ethylphenoxy)-2,2,4,4,6,6-hexahydro-6-[4-[(2-methyl-1-oxo-2-propenyl)oxy]phenoxy]-, polymer with ethenol, graft (9CI)

OTHER NAMES:

CN (4-Methacryloyloxy) pentakis (4-ethylphenoxy) cyclotriphosphazene-vinyl alcohol graft copolymer .

MF (C50 H54 N3 O8 P3 . C2 H4 O) \times

CI PMS

PCT Polyacrylic, Polyphosphazene, Polyphosphazene formed, Polyvinyl

SR CA

LC STN Files: CA, CAPLUS

CM 1

CRN 200334-89-0 CMF C50 H54 N3 O8 P3

CRN 557-75-5 CMF C2 H4 O

 $H_2C = CH - OH$

1 REFERENCES IN FILE CA (1907 TO DATE)
1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 18 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN

RN 200334-90-3 REGISTRY

ED Entered STN: 27 Jan 1998

CN 1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6-pentakis(4-ethylphenoxy)-2,2,4,4,6,6-hexahydro-6-[4-[(2-methyl-1-oxo-2-propenyl)oxy]phenoxy]-, homopolymer (9CI) (CA INDEX NAME)

OTHER NAMES:

CN (4-Methacryloyloxy)pentakis(4-ethylphenoxy)cyclotriphosphazene homopolymer

MF (C50 H54 N3 O8 P3)x

CI PMS

PCT Polyacrylic, Polyphosphazene, Polyphosphazene formed

SR CA

LC STN Files: CA, CAPLUS

CM 1

CRN 200334-89-0 CMF C50 H54 N3 O8 P3

1 REFERENCES IN FILE CA (1907 TO DATE)
1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 19 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN

RN 200334-89-0 REGISTRY

ED Entered STN: 27 Jan 1998

CN 1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6-pentakis(4-ethylphenoxy)-2,2,4,4,6,6-hexahydro-6-[4-[(2-methyl-1-oxo-2-propenyl)oxy]phenoxy]-(9CI) (CA INDEX NAME)

OTHER NAMES:

CN (4-Methacryloyloxy)pentakis(4-ethylphenoxy)cyclotriphosphazene

MF C50 H54 N3 08 P3

CI COM

SR CA

LC STN Files: CA, CAPLUS

***PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT**

1 REFERENCES IN FILE CA (1907 TO DATE)

1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 20 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN

RN 192944-35-7 REGISTRY

ED Entered STN: 22 Aug 1997

CN 2-Propenoic acid, ethyl ester, polymer with 2,2,4,4,6-pentachloro-6-[(4'-ethenyl[1,1'-biphenyl]-4-yl)oxy]-2,2,4,4,6,6-hexahydro-1,3,5,2,4,6-triazatriphosphorine (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES:

CN 1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6-pentachloro-6-[(4'-ethenyl[1,1'-biphenyl]-4-yl)oxy]-2,2,4,4,6,6-hexahydro-, polymer with ethyl 2-propenoate (9CI)

OTHER NAMES:

'CN Ethyl acrylate-2-(4'-vinyl-4-biphenylyloxy)pentachlorocyclotriphospha zene copolymer

MF (C14 H11 C15 N3 O P3 . C5 H8 O2)x

CI PMS

PCT Polyacrylic, Polyphosphazene, Polyphosphazene formed, Polystyrene

SR CA

LC STN Files: CA, CAPLUS

CM 1

CRN 115529-67-4

CMF C14 H11 C15 N3 O P3

$$\begin{array}{c|c}
C1 & C1 \\
\hline
O & P & N \\
\hline
N & P & C1 \\
\hline
C1 & C1 \\
\hline
C1 & C1
\end{array}$$

CRN 140-88-5 CMF C5 H8 O2

1 REFERENCES IN FILE CA (1907 TO DATE)
1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 21 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN

RN 192944-34-6 REGISTRY

ED Entered STN: 22 Aug 1997

CN 2-Propenoic acid, methyl ester, polymer with 2,2,4,4,6-pentachloro-6-[(4'-ethenyl[1,1'-biphenyl]-4-yl)oxy]-2,2,4,4,6,6-hexahydro-1,3,5,2,4,6-triazatriphosphorine (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES:

CN 1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6-pentachloro-6-[(4'-ethenyl[1,1'-biphenyl]-4-yl)oxy]-2,2,4,4,6,6-hexahydro-, polymer with methyl 2-propenoate (9CI)

OTHER NAMES:

CN Methyl acrylate-2-(4'-vinyl-4-biphenylyloxy)pentachlorocyclotriphosph azene copolymer

MF (C14 H11 C15 N3 O P3 . C4 H6 O2) \times

CI PMS

PCT Polyacrylic, Polyphosphazene, Polyphosphazene formed, Polystyrene

SR CA

LC STN Files: CA, CAPLUS

CM 1

CRN 115529-67-4

CMF C14 H11 C15 N3 O P3

$$\begin{array}{c|c} C1 & C1 \\ \hline \\ O & P \\ \hline \\ N & N \\ \hline \\ C1 & C1 \\ \hline \\ \\ C1 & C1 \\ \end{array}$$

CM 2

CRN 96-33-3 CMF C4 H6 O2

1 REFERENCES IN FILE CA (1907 TO DATE)

1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

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·ANSWER 22 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN
L4
RN
     181940-31-8 REGISTRY
     Entered STN: 15 Oct 1996
ED
     2-Propenoic acid, 2-methyl-, methyl ester, polymer with
CN
     2,2,4,4,6-pentachloro-2,2,4,4,6,6-hexahydro-6-[2-(2-propenyl)phenoxy]-
     1,3,5,2,4,6-triazatriphosphorine (9CI) (CA INDEX NAME)
OTHER CA INDEX NAMES:
     1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6-pentachloro-2,2,4,4,6,6-
     hexahydro-6-[2-(2-propenyl)phenoxy]-, polymer with methyl
     2-methyl-2-propenoate (9CI)
OTHER NAMES:
     2-Allylphenoxypentachlorocyclotriphosphazene-methyl methacrylate
CN
     copolymer
MF
     (C9 H9 C15 N3 O P3 . C5 H8 O2)x
CI
PCT
     Polyacrylic, Polyphosphazene, Polyphosphazene formed, Polyvinyl
SR
                  CA, CAPLUS
LC
     STN Files:
     CM
          1
     CRN
          137856-38-3
     CMF C9 H9 C15 N3 O P3
    H_2C = CH
          CH<sub>2</sub>
          Cl
          2
     CM
     CRN
          80-62-6
     CMF C5 H8 O2
 H<sub>2</sub>C
      0
      \parallel
Me-C-C-OMe
             . 3 REFERENCES IN FILE CA (1907 TO DATE)
               3 REFERENCES IN FILE CAPLUS (1907 TO DATE)
L4
     ANSWER 23 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN
     180090-93-1 REGISTRY
RN
ED
     Entered STN: 27 Aug 1996
CN
     2-Propenoic acid, 2-methyl-, methyl ester, polymer with
     2,2,4,4,6,6-hexahydro-2-[2-(2-propenyl)phenoxy]-2,4,4,6,6-pentakis(2,2,2-
     trifluoroethoxy)-1,3,5,2,4,6-triazatriphosphorine (9CI) (CA INDEX NAME)
OTHER CA INDEX NAMES:
```

1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6,6-hexahydro-2-[2-(2-

propenyl)phenoxy]-2,4,4,6,6-pentakis(2,2,2-trifluoroethoxy)-, polymer with

methyl 2-methyl-2-propenoate (9CI)

OTHER NAMES:

CN 2-Allylphenoxypenta(2,2,2-trifluoroethoxy)cyclotriphosphazene-methyl methacrylate copolymer

MF (C19 H19 F15 N3 O6 P3 . C5 H8 O2)x

CI PMS

PCT Polyacrylic, Polyphosphazene, Polyphosphazene formed, Polyvinyl

SR CA

LC STN Files: CA, CAPLUS

CM 1

CRN 180090-88-4

CMF C19 H19 F15 N3 O6 P3

CM 2

CRN 80-62-6 CMF C5 H8 O2

$$\begin{array}{ccc} ^{H_2C} & \text{O} \\ & \parallel & \parallel \\ \text{Me-} & \text{C-} & \text{C-} & \text{OMe} \end{array}$$

3 REFERENCES IN FILE CA (1907 TO DATE)

3 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 24 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN

RN 180090-90-8 REGISTRY

ED Entered STN: 27 Aug 1996

CN 2-Propenoic acid, 2-methyl-, methyl ester, polymer with

2,2,4,4,6,6-hexahydro-2,2,4,4,6-pentaphenoxy-6-[2-(2-propenyl)phenoxy]-

1,3,5,2,4,6-triazatriphosphorine (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES:

CN 1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6,6-hexahydro-2,2,4,4,6pentaphenoxy-6-[2-(2-propenyl)phenoxy]-, polymer with methyl
2-methyl-2-propenoate (9CI)

OTHER NAMES:

CN 2-Allylphenoxypentaphenoxycyclotriphosphazene-methyl methacrylate copolymer

MF (C39 H34 N3 O6 P3 . C5 H8 O2)x

CI PMS

PCT Polyacrylic, Polyphosphazene, Polyphosphazene formed, Polyvinyl

SR CA

LC STN Files: CA, CAPLUS

CM 1

CRN 137856-40-7 CMF C39 H34 N3 O6 P3

CM 2

CRN 80-62-6 CMF C5 H8 O2

$$^{\text{H}_2\text{C}}_{\|}$$
 $^{\text{O}}_{\|}$ $^{\text{Me}-\text{C}-\text{C}-\text{OMe}}$

3 REFERENCES IN FILE CA (1907 TO DATE)
3 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 25 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN

RN 171729-46-7 REGISTRY

ED Entered STN: 22 Dec 1995

CN 2-Propenamide, N-(1-methylethyl)-, polymer with 1-ethenyl-2-pyrrolidinone and 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethyl]-1,3,5,2,4,6-triazatriphosphorine (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES:

CN 1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethyl]-, polymer with 1-ethenyl-2-pyrrolidinone and N-(1-methylethyl)-2-propenamide (9CI)

CN 2-Pyrrolidinone, 1-ethenyl-, polymer with 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethyl]-1,3,5,2,4,6triazatriphosphorine and N-(1-methylethyl)-2-propenamide (9CI)

OTHER NAMES:

CN Hexakis(2-methacryloxyethyl)cyclotriphosphazene-N-isopropylacrylamide-N-vinyl-2-pyrrolidone copolymer

MF (C36 H54 N3 O12 P3 . C6 H11 N O . C6 H9 N O)x

CI PMS

PCT Polyacrylic, Polyphosphazene, Polyphosphazene formed, Polyvinyl

SR CA

LC STN Files: CA, CAPLUS

CM 1

CRN 171729-45-6

CRN 2210-25-5 CMF C6 H11 N O

CM 3

CRN 88-12-0 CMF C6 H9 N O

1 REFERENCES IN FILE CA (1907 TO DATE)
1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 26 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN

RN 155880-83-4 REGISTRY

ED Entered STN: 22 Jun 1994

CN 11,15-Dioxa-2,9-diazaoctadec-17-enoic acid, 10,16-dioxo-13,13-bis[[(1-oxo-2-propenyl)oxy]methyl]-, 3-[(1-oxo-2-propenyl)oxy]-2,2-bis[[(1-oxo-2-propenyl)oxy]methyl]propyl ester, polymer with 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine and Sartomer C 9505 (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES:

CN 1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-, polymer with 3-[(1-oxo-2-propenyl)oxy]-2,2-bis[[(1-oxo-2-propenyl)oxy]methyl]propyl 10,16-dioxo-13,13-bis[[(1-oxo-2-propenyl)oxy]methyl]-11,15-dioxa-2,9-diazaoctadec-17-enoate and Sartomer C 9505 (9CI)

CN Sartomer C 9505, polymer with 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine and 3-[(1-oxo-2-propenyl)oxy]-2,2-bis[[(1-oxo-2-propenyl)oxy]methyl]propyl 10,16-dioxo-13,13-bis[[(1-oxo-2-propenyl)oxy]methyl]-11,15-dioxa-2,9-diazaoctadec-17-enoate (9CI)

OTHER NAMES:

CN 1,1,3,3,5,5-hexa(methacryloylethylenedioxy)cyclotriphosphazene-Sartomer 9505-UA 306H copolymer

MF (C36 H54 N3 O18 P3 . C36 H48 N2 O16 . Unspecified)x

CI PMS

PCT Manual component, Polyacrylic, Polyother, Polyphosphazene, Polyphosphazene formed

SR CA

LC STN Files: CA, CAPLUS

CM 1

CRN 126904-04-9 CMF Unspecified CCI PMS, MAN

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***

CM 2

CRN 92832-53-6 CMF C36 H54 N3 O18 P3

CM 3

CRN 77001-81-1 CMF C36 H48 N2 O16

1 REFERENCES IN FILE CA (1907 TO DATE)
1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 27 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN

RN 122847-49-8 REGISTRY

ED Entered STN: 22 Sep 1989

CN 2-Propenoic acid, 2-methyl-, methyl ester, polymer with 2,2,4;4,6-pentachloro-6-[(4'-ethenyl[1,1'-biphenyl]-4-yl)oxy]-2,2,4,4,6,6-hexahydro-1,3,5,2,4,6-triazatriphosphorine (9CI) (CA INDEX NAME) OTHER CA INDEX NAMES:

CN 1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6-pentachloro-6-[(4'-ethenyl[1,1'-biphenyl]-4-yl)oxy]-2,2,4,4,6,6-hexahydro-, polymer with methyl 2-methyl-2-propenoate (9CI)

OTHER NAMES:

CN Methyl methacrylate-2-(4'-vinyl-4-biphenylyloxy)pentachlorocyclotriph osphazene copolymer

MF (C14 H11 C15 N3 O P3 . C5 H8 O2)x

CI PMS

PCT Polyacrylic, Polyphosphazene, Polyphosphazene formed, Polystyrene

SR CA

LC STN Files: CA, CAPLUS

CM 1

CRN 115529-67-4

CMF C14 H11 C15 N3 O P3

$$\begin{array}{c|c} C1 & C1 \\ \hline \\ O & P & P \\ \hline \\ N & N \\ \end{array}$$

CM 2

CRN 80-62-6 CMF C5 H8 O2

2 REFERENCES IN FILE CA (1907 TO DATE) 2 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 28 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN

RN 99798-90-0 REGISTRY

ED Entered STN: 18 Jan 1986

CN 2-Propenoic acid, 2-methyl-, methyl ester, polymer with 2-(1-ethoxyethenyl)-2,4,4,6,6-pentafluoro-2,2,4,4,6,6-hexahydro-1,3,5,2,4,6-triazatriphosphorine (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES:

CN 1,3,5,2,4,6-Triazatriphosphorine, 2-(1-ethoxyethenyl)-2,4,4,6,6-pentafluoro-2,2,4,4,6,6-hexahydro-, polymer with methyl 2-methyl-2-propenoate (9CI)

OTHER NAMES:

CN $(\alpha-\text{Ethoxyvinyl})$ pentafluorocyclotriphosphazene-methyl methacrylate copolymer

MF (C5 H8 O2 . C4 H7 F5 N3 O P3) \times

CI PMS

PCT Polyacrylic, Polyphosphazene, Polyphosphazene formed, Polyvinyl

SR CA

LC STN Files: CA, CAPLUS

CM 1

CRN 80297-67-2 CMF C4 H7 F5 N3 O P3

CM 2

CRN 80-62-6 CMF C5 H8 O2

- 1 REFERENCES IN FILE CA (1907 TO DATE)
- 1 REFERENCES IN FILE CAPLUS (1907 TO DATE)
- L4 ANSWER 29 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN -
- RN 92832-54-7 REGISTRY
- ED Entered STN: 17 Dec 1984
- CN 2-Propenoic acid, 2-methyl-, 1,2-ethanediylbis(oxy-2,1-ethanediyl) ester, polymer with 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-1,3,5,2,4,6-triazatriphosphorine (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES:

CN 1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]-, polymer with 1,2-ethanediylbis(oxy-2,1-ethanediyl) bis(2-methyl-2-propenoate) (9CI) OTHER NAMES:

CN Hexakis(methacryloyloxyethoxy)cyclotriphosphazene-triethylene glycol dimethacrylate copolymer

MF (C36 H54 N3 O18 P3 . C14 H22 O6) x

CI PMS

PCT Polyacrylic, Polyphosphazene, Polyphosphazene formed

LC STN Files: CA, CAPLUS, USPATFULL

CM 1

CRN 92832-53-6

CMF C36 H54 N3 O18 P3

CM 2

CRN 109-16-0 CMF C14 H22 O6

4 REFERENCES IN FILE CA (1907 TO DATE)

4 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 30 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN

RN 92832-53-6 REGISTRY

ED Entered STN: 17 Dec 1984

CN 1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4,6,6-hexahydro-2,2,4,4,6,6-hexakis[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethoxy]- (9CI) (CA INDEX NAME) OTHER NAMES:

CN Hexakis (methacryloyloxyethoxy) cyclotriphosphazene

DR 123522-97-4

MF C36 H54 N3 O18 P3

CI COM

LC STN Files: CA, CAPLUS, CASREACT, CHEMLIST, MEDLINE, USPATFULL

PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

48 REFERENCES IN FILE CA (1907 TO DATE)

3 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA

48 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L4 ANSWER 31 OF 31 REGISTRY COPYRIGHT 2006 ACS on STN

RN 41407-17-4 REGISTRY

ED Entered STN: 16 Nov 1984

CN 1,3,5,2,4,6-Triazatriphosphorine, 2,2,4,4(or 2,2,4,6)tetrakis(dimethylamino)-2,2,4,4,6,6-hexahydro-6,6(or 4,6)-bis[[2-[(1-oxo-2propenyl)oxy]ethyl]amino]- (9CI) (CA INDEX NAME)

OTHER NAMES:

CN Bis (acryloyloxyethylamino) tetrakis (dimethylamino) cyclotriphosphazene*

MF C18 H40 N9 O4 P3

CI IDS.

LC STN Files: CA, CAPLUS, IFICDB, IFIPAT, IFIUDB

- 1 REFERENCES IN FILE CA (1907 TO DATE)
- 1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

```
=> S BIURET
L5
          1216 BIURET
=> S BIURET?
          1216 BIURET?
1.6
=> S L6 AND (ACRYL ENE ALLYL)
        103858 ACRYL
       6304255 ENE
         58502 ALLYL
             O ACRYL ENE ALLYL
                  (ACRYL (W) ENE (W) ALLYL)
L7
             0 L6 AND (ACRYL ENE ALLYL)
=> D HIS
     (FILE 'HOME' ENTERED AT 08:07:37 ON 04 SEP 2006)
     FILE 'REGISTRY' ENTERED AT 08:07:48 ON 04 SEP 2006
L1
            492 S PHOSPHAZENE
L2
             31 S L1 AND ACRYL?
L3
              6 S L2 AND ENE
L4
             31 S L2 OR L3
L5
           1216 S BIURET
L6
           1216 S BIURET?
L7
              O S L6 AND (ACRYL ENE ALLYL)
=> FILE REG
COST IN U.S. DOLLARS
                                                   SINCE FILE
                                                                    TOTAL
                                                        ENTRY
                                                                  SESSION
FULL ESTIMATED COST
                                                        130.06
                                                                   130.27
DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)
                                                   SINCE FILE
                                                                    TOTAL
                                                        ENTRY
                                                                  SESSION
CA SUBSCRIBER PRICE
                                                          -4.26
                                                                     -4.26
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REGISTRY includes numerically searchable data for experimental and predicted properties as well as tags indicating availability of experimental property data in the original document. For information on property searching in REGISTRY, refer to:

http://www.cas.org/ONLINE/UG/regprops.html

=> S L4 L8 31 L2 OR L3

=> FILE CAPLUS

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http://www.cas.org/infopolicy.html

=> S L8

L9 70 L8

=> D ALL 1-70

- L9 ANSWER 1 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
- AN 2006:2108 CAPLUS
- DN 144:150886
- ED Entered STN: 03 Jan 2006
- TI Multimorphological Self-Assemblies of Amphiphilic Graft Polyphosphazenes with Oligopoly(N-isopropylacrylamide) and Ethyl 4-Aminobenzoate as Side Groups
- AU Zhang, Jian Xiang; Qiu, Li Yan; Jin, Yi; Zhu, Kang Jie
- CS College of Pharmaceutical Sciences, Zhejiang University, Hangzhou, 310031, Peop. Rep. China
- SO Macromolecules (2006), 39(1), 451-455 CODEN: MAMOBX; ISSN: 0024-9297
- PB American Chemical Society
- DT Journal
- LA English
- CC 36-2 (Physical Properties of Synthetic High Polymers) Section cross-reference(s): 35
- AB Both TEM and AFM observations indicated that the self-assembly morphol. of the title graft copolymer were regulated along with the variation of copolymer composition which could be achieved due to the synthetic flexibility and versatility of polyphosphazene. Depending on various solvents used in aggregate preparation procedure, multimorphol. assemblies ranging from network structure and nanospheres to high-genus particles were observed for the copolymer containing a relatively high Et aminobenzoate content. The

temperature would induce the morphol. change of aggregates, which was resulted from the thermosensitivity of polyisopropylacrylamide segments in the copolymers.

```
ST
     multimorphol self assembly graft polyphosphazene
     oligopolyisopropylacrylamide aminobenzoate
ΙT
     Critical solution temperature
        (lower; multimorphol. self-assemblies of amphiphilic graft
        polyphosphazenes with oligopoly(isopropylacrylamide) and Et
        aminobenzoate as side groups)
IT
     Polymer morphology
     Self-assembly
        (multimorphol. self-assemblies of amphiphilic graft polyphosphazenes
        with oligopoly(isopropylacrylamide) and Et aminobenzoate as side
ΙT
     Polyphosphazenes
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (multimorphol. self-assemblies of amphiphilic graft polyphosphazenes
        with oligopoly(isopropylacrylamide) and Et aminobenzoate as side
        groups)
ΙT
     156-57-0DP, 2-Aminoethanethiol hydrochloride, reaction products with
     polyisopropylacrylamide
                               25189-55-3DP, Polyisopropylacrylamide, reaction
     products with aminoethanethiol hydrochloride
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (in preparation of amphiphilic graft polyphosphazenes with
        oligopoly(isopropylacrylamide) and Et aminobenzoate as side groups)
ΙT
     94-09-7DP, Ethyl 4-aminobenzoate, reaction products with
     hexachlorocyclotriphosphazene-isopropylacrylamide graft copolymer
     794574-57-5DP, Hexachlorocyclotriphosphazene-isopropylacrylamide
     graft copolymer, reaction products with Et aminobenzoate
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (multimorphol. self-assemblies of amphiphilic graft polyphosphazenes
        with oligopoly(isopropylacrylamide) and Et aminobenzoate as side
        groups)
RE.CNT
        44
              THERE ARE 44 CITED REFERENCES AVAILABLE FOR THIS RECORD .
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(28) Soo, P; J Polym Sci, Part B: Polym Phys 2004, V42, P923 CAPLUS
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L9
     ANSWER 2 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
ΑN
     2005:1258917 CAPLUS
DN
     144:129539
ED
     Entered STN: 01 Dec 2005
TI
     Solvent controlled multi-morphological self-assembly of amphiphilic graft
ΑU
     Zhang, Jian Xiang; Qiu, Li Yan; Zhu, Kang Jie
CS
     College of Pharmaceutical Sciences, Zhejiang University, Hangzhou, 310027,
     Peop. Rep. China
SO
     Macromolecular Rapid Communications (2005), 26(21), 1716-1723
     CODEN: MRCOE3; ISSN: 1022-1336
PΒ
     Wiley-VCH Verlag GmbH & Co. KGaA
DT
     Journal
LA
     English
CC
     36-7 (Physical Properties of Synthetic High Polymers)
AB
     Amphiphilic graft polyphosphazenes (tryptophan Et ester/poly(N-
     isopropylacrylamide) - hexachloro-cyclo-tri-phosphazene) with different mole
     ratios of hydrophobic groups to hydrophilic segments were synthesized by
     ring-opening polymerization and subsequent substitution reactions. The
     self-assembly behavior of these graft copolymers was studied in detail.
     Depending on the copolymer composition and common organic solvent employed in
     dialysis process, supramol. aggregates ranging from network, nano-spheres,
     high-genus particles to macrophage-like aggregates were produced with
     graft copolymers.
ST
     polyphosphazene graft morphol self assembly solvent amphiphiles
IT
     Solvents
        (organic; solvent controlled multi morphol. self-assembly of amphiphilic
        graft copolymers)
ΙT
     Aggregates
     Amphiphiles
     Polymer morphology
     Self-assembly
        (solvent controlled multi morphol. self-assembly of amphiphilic graft
        copolymers)
     67-56-1, Methanol, uses
                               67-64-1, Acetone, uses
                                                         67-68-5,
ΙT
     Dimethylsulfoxide, uses
                               109-99-9, uses
                                                127-19-5, N, N-Dimethylacetamide
     RL: NUU (Other use, unclassified); USES (Uses)
        (solvent controlled multi morphol. self-assembly of amphiphilic graft
        copolymers)
     6519-66-0D, reaction products with hexachloro-cyclo -tri-phosphazene-
IΤ
     isopropylacrylamide graft copolymer 794574-57-5D, reaction
     products with tryptophan Et ester
     RL: PRP (Properties)
        (solvent controlled multi morphol. self-assembly of amphiphilic graft
        copolymers)
              THERE ARE 42 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE.CNT
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(29) van Hest, J; Science 1995, V268, P1592 CAPLUS
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     ANSWER 3 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
ΑN
     2005:1019366 CAPLUS
DN
     144:489226
ED
     Entered STN: 22 Sep 2005
     Thermosensitive self-assembly behaviors of novel amphiphilic
TI
     polyphosphazenes
ΑU
     Qiu, Liyan; Zhang, Jianxiang; Jin, Yi; Zhu, Kangjie
     College of Pharmaceutical Sciences, Zhejiang University, Hangzhou, 310031,
CS
     Peop. Rep. China
     Chinese Science Bulletin (2005), 50(14), 1453-1455
SO
     CODEN: CSBUEF; ISSN: 1001-6538
`PB
     Science in China Press
DT
     Journal
LA
     English
     36-6 (Physical Properties of Synthetic High Polymers)
CC
AΒ
     Amphiphilic thermosensitive polyphosphazenes (PNIPAm-g-PPP) bearing
     N-isopropylacrylamide oligomers (oligo-PNIPAm) and glycine Et groups
     (GlyEt) as co-substituents were synthesized via polymer substitute
     reaction. UV-visible spectra indicated that the aqueous solution of
PNIPAm-g-PPP
     exhibited the lower critical solution temperature (LCST). Also, the LCST was
     seriously influenced by the substitution ratios of PNIPAm to GlyEt in the
     copolymer. The more GlyEt the copolymer contained, the lower LCST it had.
     The critical association concentration (CAC) of copolymers was determined by
fluorescence
     probe method. Cac was decreased with increasing GlyEt content of
```

polyphosphazene. Also the formation of self-assembled micelles or

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nano-particles was confirmed by TEM.
ST
    self assembly amphiphilic polyphosphazene; polyisopropylacrylamide
    polyphosphazene graft copolymer thermosensitive micelle
ΙT
    Critical solution temperature
       (lower; self-assembly behavior of amphiphilic grafted polyphosphazenes)
IT
    Micelles
    Nanoparticles
    Polymer morphology
    Self-assembly
       (self-assembly behavior of amphiphilic grafted polyphosphazenes)
ΙT
    Polyphosphazenes
    RL: PEP (Physical, engineering or chemical process); PNU (Preparation,
    unclassified); PRP (Properties); PYP (Physical process); PREP
     (Preparation); PROC (Process)
       (self-assembly behavior of amphiphilic grafted polyphosphazenes)
    459-73-4DP, reaction products with hexachlorocyclotriphosphazene-
ΙT
    isopropylacrylamide graft copolymer 794574-57-5DP, reaction
    products with glycine Et ester
    RL: PEP (Physical, engineering or chemical process); PNU (Preparation,
    unclassified); PRP (Properties); PYP (Physical process); PREP
     (Preparation); PROC (Process)
       (self-assembly behavior of amphiphilic grafted polyphosphazenes)
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RE.CNT
RE
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    ANSWER 4 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
    2005:75417 CAPLUS
ΑN
DN
    142:144371
ED
    Entered STN: 28 Jan 2005
TI
    Gas-barrier plastic laminates with good adhesion and durability, and
    displays having them
ΙN
    Watanabe, Takuzo
PA
    Toppan Printing Co., Ltd., Japan
SO
    Jpn. Kokai Tokkyo Koho, 10 pp.
    CODEN: JKXXAF
DT
    Patent
    Japanese
LA
IC
    ICM B32B009-04
     ICS B32B027-00; H05B033-04; H05B033-14
    74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other
CC
     Reprographic Processes)
     Section cross-reference(s): 38
FAN.CNT 1
                              DATE
                                        APPLICATION NO.
                                                               DATE
     PATENT NO.
                      KIND
                             -----
                                         _____
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                              20050127 JP 2003-270610
    JP 2005022375
                       A2
                                                                20030703
                              20030703
PRAI JP 2003-270610
CLASS
 PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
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                       B32B009-04
 JP 2005022375 ICM
                       B32B027-00; H05B033-04; H05B033-14
               ICS
                       B32B0009-04 [ICM,7]; B32B0027-00 [ICS,7]; H05B0033-04
               IPCI
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[ICS,7]; H05B0033-14 [ICS,7]

IPCR B32B0009-04 [I,A]; B32B0009-04 [I,C*]; B32B0027-00 [I,A]; B32B0027-00 [I,C*]; H05B0033-04 [I,A]; H05B0033-04 [I,C*]; H05B0033-14 [I,A]; H05B0033-14 [I,C*]

FTERM 3K007/AB12; 3K007/AB13; 3K007/AB15; 3K007/BB01; 3K007/CA06; 3K007/DB03; 3K007/FA02; 4F100/AA01B; 4F100/AA20; 4F100/AH07C; 4F100/AK42; 4F100/AT00A; 4F100/BA03; 4F100/BA04; 4F100/BA05; 4F100/BA10A; 4F100/BA10C; 4F100/EJ66; 4F100/EJ24; 4F100/EJ24C; 4F100/EJ53; 4F100/JB14C; 4F100/JD02; 4F100/JJ02; 4F100/JK06; 4F100/JN01

OS MARPAT 142:144371 GI

AB The laminates, useful for electroluminescence (EL) displays, liquid crystal displays (LCD), etc., contain substrate layers, inorg. layers, and layers containing cyclophosphazene derivs. manufactured from I (R1 = radiation-reactive

group). The radiation-reactive groups may be H2C:R2CO2(CH2)nO (R2 = H, Me; n = natural number).

ST gas barrier laminate acrylic polycyclophosphazene display

IT Polyphosphazenes

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(acrylic, cyclomatrix; gas-barrier plastic laminates with good adhesion and durability for displays)

IT Polyphosphazenes

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(cyclomatrix; gas-barrier plastic laminates with good adhesion and durability for displays)

IT Electroluminescent devices

(displays; gas-barrier plastic laminates with good adhesion and durability for displays)

IT Luminescent screens

(electroluminescent; gas-barrier plastic laminates with good adhesion and durability for displays)

IT Laminated plastic films

(gas-barrier plastic laminates with good adhesion and durability for displays)

IT Polyesters, uses

RL: TEM (Technical or engineered material use); USES (Uses) (substrate; gas-barrier plastic laminates with good adhesion and durability for displays)

1T 93891-06-6P, PPZ 827320-64-9P, Hexakis(methacryloyloxyethoxy)cyc
lotriphosphazene-2-hydroxyethyl acrylate copolymer 827320-65-0P,
Aronix M 220-hexakis(methacryloyloxyethoxy)cyclotriphosphazene copolymer
RL: IMF (Industrial manufacture); TEM (Technical or engineered material
use); PREP (Preparation); USES (Uses)

```
(cyclomatrix; gas-barrier plastic laminates with good adhesion and
        durability for displays)
                                7631-86-9, Silica, uses
ΙT
     1344-28-1, Alumina, uses
                                                          50926-11-9, ITO
     RL: TEM (Technical or engineered material use); USES (Uses)
        (gas-barrier plastic laminates with good adhesion and durability for
        displays)
TΤ
     25230-87-9
     RL: TEM (Technical or engineered material use); USES (Uses)
        (substrate, assumed monomers; gas-barrier plastic laminates with good
        adhesion and durability for displays)
ΙT
     24968-11-4, PEN
                      25038-59-9, PET polymer, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
        (substrate; gas-barrier plastic laminates with good adhesion and
        durability for displays)
     ANSWER 5 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
ΑN
     2004:793032 CAPLUS
     141:424682
DN
ED
     Entered STN: 29 Sep 2004
     Thermosensitive micelles self-assembled by novel N-isopropylacrylamide
ΤI
     oligomer grafted polyphosphazene
ΑU
     Zhang, Jian Xiang; Qiu, Li Yan; Zhu, Kang Jie; Jin, Yi
     College of Pharmaceutical Sciences, Zhejiang University, Hangzhou, 310031,
CS
     Peop. Rep. China
     Macromolecular Rapid Communications (2004), 25(17), 1563-1567
SO
     CODEN: MRCOE3; ISSN: 1022-1336
PΒ
     Wiley-VCH Verlag GmbH & Co. KGaA
     Journal
DT
LA
     English
CC
     36-7 (Physical Properties of Synthetic High Polymers)
     Section cross-reference(s): 35
     A novel thermosensitive amphiphilic copolymer of poly(N-
AB
     isopropylacrylamide) grafted polyphosphazene (PNIPAm-g-PPP) has been
     synthesized through a co-substitution reaction of
     poly(dichlorophosphazene) with amino-terminated NIPAm oligomer and glycine
     Et ester (GlyEt). The polymer obtained had a molar ratio of 1:5.25 PNIPAm
     to GlyEt and had a lower critical solution temperature (LCST) near 30°. The
     formation of polymeric micelles was confirmed by fluorescence, dynamic
     light scattering and transmission electron microscopy measurements.
ST
     polyisopropylacrylamide polyphosphazene graft copolymer thermosensitive
     micelle
IT
     Fluorescence
     Hydrodynamic radius
     Micelles
     Polymer morphology
     Polymerization
        (preparation and properties of thermosensitive micelles self-assembled by
        isopropylacrylamide oligomer grafted polyphosphazene)
ΙT
     Polyphosphazenes
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (preparation and properties of thermosensitive micelles self-assembled by
        isopropylacrylamide oligomer grafted polyphosphazene)
ΙT
     156-57-0DP, 2-Aminoethanethiol hydrochloride, reaction products with
     polyisopropylacrylamide
                               25189-55-3DP, Poly(N-isopropylacrylamide),
     reaction products with aminoethanethiol hydrochloride
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (intermediate; preparation and properties of thermosensitive micelles
        self-assembled by isopropylacrylamide oligomer grafted polyphosphazene)
ΙT
     459-73-4DP, reaction products with hexachlorocyclotriphosphazene-
     isopropylacrylamide graft copolymer 794574-57-5DP, reaction.
     products with glycine Et ester
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (preparation and properties of thermosensitive micelles self-assembled by
        isopropylacrylamide oligomer grafted polyphosphazene)
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RE.CNT 24
             THERE ARE 24 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE
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(23) Topp, M; Macromolecules 2000, V33, P4986 CAPLUS
(24) Wilhelm, M; Macromolecules 1991, V24, P1033 CAPLUS
    ANSWER 6 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
    2004:631816 CAPLUS
AN
DN
    141:141213
ED
    Entered STN: 06 Aug 2004
    Three-dimensionally crosslinked resins and transparent molded products
    Kawasaki, Noboru; Kogo, Osamu; Enya, Masahiro
IN
    Mitsui Chemicals Inc., Japan
    Jpn. Kokai Tokkyo Koho, 20 pp.
    CODEN: JKXXAF
DT
    Patent
LΑ
    Japanese
     ICM C08F220-14
IC
     ICS C08F220-18; C08F220-28; C08F220-36; C08F230-02; C08F290-06
     37-3 (Plastics Manufacture and Processing)
FAN.CNT 1
                                      APPLICATION NO. DATE
                       KIND
     PATENT NO.
                              DATE
     _____
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                               -----
                                          -----
    JP 2004217700
                        A2
                               20040805 · JP 2003-3682
                                                                  20030109
PRAI JP 2003-3682
                               20030109
CLASS
               CLASS PATENT FAMILY CLASSIFICATION CODES
PATENT NO.
                _____
                       C08F220-14
JP 2004217700 ICM
                ICS
                       C08F220-18; C08F220-28; C08F220-36; C08F230-02;
                       C08F290-06
                IPCI
                       C08F0220-14 [ICM, 7]; C08F0220-18 [ICS, 7]; C08F0220-28
                       ·[ICS,7]; C08F0220-36 [ICS,7]; C08F0220-00 [ICS,7,C*];
                       C08F0230-02 [ICS,7]; C08F0230-00 [ICS,7,C*];
                       C08F0290-06 [ICS,7]; C08F0290-00 [ICS,7,C*]
                 IPCR
                       C08F0220-00 [I,C*]; C08F0220-14 [I,A]; C08F0220-18
                       [I,A]; C08F0220-28 [I,A]; C08F0220-36 [I,A];
                       C08F0230-00 [I,C*]; C08F0230-02 [I,A]; C08F0290-00
                       [I,C*]; C08F0290-06 [I,A]
                 FTERM
                       4J027/AG03; 4J027/AG04; 4J027/AG14; 4J027/AG23;
                        4J027/AG24; 4J027/BA07; 4J027/CB03; 4J027/CC02;
                        4J027/CD01; 4J100/AL03P; 4J100/AL08T; 4J100/AL66Q;
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4J100/AL66T; 4J100/AL67R; 4J100/AL74S; 4J100/BA02R;

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4J100/BC43Q; 4J100/BC43S; 4J100/BC51R; 4J100/BC75R;
                        4J100/CA03; 4J100/CA04; 4J100/CA05; 4J100/CA06;
                        4J100/DA25; 4J100/DA28; 4J100/DA37; 4J100/DA49;
                        4J100/JA58
AB
     Compns. comprising Me methacrylate, R1 (OCONHQNHCO2-Pol) nOCONHQNHCO2R2 [R1,
     R2 = hydroxy (meth)acrylate residue; Q = residue of aliphatic, aromatic, or
     alicyclic diisocyanate; Pol = polyether diol or polyester diol residue; n
     = 1-5], and radical initiators are polymerized to give the resins with no
     runaway reaction. Thus, a mixture of 100 parts Me methacrylate and 40 parts
     of polyether-polyurethane acrylate (3:4:2 triethylene glycol-IPDI-2-
     hydroxyethyl acrylate condensate) was mixed with 0.2 part tert-Bu
     peroxy-2-ethylhexanoate and 0.1 part tert-Bu peroxy-3,5,5-
     trimethylhexanoate, poured in a space between glass sheets, and heated to
     70-130° to give a transparent resin sheet showing Tg 115°,
     flexural modulus 3.2 GPa, H2O absorption 0.45%, no cracks at mold
     releasing, and good chemical resistance.
ST
     polyether polyurethane acrylate methyl methacrylate polymer prepn
ΙT
     Polyurethanes, preparation
     RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or
     engineered material use); PREP (Preparation); USES (Uses)
        (acrylic-polyether-; three-dimensionally crosslinked Me
        methacrylate-polyurethane (meth)acrylate resins for transparent molded
        products)
ΙT
     Polyesters, preparation
     Polyethers, preparation
     RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or
     engineered material use); PREP (Preparation); USES (Uses)
        (acrylic-polyurethane-; three-dimensionally crosslinked Me
        methacrylate-polyurethane (meth)acrylate resins for transparent molded
        products)
ΙT
     Transparent materials
        (three-dimensionally crosslinked Me methacrylate-polyurethane
        (meth)acrylate resins for transparent molded products)
ΙT
     2094-99-7, 3-Isopropenyl-\alpha, \alpha-dimethylbenzyl isocyanate
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction with hydroxy-containing methacrylate)
     727704-32-7P, 2-Hydroxyethyl acrylate-isophorone diisocyanate-methyl
ΙT
     methacrylate-triethylene glycol copolymer 727704-34-9P,
     1,1,3,3,5,5-Hexakis(acryloylethylenedioxy)cyclotriphosphazene-2-
     hydroxyethyl methacrylate-methyl methacrylate-2,4-TDI-tripropylene glycol
                 727704-36-1P, Ethylene oxide-glycidol methacrylate-N-(3-
     isopropenyl-a, a-dimethylbenzyl)-1-methacryloyloxypropan-2-yl
     carbamate-methyl methacrylate-propylene oxide-2,4-TDI copolymer
     727704-38-3P, Dimethyloltricyclodecane diacrylate-ethoxylated bisphenol
     A-hexamethylene diisocyanate-2-hydroxyethyl methacrylate-methyl
     methacrylate copolymer
                              727704-40-7P, Dibutylene glycol-dicyclopentanyl
     acrylate-2-hydroxyethyl acrylate-isophorone diisocyanate-N-(3-isopropenyl-
     \alpha, \alpha-dimethylbenzyl)-2-methacryloyloxyethyl carbamate-methyl
     methacrylate-tris[N-(2-acryloyloxyethyl)] isocyanurate copolymer
     727704-42-9P, Adipic acid-ethylene glycol-1,6-hexanediol-isophorone
     diisocyanate-methyl methacrylate-pentaerythritol triacrylate copolymer
     727704-44-1P, Ethylene glycol-1,1,3,3,5,5-
     hexakis(acryloylethylenedioxy)cyclotriphosphazene-1,6-hexanediol-2-
     hydroxypropyl acrylate-methylenebis(cyclohexyl isocyanate)-methyl
     methacrylate-terephthalic acid copolymer
                                                727704-47-4P, Adipic
     acid-1,4-butanediol-ethylene glycol-2-hydroxypropyl acrylate-N-(3-
     isopropenyl-α, α-dimethylbenzyl)-1-methacryloyloxypropan-2-yl
     carbamate-methyl methacrylate-trimethylhexamethylene diisocyanate
                 727704-49-6P, Cyclohexyl methacrylate-ethylene
     glycol-1,6-hexanediol-2-hydroxyethyl methacrylate-isophthalic
     acid-methylenebis(cyclohexyl isocyanate)-methyl methacrylate copolymer
     727704-51-0P, Adipic acid-ethylene glycol-1,6-hexanediol-2-hydroxyethyl
```

acrylate-isobornyl methacrylate-isophorone diisocyanate-N-(3-isopropenyl-

4J100/BA02T; 4J100/BA04R; 4J100/BA34Q; 4J100/BA38S; 4J100/BC04Q; 4J100/BC04T; 4J100/BC07T; 4J100/BC26T;

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\alpha, \alpha-dimethylbenzyl)-2-methacryloyloxyethyl carbamate-methyl
     methacrylate-tris[N-(2-acryloyloxyethyl)] isocyanurate copolymer
     RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or
     engineered material use); PREP (Preparation); USES (Uses)
        (three-dimensionally crosslinked Me methacrylate-polyurethane
        (meth)acrylate resins for transparent molded products)
ΙT
     126710-00-7P
                    126710-08-5P
     RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT
     (Reactant or reagent)
        (three-dimensionally crosslinked Me methacrylate-polyurethane
        (meth)acrylate resins for transparent molded products)
L9
     ANSWER 7 OF 70 GAPLUS COPYRIGHT 2006 ACS on STN
     2004:473452 CAPLUS
ΑN
     141:44886
DN
ED
     Entered STN: 11 Jun 2004
TΤ
     Photopolymer composition suitable for lithographic printing plates
     Baumann, Harald; Flugel, Michael; Dwars, Udo; Glatt, Hans Horst
ΙN
PA
     Kodak Polychrome Graphics GmbH, Germany
     PCT Int. Appl., 32 pp.
SO
     CODEN: PIXXD2
DT
     Patent
LA
     English
IC
     ICM G03F007-027
     74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other
CC
     Reprographic Processes)
FAN.CNT 1
                         KIND
                                           APPLICATION NO.
                                                                  DATE
     PATENT NO.
                               DATE
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                                           ______
                                                                  _____
                               20040610
                                                                  20031128
PΙ
     WO 2004049069
                         A1
                                           WO 2003-EP13432
         W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,
             CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH,
             GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR,
             LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NI, NO, NZ, OM,
             PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN,
             TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW
         RW: BW, GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ,
             BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE,
             ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE, SI, SK,
             TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG
                                           DE 2002-10255664
     DE 10255664
                         Α1
                                20040708
                                                                   20021128
     DE 10255664
                          В4
                                20060504
                         Α1
                                20040618
                                           AU 2003-292156
                                                                  20031128
     AU 2003292156
                                           EP 2003-767702
                                20050824
                                                                  20031128
     EP 1565788
                         Α1
             AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK
                                           US 2005-536510
                         Α1
                                20060413
                                                                  20050525
     US 2006078819
                          Α
                                20021128
PRAI DE 2002-10255664
     WO 2003-EP13432
                                20031128
                          W
CLASS
 PATENT NO.
                 CLASS
                        PATENT FAMILY CLASSIFICATION CODES
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                        G03F007-027
 WO 2004049069
                 ICM
                        G03F0007-027 [ICM, 7]
                 IPCI
                        G03F0007-027 [I,A]; G03F0007-027 [I,C*]
                 IPCR
                        C08G018/67B2; C08G018/78B4F; G03F007/027
                 ECLA
 DE 10255664
                 IPCI
                        G03F0007-027 [I,A]; G03F0007-028 [I,A]
                        G03F0007-027 [I,A]; G03F0007-027 [I,C*]
                 IPCR
                        C08G018/67B2; C08G018/78B4F; G03F007/027
                 ECLA
                        G03F0007-027 [ICM, 7]
 AU 2003292156
                 IPCI
                        G03F0007-027 [I,A]; G03F0007-027 [I,C*]
                 IPCR
 EP 1565788
                 IPCI
                        G03F0007-027 [ICM, 7]
                        G03F0007-027 [I,A]; G03F0007-027 [I,C*]
                 IPCR
                        C08G018/67B2; C08G018/78B4F; G03F007/027
                 ECLA
 US 2006078819
                 IPCI
                        G03C0001-76 [I,A]
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NCL 430/270.100
ECLA C08G018/67B2; C08G018/78B4F; G03F007/027
ensitive elements are described whose radiation-s

- AB Radiation-sensitive elements are described whose radiation-sensitive coating comprises both a photopolymerizable oligomer with a biuret structural unit and a photopolymerizable phosphazene oligomer. The object of the present invention is to provide novel radiation-sensitive elements exhibiting improved properties compared to the ones known in the prior art, in particular high photosensitivity, high degree of thermal stability, high resolution in combination with good storage stability, and in the case of printing plates yielding a large number of copies on the printing machine.
- ST photopolymer compn lithog printing plate
- IT Lithographic plates

(photopolymer composition suitable for lithog. printing plates)

IT Polyvinyl acetals

RL: TEM (Technical or engineered material use); USES (Uses) (photopolymer composition suitable for lithog. printing plates)

IT 32435-46-4, Kayamer PM-2

RL: TEM (Technical or engineered material use); USES (Uses) (Kayamer PM-2; photopolymer composition suitable for lithog. printing plates)

IT 147-14-8, Copper phthalocyanine 7429-90-5, Aluminum, uses 25035-81-8, Methacrylic acid-methylmethacrylate-styrene copolymer 92832-53-6 115965-96-3, Airvol 203 185396-46-7, Desmodur N 100/2-hydroxyethyl acrylate/pentaerythritol triacrylate copolymer 701921-54-2, Desmodur N 100-pentaerythritol triacrylate copolymer

RL: TEM (Technical or engineered material use); USES (Uses)

(photopolymer composition suitable for lithog. printing plates)

RE.CNT 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD RE

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- (2) Faust, R; US 4019972 A 1977 CAPLUS
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- (4) Nichigo Morton Co Ltd; EP 0961171 A 1999 CAPLUS
- L9 ANSWER 8 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
- AN 2003:358950 CAPLUS
- DN 139:135193
- ED Entered STN: 12 May 2003
- TI Synthesis of hexa(2-hydroxyethyl methacrylate) cyclotriphosphazene
- AU Nie, Xu-Wen; Cui, Yan-Jun; Tang, Xiao-Zhen
- CS School of Chemistry and Chemical Technology, Shanghai Jiaotong University, Shanghai, 200240, Peop. Rep. China
- SO Yingyong Huaxue (2003), 20(4), 385-387 CODEN: YIHUED; ISSN: 1000-0518
- PB Yingyong Huaxue Bianji Weiyuanhui
- DT Journal
- LA Chinese
- CC 45-4 (Industrial Organic Chemicals, Leather, Fats, and Waxes)
- OS CASREACT 139:135193
- AB Hexa(2-hydroxyethyl methacrylate)cyclotriphosphazene(HHMP) was synthesized from 2-hydroxyethyl methacrylate and hexachlorocyclotriphosphazene through nucleophilic substitution-reaction in the presence of cuprous chloride as a catalyst. Raman, FTIR and NMR techniques were used to characterize and confirm the structure of the product.
- ST hexachlorocyclotriphosphazene hydroxyethyl methacrylate synthesis hexakismethacryloyloxyethoxycyclotriphosphazene
- IT 868-77-9, 2-Hydroxyethyl methacrylate
 - RL: RCT (Reactant); RACT (Reactant or reagent)
 - (in reaction with hexachlorocyclotriphosphazene)
- IT 940-71-6P, Hexachlorocyclotriphosphazene
 - RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)
 - (intermediate; preparation of, and reaction with hydroxyethyl methacrylate)
- IT 10026-13-8, Phosphorus pentachloride

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RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction with ammonium chloride in preparation of
        hexachlorocyclotriphosphazene)
TΤ
     12125-02-9, Ammonium chloride, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction with phosphorus pentachloride in preparation of
        hexachlorocyclotriphosphazene)
ΙT
     92832-53-6P, Hexakis (methacryloyloxyethoxy) cyclotriphosphazene
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (synthesis, of)
L9
    ANSWER 9 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
ΑN
     2003:289667 CAPLUS
DN
     139:85871
ED
     Entered STN: 15 Apr 2003
     Fluorescence behavior of water-soluble copolymers with pendant
TΤ
     (4-carboxylatophenoxy)cyclotriphosphazene/europium ion complexes
     Itaya, Tomoyuki; Honda, Tetsuji; Kusumoto, Noriyuki; Matsumoto, Akira;
ΑU
     Inoue, Kenzo
     Chemical Laboratory, Nagano National College of Technology, Nagano,
CS
     381-8550, Japan
SO
     Polymer (2003), 44(10), 2927-2932
     CODEN: POLMAG; ISSN: 0032-3861
PB
     Elsevier Science Ltd.
     Journal
DT
     English
LA
     36-5 (Physical Properties of Synthetic High Polymers)
CC
AB
    The water-soluble copolymers composed of styrene derivative with
     (4-carboxylatophenoxy)cyclotriphosphazene (CCP) and N-isopropylacrylamide
     (NIPAM) have been prepared as host polymers of Eu3+ ion. When CCP units
     were excited at 260 nm, the characteristic fluorescence of Eu3+ was
     remarkably enhanced compared to that of the excitation of Eu3+ at 396 nm,
     suggesting occurrence of efficient energy transfer from CCP units to Eu3+
          The content of CCP units in the polymer chains significantly
     affected the fluorescence behaviors; for the copolymer with 1.7 mol% CCP
     unit (6a) the intensity increased with increasing of Eu3+ ion and reached
     a limiting value at [Eu3+]/[CCP unit] = 1.0, whereas the fluorescence
    intensity of the copolymer with 12.4 mol% CCP unit (6d) reached a maximum at
     [Eu3+]/[CCP\ unit] = 0.25, decreased by the further addition of Eu3+, and then
     has a constant value. In the 6d-Eu3+ complexes, the number of water mols.
     coordinated with Eu3+ ions, which act as a quencher, varies from 2.8 at
     [Eu3+]/[CCP unit] = 0.25 to 4.1 at [Eu3+]/[CCP unit] = 1.5. The reduced
     viscosity of 6d decreased with increasing the concentration of Eu3+ ions,
     indicating the contraction of polymer chains. The environment of Eu3+
     bound to the copolymer is not affected by the coil-globule transition of
     the copolymer chain, although the 6a-Eu3+ system showed a lower critical
     solution temperature at 38°. The fluorescence behaviors of Eu3+ trapped by
     new host polymers based on cyclotriphosphazene are described.
     carboxylatophenoxycyclotriphosphazene styrene deriv isopropylacrylamide
ST
     copolymer europium complex fluorescence
TΤ
     Complexation
     Fluorescence
     Optical absorption
     Viscosity
        (preparation and fluorescence behavior of water-soluble copolymers with
pendant
        (4-carboxylatophenoxy)cyclotriphosphazene/europium ion complexes)
     552890-89-8P, 2,4,4,6,6-Pentakis(4-Ethoxycarbonylphenoxy)-2-(4-
ΙT
     vinylphenoxy) cyclotriphosphazene
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (monomer; preparation and fluorescence behavior of water-soluble copolymers
        with pendant (4-carboxylatophenoxy)cyclotriphosphazene/europium ion
        complexes)
ΙT
     552890-89-8D, 2,4,4,6,6-Pentakis(4-Ethoxycarbonylphenoxy)-2-(4-
```

```
vinylphenoxy)cyclotriphosphazene, hydrolyzed, complexes with europium
     RL: PRP (Properties)
        (preparation and fluorescence behavior of water-soluble copolymers with
pendant
        (4-carboxylatophenoxy)cyclotriphosphazene/europium ion complexes)
     7440-53-1DP, Europium, complexes with 2,4,4,6,6-pentakis(4-
ΙT
     ethoxycarbonylphenoxy) -2-(4-vinylphenoxy) cyclotriphosphazene-N-
     isopropylacrylamide copolymer 552890-90-1DP,
     2,4,4,6,6-Pentakis(4-Ethoxycarbonylphenoxy)-2-(4-
     vinylphenoxy)cyclotriphosphazene-N-isopropylacrylamide copolymer,
     hydrolyzed, potassium salts 552890-90-1DP, 2,4,4,6,6-Pentakis(4-
     Ethoxycarbonylphenoxy)-2-(4-vinylphenoxy)cyclotriphosphazene-N-
     isopropylacrylamide copolymer, hydrolyzed, potassium salts, complexes with
     europium
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (preparation and fluorescence behavior of water-soluble copolymers with
pendant
        (4-carboxylatophenoxy)cyclotriphosphazene/europium ion complexes)
IT
     120-47-8, Ethyl 4-hydroxybenzoate
                                        136144-85-9, 2-(4-
     Vinylphenoxy) pentachlorocyclotriphosphazene
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (starting material; preparation and fluorescence behavior of water-soluble
        copolymers with pendant (4-carboxylatophenoxy)cyclotriphosphazene/europ
        ium ion complexes)
RE.CNT
              THERE ARE 24 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE
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(14) Mwaura, J; J Am Chem Soc 2000, V123, P2647
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(22) Wang, L; Chem Mater 2000, V12, P2212 CAPLUS
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L9
     ANSWER 10 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
     2002:483938 CAPLUS
AN
DN
     137:202669
ED
     Entered STN: 27 Jun 2002
     Curing and combustion properties of a PU-coating system with UV-reactive
ΤI
     phosphazene
ΑU
     Huang, Woei-Kae; Chen, Kwei-Ju; Yeh, Jen-Taut; Chen, Kan-Nan
CS
     Department of Chemistry, Tamkang University, Tamsui, 251, Taiwan
     Journal of Applied Polymer Science (2002), 85(9), 1980-1991
SO
     CODEN: JAPNAB; ISSN: 0021-8995
PΒ
     John Wiley & Sons, Inc.
DT
     Journal
LA
     English
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42-3 (Coatings, Inks, and Related Products)

CC

- AB A UV-curable polyurethane (PU)-coating system containing phosphorus is formulated by the combination of photoinitiator, PU acrylate oligomer, and UV-reactive phosphazene monomer. PU acrylate oligomer is prepared by the addition of 2-hydroxyethyl methacrylate (HEMA) to NCO-terminated PU prepolymer. UV-reactive phosphazene monomer is derived from the HEMA substitution reaction to hexachlorocyclotriphosphazene (NPCl2)3. The curing reaction of this PU-coating system is carried out by UV irradiation The resultant UV-cured PU-coated films demonstrated better performance properties than those of original UV-cured PU acrylate (UV-PU) without UV-reactive phosphazene monomer. Furthermore, their thermal properties are investigated by a thermogravimetric analyzer and a dynamic mech. thermal analyzer, resp. The combustion behaviors of these UV-cured PU-coated films are evaluated by the measurements of a limiting oxygen index and a cone calorimeter.

 ST polyurethane acrylate hexachlorocyclotriphosphazene curing UV irradn:
- ST polyurethane acrylate hexachlorocyclotriphosphazene curing UV irradn; combustion polyurethane acrylate hexachlorocyclotriphosphazene UV curable coating
- IT Coating materials

(UV-curable; curing and combustion properties of polyurethane coating system with UV-reactive phosphazene)

IT Bending strength

Combustion

Elongation, mechanical

Flammability

Mechanical loss

Storage modulus

Stress-strain relationship

Tensile strength

(curing and properties of polyurethane coating system with UV-reactive phosphazene)

IT Crosslinking

(photochem.; curing and combustion properties of polyurethane coating system with UV-reactive phosphazene)

IT Polyurethanes, uses

RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PRP (Properties); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); PROC (Process); USES (Uses) (polyoxyalkylene-, acrylate-terminated, crosslinked; curing and combustion properties of polyurethane coating system with UV-reactive phosphazene)

IT 868-77-9DP, 2-Hydroxyethyl methacrylate, reaction products with IPDI-polypropylene glycol copolymer, polymers with phosphazene derivs. 39323-37-0DP, IPDI-polypropylene glycol copolymer, reaction products with hydroxyethyl methacrylate, polymers with phosphazene derivs. 92832-53-6DP, polymers with polyurethane acrylate RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PRP (Properties); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); PROC (Process); USES (Uses) (curing and combustion properties of polyurethane coating system with UV-reactive phosphazene)

IT 92832-53-6P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(monomer; curing and combustion properties of polyurethane coating system with UV-reactive phosphazene)

IT 868-77-9, 2-Hydroxyethyl methacrylate 940-71-6,

Hexachlorocyclotriphosphazene

RL: RCT (Reactant); RACT (Reactant or reagent)

(starting material; curing and combustion properties of polyurethane coating system with UV-reactive phosphazene)

- RE.CNT 16 THERE ARE 16 CITED REFERENCES AVAILABLE FOR THIS RECORD
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    ANSWER 11 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
    2002:313388 CAPLUS
AN
DN
    136:334101
    Entered STN: 26 Apr 2002
ED
TΙ
    Magnetic recording media
    Hashimoto, Hiroshi; Murayama, Yuichiro
IN
PΑ
    Fuji Photo Film Co., Ltd., Japan
SO
    Jpn. Kokai Tokkyo Koho, 13 pp.
    CODEN: JKXXAF
DT
    Patent
LA
    Japanese
    ICM G11B005-702
    ICS C09D004-00; C09D005-23; C09D201-00
CC
    77-8 (Magnetic Phenomena)
FAN.CNT 1
                      KIND DATE
                                        APPLICATION NO.
    PATENT NO.
                                                                DATE
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                                         ______
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                                                                _____
                              -----
                              20020426 JP 2000-311455
                       A2
                                                                20001012
    JP 2002123922
                              20001012
PRAI JP 2000-311455
CLASS
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
               ____
JP 2002123922 ICM
                       G11B005-702
                       C09D004-00; C09D005-23; C09D201-00
                ICS
                       G11B0005-702 [ICM,7]; C09D0004-00 [ICS,7]; C09D0005-23
                IPCI
                       [ICS, 7]; C09D0201-00 [ICS, 7]
                       C09D0004-00 [I,A]; C09D0004-00 [I,C*]; C09D0005-23
                IPCR
                       [I,A]; C09D0005-23 [I,C*]; C09D0201-00 [I,A];
                       C09D0201-00 [I,C*]; G11B0005-702 [I,A]; G11B0005-702
                       [I,C*]
    The binders contained in the magnetic layers of the recording media are
AB
    compds. which have mol. weight of 200-1000, N-containing complex rings, and
    ≥3 radiation-hardening functional groups. The binders are hardened
    by radiation application, and the compds. are used 5-30 for ferromagnetic
    powders 100 weight parts. The magnetic recording media have high durability,
    and excellent electromagnetic conversion characteristic.
ST
    magnetic recording media binder radiation hardening
    Binders
    Hardening (mechanical)
    Magnetic recording materials
    Radiation
        (radiation-hardening compds. for magnetic recording media)
     40220-08-4 73046-84-1 92832-53-6 414909-22-1 414909-23-2
ΤТ
    RL: DEV (Device component use); USES (Uses)
        (radiation-hardening compds. for magnetic recording media)
    ANSWER 12 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
    2000:837063 CAPLUS
ΑN
DN
    134:23457
    Entered STN: 30 Nov 2000
ED
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Electrophotographic photoreceptor in process cartridge for

ΤI

```
electrophotographic image-forming apparatus
    Uematsu, Hironori; Maruyama, Akio; Sekiya, Michiyo; Amanomiya, Shoji
IN
PΑ
    Canon Inc., Japan
SO
    Jpn. Kokai Tokkyo Koho, 14 pp.
    CODEN: JKXXAF
DT
    Patent
LA
    Japanese
    ICM G03G005-147
IC
    ICS G03G005-147
    74-3 (Radiation Chemistry, Photochemistry, and Photographic and Other
CC
    Reprographic Processes)
FAN.CNT 1
                     KIND DATE APPLICATION NO.
    PATENT NO.
                                                             DATE
                                       -----
                                                             -----
    _____
                     ----
  JP 2000330314
                      A2 20001130 JP 1999-138664 19990519
PΤ
PRAI JP 1999-138664
                            19990519
CLASS
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
______
JP 2000330314 ICM G03G005-147
               ICS G03G005-147
               IPCI G03G0005-147 [ICM, 7]; G03G0005-147 [ICS, 7]
               IPCR G03G0005-147 [I,A]; G03G0005-147 [I,C*]
AB
    The title photoreceptor has a light-conducting layer and a protecting
    layer on a conductive support, wherein the protecting layer contains a
    binder resin and dispersed fine particles made of a surface-treated metal
    oxide. The protecting layer is formed by treating the surface with a
    silane having 2-3 functional groups and by treating the surface with a
    chlorosilane having a functional group. The photoreceptor has the
    improved surface characteristics such as durability, smoothness.
ST
    electrophotog photoreceptor protecting layer
ΙT
    Electrophotographic apparatus
    Electrophotographic photoconductors (photoreceptors)
       (electrophotog. photoreceptor in process cartridge for electrophotog.
       image-forming apparatus)
    40220-08-4 92832-53-6
IT
    RL: TEM (Technical or engineered material use); USES (Uses)
       (light-curable acrylic monomer in protecting layer)
IT : 1067-33-0, T 1
    RL: TEM (Technical or engineered material use); USES (Uses)
       (metal oxide fine particles in protecting layer)
ΙT
    75-77-4, Chlorotrimethylsilane, uses 75-79-6, Methyltrichlorosilane
    768-33-2, Chloro(dimethyl)phenylsilane 999-97-3 1185-55-3,
    Methyltrimethoxysilane 1825-61-2, Methoxytrimethylsilane 1825-62-3,
    Ethoxytrimethylsilane 18297-63-7 85712-15-8,
    Dimethoxy (methyl) octylsilane
    RL: TEM (Technical or engineered material use); USES (Uses)
       (silane compound in protecting layer)
L9
    ANSWER 13 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
AN
    2000:814742 CAPLUS
    133:367840
DN
    Entered STN: 21 Nov 2000
ED
TI
    Photosensitive composition
IN
    Setiabudi, Frans
    Ciba Specialty Chemicals Holding Inc., Switz.
PA
SO
    PCT Int. Appl., 23 pp.
    CODEN: PIXXD2
DT
    Patent
LA
    English
IC
    ICM G03F007-027
    74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other
CC
    Reprographic Processes)
    Section cross-reference(s): 38, 76
FAN.CNT 1
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APPLICATION NO.
                                                                DATE
     PATENT NO.
                      KIND DATE
                                        -----
                                                               -----
    WO 2000068739 A1 20001116 WO 2000-EP4009 20000504
PΙ
        W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR,
            CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU,
            ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU,
            LV, MA, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE,
            SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA,
            ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
        RW: GH, GM, KE, LS, MW, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE,
            DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF,
            CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
                            20001116
    CA 2369025
                        AA
                                        CA 2000-2369025
                                                                 20000504
    EP 1177478
                        A1
                              20020206
                                        EP 2000-931115
                                                               20000504
            AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
            IE, SI, LT, LV, FI, RO
    JP 2002544542
                        T2 20021224
                                          JP 2000-616465
                                                                 20000504
PRAI CH 1999-899
                         Α
                              19990511
    WO 2000-EP4009
                        W
                              20000504
CLASS
 PATENT NO.
              CLASS PATENT FAMILY CLASSIFICATION CODES
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               ____
WO 2000068739
                ICM
                       G03F007-027
                IPCI
                       G03F0007-027 [ICM, 7]
                IPCR
                       G03F0007-027 [I,A]; G03F0007-027 [I,C*]; H05K0003-28
                       [N,A]; H05K0003-28 [N,C*]
                ECLA
                       G03F007/027
CA 2369025
                IPCI
                       G03F0007-027 [ICM, 7]
EP 1177478
                IPCI
                       G03F0007-027 [ICM, 6]
                IPCR
                       G03F0007-027 [I,A]; G03F0007-027 [I,C*]; H05K0003-28
                       [N, A]; H05K0003-28 [N, C*]
JP 2002544542
                IPCI
                       G03F0007-027 [ICM,7]; G03F0007-027 [ICS,7]; C08F0290-00
                       [ICS,7]; C08G0079-04 [ICS,7]; C08G0079-00 [ICS,7,C*];
                       H05K0003-28 [ICS,7]
                IPCR
                       G03F0007-027 [I,A]; G03F0007-027 [I,C*]; H05K0003-28
                       [N, A]; H05K0003-28 [N, C*]
AB
    A photosensitive polymerizable composition comprises at least one cyclic and/or
    oligomeric compound composed of structural units of formula [N=P(OR)2]n (R =
    -(A)-O-C(O)-C(R1)=CH2; R1 = H, Me; A = transition group; and n = 3-18),
    and the use of this composition as photostructurable solder stopping resist for
    the production of solder masks for printed circuit boards.
ST
    photosensitive compn solder resist photoresist
TT
    Photoresists
    Solder resists
        (photosensitive composition for resists)
IT
    92832-53-6 306971-23-3, Ethoxylated trimethylolpropane
    triacrylate-tris(2-hydroxyethyl) isocyanurate triacrylate copolymer
    RL: TEM (Technical or engineered material use); USES (Uses)
        (photosensitive composition for resists)
RE.CNT
             THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD
(1) Canon Kabushiki Kaisha; EP 0464749 A 1992 CAPLUS
(2) Fujitsu Limited; EP 0453237 A 1991 CAPLUS
(3) Nippon Kayaku Kk; JP 06228253 A 1994 CAPLUS
L9
    ANSWER 14 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
    2000:392898 CAPLUS
ΑN
DN
    133:18892
    Entered STN: 14 Jun 2000
ΕD
ΤI
    Functional coat films with long service life, articles bearing the films
    and method for their manufacture
ΙN
    Higuchi, Yoshiki; Harada, Eiji; Nojima, Takayuki; Omura, Hiroshi
PA
    Nippon Oil and Fats Co., Ltd., Japan
SO
     Jpn. Kokai Tokkyo Koho, 12 pp.
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CODEN: JKXXAF

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LA
    Japanese
IC
    ICM B32B027-00
    ICS C08J007-04
CC
     42-10 (Coatings, Inks, and Related Products)
     Section cross-reference(s): 38
FAN.CNT 1
                       KIND
     PATENT NO.
                               DATE
                                        APPLICATION NO.
                                                               DATE
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                                          -----
                               -----
                                                                _____
PI JP 2000158594
                       A2
                               20000613
                                          JP 1998-333822
                                                                 19981125
PRAI JP 1998-333822
                               19981125
CLASS
PATENT NO.
              CLASS PATENT FAMILY CLASSIFICATION CODES
 _____
                _____
JP 2000158594 ICM
                       B32B027-00
                ICS
                       C08J007-04
                IPCI
                       B32B0027-00 [ICM, 7]; C08J0007-04 [ICS, 7]
                       B32B0027-00 [I,A]; B32B0027-00 [I,C*]; C08J0007-00
                       [I,C*]; C08J0007-04 [I,A]
    The films are coated on a substrate as an under layer and a top layer
AΒ
    which is chemical bonded to the under layer by 3,3',4,4'-tetra(tert-
     butylperoxycarbonyl)benzophenone (I) compound where the under layer is
     derived from radical-polymerizable monomers and the top layer is derived
     from functional monomers. Thus, coating a mixture of I 0.5, PPZ
     (phosphazene methacrylate derivative) 1.5 and propylene glycol monomethyl
     ether 98% on the surface of an acrylic resin panel to dry thickness of 0.5
     μm, drying, irradiating with UV light, coating on top with a mixture of
     N,N-dimethylacrylamide 20 and water 80% and irradiating with UV light gave
     a coated panel with a transparent film having pencil hardness 2H and good
     adhesion and resistance to fogging.
     antifogging coating reactive multilayer acrylic polymer; plastic molding
ST
     reactive multilayer acrylic polymer coating; butylperoxycarbonyl
   benzophenone deriv reactive multilayer acrylic polymer coating
    Acrylic polymers, miscellaneous
ΙT
     Polycarbonates, miscellaneous
     Polyesters, miscellaneous
     RL: MSC (Miscellaneous)
        (substrate; functional coat films with long service life, articles
       bearing films and method for manufacture)
     272119-32-1P, N,N-Dimethylacrylamide-hexakis (methacryloyloxyethoxy
ΙT
     )cyclotriphosphazene copolymer 272119-33-2P, Sodium
     2-acrylamido-2-methylpropanesulfonate-Hexakis (methacryloyloxyethoxy) cyclot
     riphosphazene copolymer 272119-34-3P, KBM 503-NK Ester A
     600-hexakis (methacryloyloxyethoxy) cyclotriphosphazene-sodium
     methacrylate-trimethylolpropane triacrylate copolymer
                                                           272119-35-4P,
     Blemmer QA-trimethylolpropane triacrylate copolymer 272119-36-5P
     , KBM 503-2-(perfluorohexyl)ethyl acrylate-Hexakis(methacryloyloxyethoxy)c
     yclotriphosphazene copolymer 272119-37-6P,
     Hexakis (methacryloyloxyethoxy) cyclotriphosphazene-KBM 503-sodium
     2-acrylamido-2-methylpropanesulfonate copolymer 272119-38-7P, NK
     Ester A 600-N, N-dimethylacrylamide-hexakis (methacryloyloxyethoxy) cyclotrip
     hosphazene-trimethylolpropane triacrylate copolymer 273221-38-8P
     , Hexakis (methacryloyloxyethoxy) cyclotriphosphazene-trioctyl (4-
     vinylbenzyl)phosphonium chloride) copolymer
     RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or
     engineered material use); PREP (Preparation); USES (Uses)
        (functional coat films with long service life, articles bearing films
       and method for manufacture)
     77473-08-6, 3,3',4,4'-Tetra(tert-butylperoxycarbonyl)benzophenone
IΤ
     RL: CAT (Catalyst use); USES (Uses)
        (polymerization initiator; functional coat films with long service life,
       articles bearing films and method for manufacture)
     25038-59-9, PET polyester, miscellaneous
TT
     RL: MSC (Miscellaneous)
```

(substrate; functional coat films with long service life, articles

DT

Patent

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ANSWER 15 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
AN
     2000:219074 CAPLUS
DN
     132:271662
     Entered STN: 05 Apr 2000
ED
TΤ
     Photosensitive, heat-resistant resin composition for forming patterns
     Tani, Motoaki; Horikoshi, Eiji; Watanabe, Isao; Miyahara, Shoichi; Ito,
IN
     Takashi; Sasaki, Makoto
PΑ
     Fujitsu Ltd., Japan
     U.S., 19 pp., Cont.-in-part of U.S. Ser. No. 684,271, abandoned.
SO
     CODEN: USXXAM
DT
     Patent
LA
     English
     ICM G03C001-73
IC
INCL 430285100
     74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other
     Reprographic Processes)
FAN.CNT 4
     PATENT NO.
                         KIND DATE
                                             APPLICATION NO.
                                                                        DATE
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                                20000404 US 1993-64664
     US 6045975
                           A
                                                                        19930521
PΙ
AZ 19920122 JP 1990-279088

JP 2980359 B2 19991122

JP 05323625 A2 19931207 JP 1992-130837

JP 2911298 B2 19990623

US 6013419 A 20000111 US 1995-418169

US 5972807 A 19991026 US 1996-772260

PRAI JP 1990-99788 A 19900416

JP 1990-279088 A 19900019

US 1991-684271 B2 19910412

JP 1992-130837 A 19920522

US 1993-64664 A3 19930521
                          A2 19920122 JP 1990-279088
     JP 04018450
                                                                        19901019
                                                                        19920522
                                                                        19950406
                                             US 1996-772260
                                                                        19961223
     JP 1992-130837
                         A3 19930521
A 19930910
B1 19931123
                                19930521
     JP 1993-248499
     US 1993-155697
US 1994-303931
                          B1 19940909
CLASS
                CLASS PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
                  ICM
 US 6045975
                          G03C001-73
                  INCL
                          430285100
                  IPCI
                          G03C0001-73 [ICM, 7]
                          G03F0007-032 [I,C*]; G03F0007-037 [I,A]; H05K0001-00
                  IPCR
                          [I,A]; H05K0001-00 [I,C*]; H05K0003-00 [I,A];
                          H05K0003-00 [I,C*]; H05K0003-38 [N,A]; H05K0003-38
                          [N,C*]; H05K0003-46 [N,A]; H05K0003-46 [N,C*]
                          430/285.100; 430/287.100; 430/288.100; 430/906.000;
                  NCL
                          430/920.000; 430/925.000
                          G03F007/037; H05K001/00B2; H05K003/00K3F
                  ECLA
                          C08L0079-08 [ICM,5]; C08L0079-00 [ICM,5,C*];
 JP 04018450
                  IPCI
                          C08G0073-10 [ICS,5]; C08G0073-00 [ICS,5,C*];
                          C09D0004-02 [ICS,5]; C09D0179-08 [ICS,5]; C09D0179-00
                          [ICS, 5, C*]
                          G03G0005-05 [ICM, 5]
 JP 05323625
                  IPCI
                          G03C0005-16 [ICM, 6]
                  I.PCI
 US 6013419
                          G03F0007-032 [I,C*]; G03F0007-037 [I,A]; H05K0003-00
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                          [N,A]; H05K0003-00 [N,C*]; H05K0003-38 [N,A];
                          H05K0003-38 [N,C*]; H05K0003-46 [N,A]; H05K0003-46
                          [N,C^*]
                          430/325.000; 430/311.000; 430/330.000
                  NCL
                  ECLA
                          G03F007/037
                          C08G0073-10 [ICM, 6]; C08G0073-00 [ICM, 6, C*];
 US 5972807
                  IPCI
                          C08L0079-08 [ICS,6]; C08L0079-00 [ICS,6,C*]
                  IPCR
                          H05K0001-00 [I,A]; H05K0001-00 [I,C*]; H05K0003-00
                          [N,A]; H05K0003-00 [N,C*]; H05K0003-38 [N,A];
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NCL
                        442/063.000; 442/069.000; 442/070.000; 442/164.000;
                        525/180.000; 525/183.000
                        H05K001/00B2
                 ECLA
AB
    A film-forming, photosensitive, heat-resistant resin composition comprises a
     varnish of a polyimide precursor having no photosensitivity in itself, a
     polymerizable monomer or oligomer compatible with the varnish and capable
     of providing a high-heat-resistant polymer upon being polymerized, and a
     polymerization initiator for the monomer or oligomer. The resin composition
is useful
     for the production of circuit substrates and semiconductor devices for high-d.
     mounting including multi-chip modules or the like, such as printed
     circuits, printed boards, wiring boards, and electronic components, since
     it can effectively avoid a reduction of the layer thickness during the film
     formation, and ensures a low cost production process. The pattern formation
     process using the resin composition is also disclosed.
ST
     photosensitive heat resistant resin compn printed circuit board
ΙT
     Photoimaging materials
        (containing polyimide precursor varnishes for forming heat-resistant
        patterns)
     Soldering
ΙT
        (masks; photosensitive compns. containing polyimide precursor varnishes for
        preparation of)
TΤ
     Polyamic acids
     RL: TEM (Technical or engineered material use); USES (Uses)
        (photosensitive compns. for forming heat-resistant images containing)
TΤ
     Printed circuit boards
        (photosensitive compns. for forming heat-resistant images for
        fabrication of)
ΙT
     25038-81-7
     RL: TEM (Technical or engineered material use); USES (Uses)
        (photosensitive compns. for forming heat-resistant images containing
        photopolymerizable compds. and)
                                               15625-89-5, Trimethylolpropane
TT
     106-90-1, Glycidyl acrylate
                                   3524-68-3
                   29570-58-9, Dipentaerythritol hexaacrylate
     triacrylate
                                                                40220-08-4,
                                                     100844-79-9
     Tris(acryloyloxyethyl)isocyanurate 92832-53-6
                                263169-76-2
                                               263169-78-4
     137741-82-3
                   251636-01-8
     RL: TEM (Technical or engineered material use); USES (Uses)
        (photosensitive compns. for forming heat-resistant images containing
        polyimide precursors and)
                                      103-01-5, N-Phenylglycine
ΙT
     94-36-0, Benzoyl peroxide, uses
                                                                   119-61-9.
                                      6542-67-2, 2,4,6-Tris(trichloromethyl)-
     Benzophenone, uses
                          2040-14-4
                      6652-28-4, Isopropyl benzoin ether 7189-82-4,
     1,3,5-triazine
     2,2'-Bis(o-chlorophenyl)-4,4',5,5'-tetraphenyl-1,2'-diimidazole
     15774-82-0, 2-Methylthioxanthone 24504-22-1
                                                                  77473-08-6,
                                                     24650-42-8
     3,3',4,4'-Tetra(tert-butylperoxycarbonyl)benzophenone
     RL: TEM (Technical or engineered material use); USES (Uses)
        (photosensitive compns. for forming heat-resistant images containing
        polyimide precursors, photopolymerizable compds. and)
RE.CNT
              THERE ARE 27 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE
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(2) Anon; DE 2914619 1979 CAPLUS
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(8) Anon; JP 54-162798 World Patents Index 1979 CAPLUS
(9) Cescon; US 3615454 1971 CAPLUS
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(11) Giles Jr; US 4395518 1983 CAPLUS
(12) Goff; US 4416973 1983 CAPLUS
(13) Goff; US 4430418 1984 CAPLUS
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H05K0003-38 [N,C*]; H05K0003-46 [N,A]; H05K0003-46

[N,C*]

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(14) Goff; US 4454220 1984 CAPLUS
(15) Hiramoto; US 4243743 1981 CAPLUS
(16) Horikoshi; ICEMM Proceedings 1993, P266
(17) Ide; US 4048035 1977
(18) Ide; US 4537855 1985 CAPLUS
(19) Irving; US 4579809 1986 CAPLUS
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(21) Kogyo Chosakai Publishing Co; Photopolymer Handbook 1989, P270
(22) Nagata; US 5438105 1995 CAPLUS
(23) St Clair; US 4389504 1983 CAPLUS
(24) Tani; US 08418169 1995
(25) Tani; US 08772260 1996
(26) Tani; Mat Res Soc Symp Proc 1992, V264, P37 CAPLUS
(27) Yoda and Hiramoto; J Macromol SCL-CHEM 1984, P1641
L9
    ANSWER 16 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
ΑN
    2000:53791 CAPLUS
    132:109482
DN
ED
    Entered STN: 23 Jan 2000
ΤI
    Weather-resistant compositions and coating materials and molded articles
ΙN
    Ogawa, Takashi; Mori, Koji; Akada, Mitsuo
PΑ
    Otsuka Chemical Co., Ltd., Japan
SO
    PCT Int. Appl., 58 pp.
    CODEN: PIXXD2
DT
    Patent
LA
     Japanese
IC
     ICM C09D004-02
     ICS C08K005-3475; C08L033-04; C09D007-12; C07D249-20; C09K003-00
CC
     42-10 (Coatings, Inks, and Related Products)
     Section cross-reference(s): 55
FAN.CNT 1
    PATENT NO.
                      KIND DATE
                                        APPLICATION NO.
                                                               DATE
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    WO 2000002964
                                          WO 1999-JP3715
                       A1 20000120
                                                               19990708
PΙ
        W: AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ,
            DE, DK, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS,
            JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK,
            MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ,
            TM, TR, TT, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ,
            MD, RU, TJ, TM
        RW: GH, GM, KE, LS, MW, SD, SL, SZ, UG, ZW, AT, BE, CH, CY, DE, DK,
            ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG,
            CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
    AU 9946504
                        A1
                              20000201
                                        AU 1999-46504
                                                                 19990708
                                                               19990708
                                         EP 1999-929769
    EP 1104790
                        A1
                              20010606
        R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
            IE, SI, LT, LV, FI, RO
                                          US 2001-720974
     US 6703139
                      В1
                               20040309
                                                                 20010103
PRAI JP 1998-196099
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                               19980710
    WO 1999-JP3715
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                              19990708
CLASS
                CLASS PATENT FAMILY CLASSIFICATION CODES
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WO 2000002964
                ICM
                       C09D004-02
                       C08K005-3475; C08L033-04; C09D007-12; C07D249-20;
                ICS
                       C09K003-00
                IPCI
                       C09D0004-02 [ICM,6]; C08K0005-3475 [ICS,6]; C08K0005-00
                       [ICS, 6, C*]; C08L0033-04 [ICS, 6]; C08L0033-00
                       [ICS, 6, C*]; C09D0007-12 [ICS, 6]; C07D0249-20 [ICS, 6];
                       C07D0249-00 [ICS, 6, C*]; C09K0003-00 [ICS, 6]
                IPCR
                       C08K0005-00 [I,C*]; C08K0005-3475 [I,A]; C09D0007-12
                       [I,A]; C09D0007-12 [I,C*]
                       C08K005/3475; C09D007/12D4A; C09D007/12D4
                ECLA
                       C08K0005-00 [I,C*]; C08K0005-3475 [I,A]; C09D0007-12
AU 9946504
                IPCR
                       [I,A]; C09D0007-12 [I,C*]
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EP 1104790
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                        C09D0004-02 [ICM, 6]; C08K0005-3475 [ICS, 6]; C08K0005-00
                        [ICS, 6, C*]; C08L0033-04 [ICS, 6]; C08L0033-00
                        [ICS, 6, C*]; C09D0007-12 [ICS, 6]; C07D0249-20 [ICS, 6];
                        C07D0249-00 [ICS, 6, C*]; C09K0003-00 [ICS, 6]
                        C08K0005-00 [I,C*]; C08K0005-3475 [I,A]; C09D0007-12
                 IPCR
                        [I,A]; C09D0007-12 [I,C*]
                        C08K005/3475; C09D007/12D4A; C09D007/12D4
                 ECLA
 US 6703139
                        B32B0027-00 [ICM,7]; C08F0126-06 [ICS,7]; C08F0126-00
                 IPCI
                        [ICS,7,C*]; C08F0112-14 [ICS,7]; C08F0112-00
                        [ICS,7,C*]; C08F0220-10 [ICS,7]; C08F0220-00
                        [ICS, 7, C*]; C08F0216-12 [ICS, 7]; C08F0216-00 [ICS, 7, C*]
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                        C08K0005-00 [I,C*]; C08K0005-3475 [I,A]; C09D0007-12
                        [I,A]; C09D0007-12 [I,C*]
                 NCL
                        428/500:000; 526/261.000; 526/313.000; 526/328.500;
                        526/332.000
                 ECLA
                        C08K005/3475; C09D007/12D4A; C09D007/12D4
OS
    MARPAT 132:109482
AΒ
    The title compns. contain the reaction products of
    bisbenzotriazolylphenols with lactones, (meth)acrylates, and curing
             Thus, a coating composition contained Aronix M 8030 40,
     tetrahydrofurfuryl acrylate 20, trimethylolpropane triacrylate 40,
     ε-caprolactone-2,2'-methylenebis[6-(2H-1,2,3-benzotriazole-2-y1)-4-
     (2-hydroxyethyl)phenol] copolymer 3, and Darocur 1173 3 parts.
ST
     weather resistant coating UV absorber; benzotriazolylhydroxyethylphenol
     lactone copolymer UV absorber; polyester benzotriazolylhydroxyethylphenol
     caprolactone UV absorber
ΙT
     Coating materials
        (abrasion-resistant; weather-resistant coating materials and moldings
        containing bisbenzotriazolylhydroxyethylphenol lactone copolymers)
ΙT
     Polymerization catalysts
        (photopolymn.; weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
IT
     Vinyl compounds, uses
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (polymers; weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
ΙT
     Polymerization catalysts
        (radical; weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
ΙT
     Polymerization
        (ring-opening; weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
IT
     Crosslinking agents
     Lacquers
     UV stabilizers
        (weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
IT
     Lactones
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
ΙT
     Coating materials
        (weather-resistant; weather-resistant coating materials and moldings
        containing bisbenzotriazolylhydroxyethylphenol lactone copolymers)
ΙT
     7473-98-5, Darocur 1173
     RL: CAT (Catalyst use); USES (Uses)
        (Darocur 1173; weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
IT
     947-19-3, Irgacure 184
     RL: CAT (Catalyst use); USES (Uses)
        (Irgacure 184; weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
ΙT
     71868-10-5, Irgacure 907
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(Irgacure 907; weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
Τ'n
     78-67-1, AIBN
                   94-36-0, Benzoyl peroxide, uses
     RL: CAT (Catalyst use); USES (Uses)
        (weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
     255384-16-8P 255384-17-9P 255384-18-0P 255384-20-4P 255384-21-5P 255827-82-8P, &-Caprolactone-2,2'-methylenebis[6-
IT
     (2H-1,2,3-benzotriazole-2-yl)-4-(2-hydroxyethyl)phenol] copolymer
     RL: IMF (Industrial manufacture); MOA (Modifier or additive use); PREP
     (Preparation); USES (Uses)
        (weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
ΙT
     38808-38-7P, Butyl acrylate-methacrylic acid-methyl methacrylate-vinyl
     acetate copolymer 77492-22-9P, Acrylic acid-butyl acrylate-2-ethylhexyl
     methacrylate-hydroxyethyl methacrylate-methyl methacrylate-styrene
     copolymer
                172156-26-2P, Acrylic acid-butyl acrylate-2-ethylhexyl
     methacrylate-hydroxyethyl methacrylate-styrene copolymer
                                                                255820-10-1P,
     Acrylic acid-butyl acrylate-2-ethylhexyl methacrylate-formaldehyde-
     hydroxyethyl methacrylate-melamine-methyl methacrylate-styrene copolymer
     255820-11-2P, Acrylic acid-butyl acrylate-2-ethylhexyl
     methacrylate-formaldehyde-hydroxyethyl methacrylate-melamine-styrene
     copolymer
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (weather-resistant coating materials and moldings containing
        bisbenzotriazolylhydroxyethylphenol lactone copolymers)
RE.CNT 9
             THERE ARE 9 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE
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(4) Nippon Paint Co Ltd; WO 9522250 A1
(5) Nippon Paint Co Ltd; JP 10503787 A 1998
(6) Otsuka Chemical Co Ltd; US 5922882 A CAPLUS
(7) Otsuka Chemical Co Ltd; EP 855393 A1 CAPLUS
(8) Otsuka Chemical Co Ltd; WO 9735847 A1 CAPLUS
(9) Otsuka Chemical Co Ltd; JP 09316060 A 1997 CAPLUS
     ANSWER 17 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
     2000:43472 CAPLUS
AN
     132:109425
DN
ED
     Entered STN: 18 Jan 2000
TΙ
     Coating compositions with good resistance to metal ion-induced
     discoloration and weather and UV absorbents for use in the compositions
     Ogawa, Takashi; Akada, Mitsuo; Mori, Hiroshi
IN
PΑ
     Ohtsuka Chemical Co., Ltd., Japan
     Jpn. Kokai Tokkyo Koho, 22 pp.
SO
     CODEN: JKXXAF
DT
     Patent
LA
     Japanese
IC
     ICM C09D005-32
     ICS C09D004-02; C09D007-12; C09D163-10; C09D167-07; C09D171-00;
          C09D175-16
ICA C08F290-06; C09K003-00
     42-5 (Coatings, Inks, and Related Products)
CC
FAN.CNT 1
     PATENT NO.
                        KIND
                                            APPLICATION NO.
                                DATE
                                                                   DATE
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     JP 2000017204
                         A2
                                20000118
                                            JP 1998-186543
                                                                   19980701
                        B2
     JP 2918543
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PRAI JP 1998-186543
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CLASS
 PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
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RL: CAT (Catalyst use); USES (Uses)

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C09D005-32
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                                       C09D171-00; C09D175-16
                                       C08F290-06; C09K003-00
                           ICA
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                                       C09D0005-32 [ICM, 6]; C09D0004-02 [ICS, 6]; C09D0007-12
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                                       C09D0171-00 [ICS,6]; C09D0175-16 [ICS,6]; C08F0290-06
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                                       C09D0005-32 [I,C*]; C09D0007-12 [I,A]; C09D0007-12
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                                       C09D0167-06 [I,C*]; C09D0167-07 [I,A]; C09D0171-00
                                       [I,A]; C09D0171-00 [I,C*]; C09D0175-14 [I,C*];
                                       C09D0175-16 [I,A]; C09D0201-00 [I,A]; C09D0201-00
                                       [I,C*]; C09K0003-00 [N,A]; C09K0003-00 [N,C*]
OS
        MARPAT 132:109425
AB
        The coating compns. contain (A) radically polymerizable monomers or/and
        curable resins, and bisbenzotriazolylphenol compds. Thus, mixing Magicron
        TC 16U Clear (aminoacrylic clear coating) with 2 phr RUVA-100
        \{2,2'-methylenebis[6-(2H-1,2,3-benzotriazol-2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-yl)-4-(2-
        hydroxyethyl)phenol]} gave a clear top coating composition for protecting
        prefinished metal sheet.
        multilayer coating clear coating UV absorbent bisbenzotriazolylphenol;
ST
        light stabilizer bisbenzotriazolylphenol clear coating; metal ion
        discoloration resistance coating
TΤ
        Epoxy resins, uses
        RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP
        (Properties); TEM (Technical or engineered material use); PREP
        (Preparation); USES (Uses)
             (acrylic; coating compns. with good resistance to metal ion-induced
             discoloration and weather and UV absorbents for use in compns.)
IT
        Aminoplasts
        RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP
        (Properties); TEM (Technical or engineered material use); PREP
         (Preparation); USES (Uses)
             (clear coating; coating compns. with good resistance to metal
             ion-induced discoloration and weather and UV absorbents for use in
             compns.)
TT
        UV stabilizers
             (coating compns. with good resistance to metal ion-induced
            discoloration and weather and UV absorbents for use in compns.)
IT
        Coating materials
             (transparent; coating compns. with good resistance to metal ion-induced
             discoloration and weather and UV absorbents for use in compns.)
IT
        Epoxy resins, uses
        RL: TEM (Technical or engineered material use); USES (Uses)
             (under coating; coating compns. with good resistance to metal
             ion-induced discoloration and weather and UV absorbents for use in
             compns.).
        Coating materials
IΤ
             (weather-resistant; coating compns. with good resistance to metal
             ion-induced discoloration and weather and UV absorbents for use in
             compns.)
                                                                                   38808-38-7P
IT
        9003-08-1P, Formaldehyde-melamine copolymer
                                                                                                          51097-42-8P,
        Acrylic acid-butyl methacrylate-2-ethylhexyl acrylate-hydroxyethyl
                                                             63747-55-7P, Acrylic acid-butyl
        methacrylate-styrene copolymer
        methacrylate-2-ethylhexyl acrylate-hydroxyethyl methacrylate-methyl
        methacrylate-styrene copolymer
                                                             255384-16-8P, Aronix M-8030-
        tetrahydrofurfuryl acrylate-trimethylolpropane triacrylate copolymer
        255384-17-9P, 2-(2-Ethoxyethoxy)ethyl acrylate-pentaerythritol
        triacrylate-Viscoat 3700 copolymer
                                                                    255384-18-0P
                                                                                             255384-19-1P, Art
        Resin UN 3320HA-pentaerythritol triacrylate-pentaerythritol
        trimethacrylate-tetrahydrofurfuryl acrylate copolymer 255384-20-4P
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trimethacrylate-Ripoxy SP 1509 copolymer 255384-21-5P
    RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP
     (Properties); TEM (Technical or engineered material use); PREP
     (Preparation); USES (Uses)
        (clear coating; coating compns. with good resistance to metal
       ion-induced discoloration and weather and UV absorbents for use in
       compns.)
ΙT
    196516-61-7, RUVA-100 196516-62-8 196516-63-9
                                                      196516-64-0
    RL: MOA (Modifier or additive use); USES (Uses)
       (light stabilizer; coating compns. with good resistance to metal
       ion-induced discoloration and weather and UV absorbents for use in
       compns.)
    255390-94-4, Magicron TC 16U Clear 255390-95-5, Magicron TA 16
IT
    RL: POF (Polymer in formulation); PRP (Properties); TEM (Technical or
    engineered material use); USES (Uses)
       (top clear coating; coating compns. with good resistance to metal
       ion-induced discoloration and weather and UV absorbents for use in
       compns.)
    ANSWER 18 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
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    1998:502650 CAPLUS
AN
DN . 129:224371
    Entered STN: 13 Aug 1998
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TΙ
    Ionic conductor and its preparation
    Okuo, Masaki; Hikuchi, Keiki; Omura, Hiroshi; Suyama, Shuji
IN
PΑ
    Nippon Oil and Fats Co., Ltd., Japan
    Jpn. Kokai Tokkyo Koho, 19 pp.
SO
    CODEN: JKXXAF
DT
    Patent
LA
    Japanese
    ICM H01B001-12
IC
    ICS C08L051-00; C08F002-46
CC
    76-2 (Electric Phenomena)
    Section cross-reference(s): 35, 38, 72
FAN.CNT 1
    PATENT NO.
                      KIND
                                        APPLICATION NO. DATE
                              DATE
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    JP 10208542
                       A2 19980807 JP 1997-8975
                                                               19970121
PΙ
PRAI JP 1997-8975
                              19970121
CLASS
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
               ____
 _____
                ICM H01B001-12
 JP 10208542
                ICS
                      C08L051-00; C08F002-46
                IPCI H01B0001-12 [ICM, 6]; C08L0051-00 [ICS, 6]; C08F0002-46
                      [ICS, 6]
    The conductor comprises (A) a support, (B) a polymer layer containing
AB
    photopolymn. initiating group, and (C) a ion conducting layer formed from
    a composition containing (a) ion conducting monomer and (b) soluble
electrolyte salt
    and/or electrolyte salt monomer, which is bonded with B via photopolymn.
    initiating group. B is formed on A, a composition containing ion conducting
    monomer and soluble electrolyte salt is contacted with B with irradiation of
    active energy to give the title ionic conductor. Ionic conductor layers
    having excellent peeling resistance and high strength are obtained.
    ionic conductor peeling resistant; photopolymn ionic conductor coating
ST
ΙT
    Polyoxyalkylenes, uses
    Polyoxyalkylenes, uses
    RL: IMF (Industrial manufacture); TEM (Technical or engineered material
    use); PREP (Preparation); USES (Uses)
        (acrylic-polyester-; graft copolymer ionic conductor layer and its
       preparation)
ΙT
    Polyesters, uses
    Polyesters, uses
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, ${\tt Hexa}\,(2{\tt -methacryloyloxyethoxy})\,{\tt cyclotriphosphazene-pentaerythritol}$

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (acrylic-polyoxyalkylene-; graft copolymer ionic conductor layer and its preparation) Polyoxyalkylenes, uses RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (acrylic; graft copolymer ionic conductor layer and its preparation) Coating materials (elec. conductive; graft copolymer ionic conductor layer and its preparation) Conducting polymers Ionic conductors (graft copolymer ionic conductor layer and its preparation) Polymers, uses RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (graft; graft copolymer ionic conductor layer and its preparation) Solid electrolytes (ionic graft copolymer conductor layer and its preparation) Polyurethanes, uses RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (polyoxyalkylene-, acrylic; graft copolymer ionic conductor layer and its preparation) 7791-03-9, Lithium perchlorate RL: MOA (Modifier or additive use); USES (Uses) (4; graft copolymer ionic conductor layer and its preparation) 13641-97-9DP, Acryloyl isocyanate, reaction products with acryloyl isocyanate, polymer with polyethylene oxides and (meth)acrylates, potassium complex RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (MAI; graft copolymer ionic conductor layer and its preparation) 56-81-5DP, 1,2,3-Propanetriol, reaction product with ethylene glycol and acrylic acid, polymer with polyethylene oxides and (meth)acrylates, uses 79-10-7DP, 2-Propenoic acid, reaction product with ethylene glycol and glycerin, polymer with polyethylene oxides and (meth)acrylates, uses 79-41-4DP, polymer with polyethylene oxides and (meth)acrylates 80-62-6DP, polymer with polyethylene oxides and (meth)acrylates 106-91-2DP, reaction products with partially saponified poly(vinyl acetate), polymer with polyethylene oxides and (meth)acrylates 107-21-1DP, 1,2-Ethanediol, reaction product with glycerin and acrylic acid, polymer with polyethylene oxides and (meth)acrylates, uses 2680-03-7DP, polymer with polyethylene oxides and (meth)acrylates, potassium complex 3524-68-3DP, Kayarad PET-30, polymer with polyethylene oxides and (meth)acrylates, potassium complex 6900-35-2DP, Potassium methacrylate, polymer with polyethylene oxides and (meth)acrylates, potassium complex 7439-93-2DP, Lithium, complexes with ion-conducting polymers, uses 7440-09-7DP, Potassium, complexes with ion-conducting polymers, uses 7440-50-8DP, Copper, complexes with ion-conducting polymers, uses 14798-03-9DP, Ammonium, complexes with ion-conducting polymers, uses 16325-47-6DP, Ammonium methacrylate, polymer with polyethylene oxides and 25805-17-8DP, Polyethyloxazoline, reaction products with (meth)acrylates acryloyl isocyanate, polymer with polyethylene oxides and (meth)acrylates, potassium complex 25852-47-5DP, polymer with polyethylene oxides and (meth)acrylates, potassium complex 26570-48-9DP, polymer with polyethylene oxides and (meth)acrylates 69488-61-5DP, Poly[(propionylimino)ethylene], reaction products with acryloyl isocyanate, polymer with polyethylene oxides and (meth)acrylates, 101027-21-8DP, Gohseran L 302, reaction product with potassium complex qlycidyl methacrylate, polymer with polyethylene oxides and (meth)acrylates 120487-52-7DP, Eleminol RS 30, polymer with polyethylene oxides and (meth)acrylates 203309-79-9DP, 1-[4-[2-[2-

(Methacryloyloxy) ethoxycarbonyloxy] ethoxy] phenyl] -2-hydroxy-2-methylpropan-

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1-one, polymer with polyethylene oxides and (meth)acrylates
212397-55-2DP, Butyl methacrylate-lauryl methacrylate-1-[4-[2-[2-
(methacryloyloxy)ethoxycarbonyloxy]ethoxy]phenyl]-2-hydroxy-2-methylpropan-
1-one-methyl methacrylate-polyethylene glycol diacrylate graft copolymer,
lithium complex
                  212397-63-2DP, polymer with polyethylene oxides and
(meth)acrylates
                  212397-64-3P, Cerium methacrylate-1-[4-[2-[2-
(methacryloyloxy) ethoxycarbonyloxy] ethoxy] phenyl] -2-hydroxy-2-methylpropan-
1-onemethyl methacrylate-polyethylene glycol diacrylate-potassium
methacrylate-vinylpyrrolidone graft copolymer
                                                212397-65-4P,
1-[4-[2-[2-(Methacryloyloxy)ethoxycarbonyloxy]ethoxy]phenyl]-2-hydroxy-2-
methylpropan-1-one-methyl methacrylate-polypropylene glycol
dimethacrylate-Eleminol RS 30 graft copolymer 212397-66-5DP,
1-[4-[2-[2-(Methacryloyloxy)ethoxycarbonyloxy]ethoxy]phenyl]-2-hydroxy-2-
methylpropan-1-one-methyl methacrylate-vinylpyridine graft copolymer,
lithium complex
                 212397-67-6DP, 1-[4-[2-(Methacryloyloxy)ethoxycarbony]
loxy]ethoxy]phenyl]-2-hydroxy-2-methylpropan-1-one-methyl
methacrylate-polyethylene glycol diacrylate-polyethylene glycol
dimethacrylate graft copolymer, complex with lithium and potassium
212397-67-6DP, 1-[4-[2-(Methacryloyloxy)ethoxycarbonyloxy]ethoxy]phenyl
]-2-hydroxy-2-methylpropan-1-one-methyl methacrylate-polyethylene glycol
diacrylate-polyethylene glycol dimethacrylate graft copolymer, potassium
         212397-68-7DP, 1-[4-[2-(Methacryloyloxy)] ethoxycarbonyloxy] eth
complex
oxy]phenyl]-2-hydroxy-2-methylpropan-1-one-KBM 503-methyl
methacrylate-polyethylene glycol diacrylate-polyethylene glycol
dimethacrylate graft copolymer, potassium complex
                                                    212397-69-8DP,
N, N-Dimethylacrylamide-1-[4-[2-(methacryloyloxy)] ethoxycarbonyloxy] ethox
y]phenyl]-2-hydroxy-2-methylpropan-1-one-methyl methacrylate-
pentaerythritol triacrylate-polyethylene glycol diacrylate-polyethylene
glycol dimethacrylate graft copolymer, potassium complex
212397-70-1DP, 3,3'-Bis (methacryloyloxyethoxycarbonyl)-4,4'-
bis(tert-butylperoxycarbonyl)benzophenone-polyethylene glycol
diacrylate-Hexakis (methacryloyloxyethoxy) cyclotriphosphazene graft
copolymer, lithium complex 212515-88-3DP, Glycidyl methacrylate-1-[4-[2-
[2-(methacryloyloxy)ethoxycarbonyloxy]ethoxy]phenyl]-2-hydroxy-2-
methylpropan-1-one-methyl methacrylate-polyethylene glycol
diacrylate-propylene glycol monomethyl ether-triethylene tetramine graft
                             212520-79-1DP, ammonium complex
copolymer, lithium complex
212520-81-5DP, Hexamethylene diisocyanate polyethylene glycol copolymer
diacrylate-1-[4-[2-[2-(methacryloyloxy)ethoxycarbonyloxy]ethoxy]phenyl]-2-
hydroxy-2-methylpropan-1-one-methyl methacrylate-polyethylene glycol
diacrylate graft copolymer, copper complex
                                            212571-38-5DP, GE 3A, polymer
with polyethylene oxides and (meth)acrylates
                                              212571-90-9DP,
Acrylonitrile-GE 3A-1-[4-[2-[2-(methacryloyloxy)ethoxycarbonyloxy]ethoxy]p
henyl]-2-hydroxy-2-methylpropan-1-one-methyl methacrylate-polyethylene
glycol diacrylate-polyethylene glycol dimethacrylate graft copolymer,
potassium complex
                  212572-38-8P
RL: IMF (Industrial manufacture); TEM (Technical or engineered material
use); PREP (Preparation); USES (Uses)
   (graft copolymer ionic conductor layer and its preparation)
            7447-39-4, Copper chloride (CuCl2), uses
Potassium iodide (KI), uses
                              7778-74-7, Potassium perchlorate
RL: MOA (Modifier or additive use); USES (Uses)
   (graft copolymer ionic conductor layer and its preparation)
212397-54-1P, 1-[4-[2-[2-(Methacryloyloxy)ethoxycarbonyloxy]ethoxy]phenyl]-
2-hydroxy-2-methylpropan-1-one-methyl methacrylate-polyethylene glycol
diacrylate graft copolymer
                             212397-55-2P, Butyl methacrylate-lauryl
methacrylate-1-[4-[2-[2-(methacryloyloxy)ethoxycarbonyloxy]ethoxy]phenyl]-
2-hydroxy-2-methylpropan-1-one-methyl methacrylate-polyethylene glycol
diacrylate graft copolymer
                             212397-56-3P, Butyl methacrylate-1-[4-[2-[2-
(methacryloyloxy) ethoxycarbonyloxy] ethoxy] phenyl] -2-hydroxy-2-methylpropan-
1-one-methyl methacrylate-polyethylene glycol diacrylate graft copolymer
212397-57-4P, Dibutyl fumarate-methyl [2-[4-(2-hydroxy-2-methyl-1-
oxopropyl)phenoxy]ethyl] fumarate-polyethylene glycol diacrylate graft
           212397-58-5P, Bis[2-[4-(2-hydroxy-2-methyl-1-
copolymer
oxopropyl)phenoxy]ethyl] itaconate-dibutyl fumarate-polyethylene glycol
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ΙT

212397-59-6P, Lauryl methacrylate-1-[4-(2diacrylate graft copolymer methacryloylethoxy)phenyl]-2,2-dimethoxy-2-phenyl ethan-1-one-methyl methacrylate-polyethylene glycol diacrylate graft copolymer 212397-60-9P, 1,2-Diphenyl-1,2-ethanedione-2-O-acryloyloxime-1-[4-[2-[2-(methacryloyloxy) ethoxycarbonyloxy] ethoxy[phenyl] - 2-hydroxy-2-methylpropan-1-one-methyl methacrylate-polyethylene glycol diacrylate graft copolymer 212397-61-0P, Hydroxyethyl methacrylate-methacrylic acid-1-[4-[2-[2-(methacryloyloxy) ethoxycarbonyloxy] ethoxy] phenyl] -2-hydroxy-2-methylpropan-1-one-methyl methacrylate-polyethylene glycol diacrylate graft copolymer 212397-62-1P, Bu methacrylate-N, N-dimethylaminoethyl methacrylate-1-[4-[2-[2-(methacryloyloxy)ethoxycarbonyloxy]ethoxy]phenyl]-2-hydroxy-2methylpropan-1-one-Me methacrylate-polyethylene glycol diacrylate-ST graft copolymer RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (ionic graft copolymer conductor layer and its preparation) ANSWER 19 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN 1998:400375 CAPLUS 129:161679 Entered STN: 01 Jul 1998 Reactions of ferrocenyl amines and alcohols with hexachlorocyclotriphosphazene Allen, Christopher W.; Myer, Charles N. Department of Chemistry, University of Vermont, Burlington, VT, 05405-0125, USA Phosphorus, Sulfur and Silicon and the Related Elements (1997), 124 & 125, 431-435 CODEN: PSSLEC; ISSN: 1042-6507 Gordon & Breach Science Publishers Journal English 29-12 (Organometallic and Organometalloidal Compounds) Section cross-reference(s): 35, 72, 80 CASREACT 129:161679 The reactions of ferrocenylmethanol, ferrocenyl-2-propanol and N-methyl-2-ferrocenylmethylamine with chlorocyclophosphazenes were examined The pentachlorocyclotriphosphazene derivative of ferrocenylmethanol undergoes rapid decomposition via a phosphazene-phosphazane rearrangement, however when the alc. function is β to the cyclopentadienyl ring modest yields of N3P3Cl5OCHMeCH2C5H4FeCp may be obtained. By way of contrast N-methyl-2-ferrocenylmethylamine gives a broad range of stable derivs., N3P3C16-n[NMeCH2C5H4FeCp]n (n = 1-3). The substitution process follows a predominantly trans nongeminal pathway. The corresponding reaction with the Bu methacrylate derivative, N3P3C15O(CH2)4OC(O)CMe:CH2 leads to the unexpected geminal product, 2,2'-N3P3Cl4[O(CH2)4OC(O)CMe:CH2]NMeCH2C5H4FeC Polymers containing the 2-ferrocenylmethylamine function were obtained by reactions of poly(dichlorophosphazene) with the ferrocenylamine and by radical addition polymerization of the aforementioned mixed ferrocenylamino Bu methacrylphosphazene. [NP(NMeCH2C5H4FeCp)x(OCH2CF3)2-x]n shows potential for use in glucose sensors. The new materials were characterized by standard methodologies including 31P NMR spectroscopy, cyclic voltammetry and gel permeation chromatog. cyclotriphosphazene ferrocenylalkoxy ferrocenylmethylamino prepn; electrochem ferrocenylmethylamino cyclotriphosphazene; glucose sensor ferrocenylmethylamino polyphosphazene; ferrocene cyclotriphosphazene polyphosphazene deriv prepn; redox active phosphazene prepn; ferrocenylmethylamino cyclotriphosphazene polyphosphazene prepn electrochem Sensors (electrochem.; poly((ferrocenylmethylamino)(trifluoroethoxy)phosphazene) for glucose) Metallocenes RL: PEP (Physical, engineering or chemical process); PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); PROC

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(Process); RACT (Reactant or reagent)
        (ferrocenes; preparation of ferrocenylalkoxy cyclotriphosphazene and
preparation
        and electrochem. properties of ferrocenylmethylamino
        cyclotriphosphazenes and poly((ferrocenylmethylamino)(trifluoroethoxy)p
        hosphazene))
     Oxidation, electrochemical
IT
        (of (ferrocenylmethylamino)cyclotriphosphazenes and related
        polyphosphazenes)
ΙT
     Polyphosphazenes
     RL: ARG (Analytical reagent use); SPN (Synthetic preparation); ANST
     (Analytical study); PREP (Preparation); USES (Uses)
        (preparation and potential use as electrochem. glucose sensor of
        poly((ferrocenylmethylamino)(trifluoroethoxy)phosphazene))
ΙT
     Cyclophosphazenes
     RL: PEP (Physical, engineering or chemical process); PRP (Properties); RCT
     (Reactant); SPN (Synthetic preparation); PREP (Preparation); PROC
     (Process); RACT (Reactant or reagent)
        (preparation of ferrocenylalkoxy cyclotriphosphazene and preparation and
        electrochem. properties of ferrocenylmethylamino cyclotriphosphazenes)
ΙT
     12093-10-6, Ferrocenyl aldehyde
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (borohydride reduction and reaction with methylamine followed by
        cyanoborohydride reduction)
IT
     50-99-7, Glucose, analysis
     RL: ANT (Analyte); ANST (Analytical study)
        (electrochem. sensing using poly((ferrocenylmethylamino)(trifluoroethox
        y)phosphazene))
IΤ
     25231-98-5DP, Hexachlorocyclotriphosphazene polymer, reaction products
     with (ferrocenylmethyl)methylamine followed by sodium trifluoroethòxide
     26085-02-9DP, Poly(dichlorophosphazene), reaction products with
     (ferrocenylmethyl) methylamine followed by sodium trifluoroethoxide
     57145-77-4DP, reaction products with poly(dichlorophosphazene) followed by
     sodium trifluoroethoxide
     RL: ARG (Analytical reagent use); SPN (Synthetic preparation); ANST
     (Analytical study); PREP (Preparation); USES (Uses)
        (preparation and potential use as glucose sensor)
     211051-09-1P, 2,2,4,4-Tetrachloro-6-(ferrocenylmethyl)(methyl)amino-6-(4-
ΙT
     (2-methyl-2-propenoyloxy)butoxy)cyclotriphosphazene
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (preparation and radical initiated copolymn. with Me methacrylate)
IΤ
     138020-35-6P, Sodium ferrocenylmethoxide
                                                211051-37-5P, Sodium
     1-ferrocenyl-2-propoxide
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (preparation and reaction with hexachlorocyclotriphosphazene)
     57145-77-4P, (Ferrocenylmethyl) (methyl) amine
IΤ
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (preparation and reactions with hexachlorocyclotriphosphazene and
        (methacryloylbutoxy)cyclophosphazene)
IT
     211051-12-6P, 2,2,4,4,6-Pentachloro-6-((ferrocenylmethyl)(methyl)amino)cyc
                        211051-14-8P, 2,2,4,4-Tetrachloro-6,6-
     lotriphosphazene
     bis((ferrocenylmethyl)(methyl)amino)cyclotriphosphazene
                                                                211051-17-1P,
     cis-2,2,4,6-Tetrachloro-4,6-bis((ferrocenylmethyl)(methyl)amino)cyclotriph
                211051-20-6P, trans-2, 2, 4, 6-Tetrachloro-4, 6-
     bis((ferrocenylmethyl)(methyl)amino)cyclotriphosphazene
                                                                211051-28-4P,
     trans-2, 4, 6-Trichloro-2, 4, 6-tris((ferrocenylmethyl)(methyl)amino)cyclotrip
     hosphazene
     RL: PEP (Physical, engineering or chemical process); PRP (Properties); RCT
     (Reactant); SPN (Synthetic preparation); PREP (Preparation); PROC
     (Process); RACT (Reactant or reagent)
        (preparation and reversible chemical/electrochem. oxidation of)
IT
     211051-23-9P, 2,2,4-Trichloro-4,6,6-tris((ferrocenylmethyl)(methyl)amino)c
```

```
yclotriphosphazene
     RL: PEP (Physical, engineering or chemical process); RCT (Reactant); SPN
     (Synthetic preparation); PREP (Preparation); PROC (Process); RACT
     (Reactant or reagent)
        (preparation and reversible chemical/electrochem. oxidation of)
IT
     211051-32-0P, Methyl methacrylate-2,2,4,4-tetrachloro-2-
     (ferrocenylmethyl) (methyl) amino-6-(4-(2-methyl-2-
     propenoyloxy) butoxy) cyclotriphosphazene copolymer
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (preparation and reversible electrochem. oxidation of)
ΙT
     211051-11-5P, 2,2,4,4,6-Pentachloro-6-(2-ferrocenyl-1-
     methylethoxy) cyclotriphosphazene
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (preparation of)
TΤ
     163130-17-4P, 2,2,4,4,6-Pentachloro-6-(4-(2-methyl-2-
     propenoyloxy) butoxy) cyclotriphosphazene
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (reaction with (ferrocenylmethyl)methylamine)
ΙT
     75-56-9P, preparation
                             12126-82-8P, 1-Ferrocenyl-2-propanol
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (reaction with ferrocenyllithium)
ΙT
     1271-15-4P, Ferrocenyllithium
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (reaction with propylene oxide)
ΙT
     940-71-6, Hexachlorocyclotriphosphazene
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reactions with sodium ferrocenylmethoxide and (ferrocenylmethyl)amine
        and ring-opening polymerization with subsequent reaction with
        (ferrocenylmethyl)amine)
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RE.CNT
        14
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     ANSWER 20 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
     1998:329124 CAPLUS
ΑN
DN
     129:28304
     Entered STN: 03 Jun 1998
ED
     Thermal analysis and characterization of 2-allylphenoxyorganocyclotriphosp.
ΤI
     hazene copolymers
ΑU
     Wu, Ho-Shing; Ke, Deng-Yuan
     Department of Chemical Engineering, Yuan-Ze University, Taoyuan, 32026,
CS
     Taiwan
     Journal of Polymer Research (1998), 5(2), 95-103
SO
     CODEN: JPOREP; ISSN: 1022-9760
PB
     Polymer Society
DΤ
     Journal
LA
     English
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CC
     35-7 (Chemistry of Synthetic High Polymers)
     Section cross-reference(s): 36
AΒ
     The radical copolymns. of 2-allylphenoxyorganocyclotriphosphazene with
     chloride, phenoxy, 2,2,2-trifluoroethoxy side group reacting with styrene,
    Me methacrylate, and vinylbenzyl chloride using AIBN, n-BuLi and UV as
     initiator were investigated. The type of the copolymn., reaction time,
     temperature were evaluated to obtain the optimum reaction condition. The
     incorporation of organophosphazene units into an organic polymer backbone
     decreased the glass transition temperature and increased the thermal stability
     of the copolymers. The weight conversion and the mol. weight had a maximum
value
     at reaction temperature of around 70 °C. The order for both weight
     conversion of the copolymn. and the thermal stability of phosphazene
     polymer with the co-monomer was VBC > STY > MMA, and with the side group
     Cl- > C6H5O- > CF3CH2O-. The phosphazene copolymer is of conductivity and
     crystallization
ST
     allylphenoxyorganocyclotriphosphazene photochem polymn vinyl monomer;
     styrene polymn allylphenoxyorganocyclotriphosphazene; methacrylate polymn
     allylphenoxyorganocyclotriphosphazene; vinylbenzyl chloride polymn
     allylphenoxyorganocyclotriphosphazene; thermal stability vinyl monomer
     allylphenoxyorganocyclotriphosphazene copolymer; glass transition vinyl
    monomer allylphenoxyorganocyclotriphosphazene copolymer; char yield vinyl
    monomer allylphenoxyorganocyclotriphosphazene copolymer; miscibility vinyl
    monomer allylphenoxyorganocyclotriphosphazene copolymer
IT
     Polymerization catalysts
        (AIBN, n-BuLi; for 2-allylphenoxyorganocyclotriphosphazene polymerization
with
        vinyl monomers)
ΙΤ
     Polyphosphazenes
     Polyphosphazenes
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (fluorine-containing; thermal anal. and characterization of
        2-allylphenoxyorganocyclotriphosphazene copolymers)
ΙT
    Miscibility
        (of 2-allylphenoxyorganocyclotriphosphazene copolymers)
IT
     Polymerization
        (photopolymn.; of 2-allylphenoxyorganocyclotriphosphazene polymerization
with
        vinyl monomers)
ΙT
     Fluoropolymers, preparation
     Fluoropolymers, preparation
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (polyphosphazene-; thermal anal. and characterization of
        2-allylphenoxyorganocyclotriphosphazene copolymers)
IT
     Glass transition
     Thermal stability
        (thermal anal. and characterization of 2-allylphenoxyorganocyclotriphos
        phazene copolymers)
ΙT
     Polyphosphazenes
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (thermal anal. and characterization of 2-allylphenoxyorganocyclotriphos
        phazene copolymers)
ΙT
                    109-72-8, Butyllithium, uses
     78-67-1, AIBN
     RL: CAT (Catalyst use); USES (Uses)
        (catalysts for 2-allylphenoxyorganocyclotriphosphazene polymerization with
        vinvl monomers)
     180090-89-5P 180090-90-8P 180090-91-9P 180090-92-0P
ΙT
                    180090-94-2P
     180090-93-1P
                                   181940-30-7P 181940-31-8P
     181940-32-9P
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (thermal anal. and characterization of 2-allylphenoxyorganocyclotriphos
        phazene copolymers).
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RE.CNT 15
RE
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     ANSWER 21 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
ΑN
     1998:52070 CAPLUS
DN
     128:61892
ED
     Entered STN: 29 Jan 1998
     Synthesis and Radical Polymerization of Methacrylate Monomers Containing
TΙ
     Cyclotriphosphazene. Thin-Layer Grafts of Their Polymers on a Poly(vinyl
     alcohol) Surface
ΑU
     Dez, Isabelle; Jaeger, Roger De
     Laboratoire de Spectrochimie Infrarouge et Raman, Universite des Sciences
CS
     et Technologies de Lille, Villeneuve d'Ascq, 59655, Fr.
SO
     Macromolecules (1997), 30(26), 8262-8269
     CODEN: MAMOBX; ISSN: 0024-9297
PB
     American Chemical Society
DT
     Journal
LA
     English
CC
     35-8 (Chemistry of Synthetic High Polymers)
AB
     The reaction between (4-hydroxyphenoxy)pentakis(4-
     ethylphenoxy) cyclotriphosphazene obtained by the reaction of the
     (4-methoxyphenoxy)pentakis(4-ethylphenoxy)cyclotriphosphazene with
     BBr3/H2O-and methacryloyl chloride led to (4-methacryloyloxy)pentakis(4-
     ethylphenoxy)cyclotriphosphazene (I). Radical polymerization of I led to the
     hybrid inorg.-organic poly(cyclophosphazene-methacrylate) (II). The
     structures of both classes of compds. were investigated by 31P, 13C, and
     1H NMR, IR spectroscopy, mass spectrometry, size exclusion chromatog., and
     elemental anal. The thermal behavior of the new polymer II was
     investigated and compared to the one of poly(Ph methacrylate). It was
     demonstrated that the incorporation of cyclophosphazene as pendant group
     increases the thermal stability of the corresponding organic polymer.
     free-radical promoted grafting of II onto poly(vinyl alc.) surface was
     described. The resultant surfaces were studied by ATR-IR, XPS, and
     contact angle techniques. It was demonstrated that the presence of Et
     functions in II increases the grafting yield.
     radical polymn methacrylate contg cyclotriphosphazene; grafting polyvinyl
     alc cyclotriphosphazene contq methacrylate
IT
     Contact angle
        (of water on cyclotriphosphazene group-containing polymethacrylate)
     123-07-9, 4-Ethylphenol 150-76-5, 4-Methoxyphenol
ΙT
                             940-71-6, Hexachlorotriphosphazene
     Methacryloyl chloride
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (in preparation of cyclotriphosphazene group-containing methacrylate
monomer)
     200334-86-7P, Chloropenta (4-ethylphenoxy) cyclotriphosphazene
     200334-87-8P, (4-Methoxyphenoxy)pentakis(4-ethylphenoxy)cyclotriphosphazen
         200334-88-9P, (4-Hydroxyphenoxy)pentakis(4-
     ethylphenoxy) cyclotriphosphazene
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (in preparation of cyclotriphosphazene group-containing methacrylate
monomer)
     200334-90-3P, (4-Methacryloyloxy)pentakis(4-
```

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ethylphenoxy)cyclotriphosphazene homopolymer 200334-91-4P,
     (4-Methacryloyloxy)pentakis(4-ethylphenoxy)cyclotriphosphazene-vinyl
     alcohol graft copolymer
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (preparation and characterization of)
TT
     200334-89-0P, (4-Methacryloyloxy)pentakis(4-
     ethylphenoxy)cyclotriphosphazene
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (preparation and polymerization of)
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L9
     ANSWER 22 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
     1997:430254 CAPLUS
DN
     127:136103
ΕĎ
     Entered STN: 11 Jul 1997
     Copolymerization of 2-(4'-vinyl-4-biphenylyloxy)pentachlorocyclotriphospha
TI
     zene with acrylate and methacrylate monomers
     Selvaraj, I. Immanuel; Chandrasekhar, V.
ΑU
CS
     Dep. Chemistry, Indian Inst. Technology, Kanpur, 208 016, India
SO
     Polymer (1997), 38(14), 3617-3623
     CODEN: POLMAG; ISSN: 0032-3861
PB
     Elsevier
DT
     Journal
LA
     English
CC
     35-4 (Chemistry of Synthetic High Polymers)
     A cyclophosphazene substituted organic monomer, 2-(4'-vinyl-4-biphenylyloxy)
AB
     pentachlorocyclotriphosphazene (CPHVB), has been copolymd. with Me
     acrylate (MA), Et acrylate (EA) and Me methacrylate (MMA) using free
     radical methods. Flame retardant thermally stable copolymers with up to a
     maximum of 90% phosphazene content were obtained. Reactivity ratios and
     Alfrey-Price parameters suggest that the biphenylyloxy unit successfully
```

insulates the vinyl moiety from the σ -electron withdrawing effect of

the cyclophosphazene ring.

```
ST
     vinyl biphenylyloxy pentachlorocyclotriphosphazene acrylate polymn;
    methacrylate vinyl biphenylyloxy pentachlorocyclotriphosphazene polymn;
     reactivity vinyl biphenylyloxy pentachlorocyclotriphosphazene acrylate
     polymn
ΙT
     Q-e value in polymerization
        (copolymn. of 2-(4'-vinyl-4-biphenylyloxy)pentachlorocyclotriphosphazen
        e with acrylate and methacrylate monomers)
ΙT
     Glass transition temperature
        (properties of 2-(4'-vinyl-4-biphenylyloxy)pentachlorocyclotriphosphaze
        ne-acrylate or methacrylate copolymers)
IT
     Polymerization
     Reactivity ratio in polymerization
        (radical; copolymn. of 2-(4'-vinyl-4-biphenylyloxy)pentachlorocyclotrip
        hosphazene with acrylate and methacrylate monomers)
ΙT
     122847-49-8P, Methyl methacrylate-2-(4'-vinyl-4-
    biphenylyloxy)pentachlorocyclotriphosphazene copolymer
     192944-34-6P, Methyl acrylate-2-(4'-vinyl-4-
    biphenylyloxy)pentachlorocyclotriphosphazene copolymer
     192944-35-7P, Ethyl acrylate-2-(4'-vinyl-4-
     biphenylyloxy)pentachlorocyclotriphosphazene copolymer
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (copolymn. of 2-(4'-vinyl-4-biphenylyloxy)pentachlorocyclotriphosphazen
        e with acrylate and methacrylate monomers)
     80-62-6, Methyl methacrylate 96-33-3, Methyl acrylate
                                                               140-88-5, Ethyl
     acrylate
                115529-67-4, 2-(4'-Vinyl-4-biphenylyloxy)pentachlorocyclotripho
     sphazene
     RL: PRP (Properties); RCT (Reactant); RACT (Reactant or reagent)
        (reactivity ratio of copolymn. of 2-(4'-vinyl-4-
       biphenylyloxy)pentachlorocyclotriphosphazene with acrylate and
       methacrylate monomers)
     ANSWER 23 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
     1997:82428 CAPLUS
AN
     126:186632
DN
     Entered STN: 05 Feb 1997
ED
     Molecular characterization of copolymer of 2-allylphenoxyorganocyclotripho
     sphazene with styrene, methyl methacrylate and vinylbenzyl chloride
ΑU
     Wu, Ho-Shing; Ke, Deng-Yuan
     Dep. Chem. Eng., Yuan-Ze Inst. Tech., Taichung, Taiwan
CS
     Polymer (1997), 38(2), 431-436
SO
     CODEN: POLMAG; ISSN: 0032-3861
PB
     Elsevier
DT
     Journal
LA
     English
     36-5 (Physical Properties of Synthetic High Polymers)
CC
     Monomers of 2-allylphenoxy-R-cyclotriphosphazene [R = pentachloro,
AB
     pentaphenoxy, and 2,2,2-trifluoroethoxy] were synthesized by reacting
     hexachlorocyclotriphosphazene with 2-allylphenol, phenol and
     2,2,2-trifluoroethanol, resp. The copolymn. of 2-allylphenoxy-R-
     cyclotriphosphazenes with styrene, Me methacrylate or vinylbenzyl chloride
     was carried out by phase-transfer catalysis. Nine copolymers were
     identified and characterized by light scattering, viscometry and 31P NMR
     spectroscopy. The Huggins constant of the copolymers was -0.11 to 0.52 in
     DMAC solution, smaller than the values expected for linear and flexible
     polymers in a good solvent. Micelles of the copolymers behave as
     impermeable spheres, as determined from hydrodynamics and thermodn.
     observations. The structure of the micellar shell was determined The
     translational diffusion coefficient, Mark-Houwink constant [0.53], hydrodynamic
     radius, micellar weight, aggregation number, characteristic ratio, and
intrinsic
     viscosity of the copolymers were obtained. The chains of the copolymers
     have a high degree of stiffness and a lack of rotational freedom in solution,
     as indicated by the high values of the ratio of length and mol. weight of
     repeating unit.
```

allylphenoxycyclotriphosphazene copolymer prepn mol parameter; micelle

```
Huggins const radius polyphosphazene copolymer
ΙT
    Viscosity
        (Mark-Houwink constant; preparation and mol. and aggregation properties of
        copolymers of allylphenoxycyclotriphosphazenes)
ΙT
     Polymer chains
        (dynamics, stiffness; preparation and mol. and aggregation properties of
        copolymers of allylphenoxycyclotriphosphazenes)
ΙT
    Micelles
    Molecular association
     Radius of gyration
        (preparation and mol. and aggregation properties of copolymers of
        allylphenoxycyclotriphosphazenes)
ΙT
     Polyphosphazenes
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (preparation and mol. and aggregation properties of copolymers of
        allylphenoxycyclotriphosphazenes)
     137856-38-3P, 2-Allylphenoxypentachlorocyclotriphosphazene
                                                                  137856-40-7P,
TΨ
     2-Allylphenoxypentaphenoxycyclotriphosphazene
                                                     180090-88-4P.
     2-Allylphenoxypenta(2,2,2-trifluoroethoxy)cyclotriphosphazene
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (monomer; preparation of allylphenoxycyclotriphosphazene monomers and
        phase-transfer catalysis polymerization)
ΙT
     180090-89-5P, 2-Allylphenoxypentaphenoxycyclotriphosphazene-styrene
     copolymer 180090-90-8P, 2-Allylphenoxypentaphenoxycyclotriphosph
     azene-methyl methacrylate copolymer
                                          180090-91-9P, 2-
     Allylphenoxypentaphenoxycyclotriphosphazene-vinylbenzyl chloride copolymer
     180090-92-0P, 2-Allylphenoxypenta(2,2,2-trifluoroethoxy)cyclotriphosphazen
     e-styrene copolymer 180090-93-1P, 2-Allylphenoxypenta(2,2,2-
     trifluoroethoxy)cyclotriphosphazene-methyl methacrylate copolymer
     180090-94-2P, 2-Allylphenoxypenta(2,2,2-trifluoroethoxy)cyclotriphosphazen
     e-vinylbenzyl chloride copolymer 181940-30-7P, 2-
     Allylphenoxypentachlorocyclotriphosphazene-styrene copolymer
     181940-31-8P, 2-Allylphenoxypentachlorocyclotriphosphazene-methyl
     methacrylate copolymer
                             181940-32-9P, 2-Allylphenoxypentachlorocyclotriph
     osphazene-vinylbenzyl chloride copolymer
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (preparation and mol. and aggregation properties of copolymers of
        allylphenoxycyclotriphosphazenes)
IT
     75-89-8, 2,2,2-Trifluoroethanol
                                       108-95-2, Phenol, reactions
                                                                      940-71-6,
                                     1745-81-9, 2-Allylphenol
     Hexachlorocyclotriphosphazene
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (preparation of allylphenoxycyclotriphosphazene monomers and phase-transfer
        catalysis polymerization)
     ANSWER 24 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
     1996:625334 CAPLUS
AN
DN
     125:248578
     Entered STN: 23 Oct 1996
ED
     Kinetics and Characterization of Copolymerization of (2-
     Allylphenoxy)pentachlorocyclotriphosphazene with Styrene, Methyl
     Methacrylate, and Vinylbenzyl Chloride
     Wu, Ho-Shing; Ke, Deng-Yuan
ΑU
     Department of Chemical Engineering, Yuan-Ze Institute of Technology,
CS
     Taoyuan, 135, Taiwan
     Industrial & Engineering Chemistry Research (1996), 35(11), 4291-4295
SO
     CODEN: IECRED; ISSN: 0888-5885
     American Chemical Society
PB
DT
     Journal
LA
     English
     35-3 (Chemistry of Synthetic High Polymers)
CC
     Section cross-reference(s): 36, 37
     The monomer (2-allylphenoxy)pentachlorocyclotriphosphazene (Ia) was
AΒ
     synthesized on reacting hexachlorocyclotriphosphazene with 2-allylphenol
```

shell .structure allylphenoxycyclotriphosphazene copolymer; aggregation

with a phase-transfer catalyst in a dichloromethane/alkaline solution The copolymns. of Ia with styrene, Me methacrylate, and vinylbenzyl chloride were investigated. No improvement in the weight conversion of the copolymn. reaction was observed when the duration of copolymn. exceeded 30 h and the reaction temperature was controlled between 60 and 80 °C. Mol. wts. of the copolymers decreased with increasing molar fraction of Ia in the feed, but there was no correlation with the content of Ia in the copolymer. Reactivity ratios and Alfrey-Price parameters for the copolymers were obtained. The copolymers were characterized by using EA, GPC, TGA, and DSC. The incorporation of the phosphazene unit affected the flame-retarding properties of the copolymer. Tensile measurement showed that the copolymer of Ia reacted with Me methacrylate was hard and reactivity ratio allylphenoxypentachlorocyclotriphosphazene polymn vinyl monomer; Alfrey Price parameter allylphenoxypentachlorocyclotriphosphazene polymn vinyl; vinylbenzyl chloride copolymer allylphenoxypentachlorocyclot riphosphazene; methyl methacrylate copolymer allylphenoxypentachlorocyclot riphosphazene; styrene copolymer allylphenoxypentachlorocyclotriphosphazen e; tensile property allylphenoxypentachlorocyclotriphosphazene copolymer; flame retardant allylphenoxypentachlorocyclotri phosphazene copolymer Phosphazene polymers RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (characterization of (2-allylphenoxy)pentachlorocyclotriphosphaze polymers with styrene, Me methacrylate, and vinylbenzyl chloride) Fire-resistant materials (characterization of (2-allylphenoxy)pentachlorocyclotriphosphaze polymerization with styrene, Me methacrylate, and vinylbenzyl chloride) Glass temperature and transition (of (2-allylphenoxy)pentachlorocyclotriphosphaze copolymers with styrene, Me methacrylate, and vinylbenzyl chloride) Q-e value in polymerization Reactivity ratio in polymerization (of. (2-allylphenoxy)pentachlorocyclotriphosphaze with styrene, Me methacrylate, and vinylbenzyl chloride) 80-62-6, Methyl Methacrylate 100-42-5, Styrene, reactions 30030-25-2 RL: PRP (Properties); RCT (Reactant); RACT (Reactant or reagent) (kinetics and product characterization of (2allylphenoxy)pentachlorocyclotriphosphaze polymerization with styrene, Me methacrylate, and vinylbenzyl chloride) 137856-38-3P, (2-Allylphenoxy)pentachlorocyclotriphosphazene RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent) (kinetics and product characterization of (2allylphenoxy)pentachlorocyclotriphosphaze polymerization with styrene, Me methacrylate, and vinylbenzyl chloride) 181940-30-7P, (2-Allylphenoxy)pentachlorocyclotriphosphazene-styrene copolymer 181940-31-8P, (2-Allylphenoxy)pentachlorocyclotriphosp hazene-methyl methacrylate copolymer 181940-32-9P RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (kinetics and product characterization of (2allylphenoxy)pentachlorocyclotriphosphaze polymerization with styrene, Me methacrylate, and vinylbenzyl chloride) 940-71-6, Hexachlorocyclotriphosphazene 1745-81-9, 2-Allylphenol RL: RCT (Reactant); RACT (Reactant or reagent) (monomer synthesis; kinetics and product characterization of (2-allylphenoxy)pentachlorocyclotriphosphaze polymerization with styrene, Me methacrylate, and vinylbenzyl chloride) ANSWER 25 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN

L9

1996:551147 CAPLUS AN

DN 125:171007

ΙT

IT

ΙT

ΙT

ΙT

ΙT

IT

ΙT

Entered STN: 17 Sep 1996 ΕD

Weather-resistant (meth)acrylic coating compositions ΤI

IN Tada, Juji; Akata, Atsuo

PΑ Otsuka Kagaku Kk, Japan SO Jpn. Kokai Tokkyo Koho, 19 pp.

CODEN: JKXXAF

DT Patent LA Japanese

IC ICM C08F220-10

ICS C08F220-30; C08F220-36; C08F290-04; C09D004-02

CC 42-7 (Coatings, Inks, and Related Products)

FAN.CNT 1

| PATENT NO. | KIND | DATE | APPLICATION NO. | DATE |
|--|----------|----------------------|--|----------|
| | | | | |
| PI JP 08151415
PRAI JP 1994-234707
CLASS | A2
A1 | 19960611
19940929 | JP 1995-252476 | 19950929 |
| PATENT NO. CLASS | PATENT | FAMILY CLASS | IFICATION CODES | |
| JP 08151415 ICM
ICS
IPCI | | 0-30; C08F22C | -36; C08F290-04; C09; C08F0220-30 [ICS,6 | |

[ICS, 6]; C08F0290-04 [ICS, 6]; C09D0004-02 [ICS, 6]

OS MARPAT 125:171007

GI

AB A weather-resistant composition suitable for surface protective coating of various molded materials comprises (1) (meth)acryloyloxy group-containing (meth)acrylic compds., which can be either oligomers or monomeric compds., and (2) UV-absorbing agents selected from I and II (X = H, halogen; R1 = H, C1-6 alkyl; R2 = C2-12 alkylene; R3 = H, Me; R4 = C2-6 alkylene which may be substituted by OH; Y = H, C1-6 alkoxy; Z = H, OH). A coating composition contained 40 parts of Aronix M 8030 (polyester acrylate), 20 parts tetrahydrofurfuryl acrylate, 40 parts trimethylolpropane triacrylate, 3 parts 2-(2'-hydroxy-5'-acryloyloxyethylphenyl)-2H-benzotriazole, and 3 parts of a polymerization initiator. The composition was applied on a polycarbonate

plate and irradiated with a high-pressure mercury lamp to form a coating with pencil hardness 2H and showing good results in a weather exposure test.

ST methacrylic coating weather resistance; UV absorbent methacrylic benzotriazole

IT Acrylic polymers, uses

```
RL: POF (Polymer in formulation); PRP (Properties); TEM (Technical or
     engineered material use); USES (Uses)
        (weather-resistant (meth)acrylic coating compns.)
IT
     Light stabilizers
        (UV, weather-resistant (meth)acrylic coating compns. containing reactive UV
        absorbents)
IT
     Coating materials
        (weather-resistant, weather-resistant (meth)acrylic coating compns.)
IT
     2399-48-6, Tetrahydrofurfuryl acrylate
                                              3524-66-1, Pentaerythritol
     trimethacrylate
                       3524-68-3, Pentaerythritol triacrylate
                                                                4687-94-9,
     Ripoxy SP 1509
                      7328-17-8
                                  13048-33-4
                                               15625-89-5, Trimethylolpropane
     triacrylate
                   16432-81-8
                               29570-58-9, Dipentaerythritol hexaacrylate
     61287-25-0, Aronix M 8030 92832-53-6
                                            95523-89-0, Viscoat 3700
     96478-09-0
                 108251-12-3, Aronix M 1200
                                               130504-36-8
                                                             149531-40-8, Art
     resin UN 3320HA
                       170103-27-2
                                     180782-04-1
                                                   180782-06-3
                                                                 180782-11-0
     180782-14-3
                   180782-16-5
                                 180782-19-8
                                               180782-21-2
                                                             180782-23-4
     180782-26-7
                   180782-29-0
                                 180782-30-3
                                               180782-32-5
                                                             180782-34-7
     180782-36-9
                   180782-38-1
                                 180782-41-6
                                               180782-43-8
                                                             180782-45-0
                                 180782-52-9
                                               180782-55-2
                                                             180782-58-5
     180782-47-2
                   180782-49-4
     180782-61-0
                   180782-64-3
                                 180782-66-5
                                               180782-68-7
                                                             180782-70-1
                   180782-76-7
                                 180782-79-0
     180782-73-4
                                               180782-82-5
                                                             180782-85-8
     180782-88-1
     RL: POF (Polymer in formulation); PRP (Properties); TEM (Technical or
     engineered material use); USES (Uses)
        (weather-resistant (meth)acrylic coating compns.)
L9~
     ANSWER 26 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
     1996:467605 CAPLUS
AN
DN
     125:143448
ED
     Entered STN: 08 Aug 1996
ΤI
     Synthesis, characterization, and modification of 2-
     allylphenoxyorganocyclotriphosphazene copolymers
ΑU
     Wu, Ho-Shing; Ke, Deng-Yuan
     Dep. Chem. Engineering, Yuan-Ze Institute Technology, Taichung, Peop. Rep.
CS
SO
     Journal of Applied Polymer Science (1996), 61(8), 1351-1358
     CODEN: JAPNAB; ISSN: 0021-8995
PΒ
     Wiley
DT
     Journal
LA
     English
     35-6 (Chemistry of Synthetic High Polymers)
CC
     The monomer 2-allylphenoxyorganocyclotriphosphazene (I) with phenoxy (or
AB
     2,2,2-trifluoroethoxy) side group was synthesized on reacting
     hexachlorocyclotriphosphazene with phenol (or 2,2,2-trifluoroethanol) and
     2-allylphenol with phase-transfer catalysis in a dichloromethane/alkaline
     solution The synthesis had a large yield (>80%) and narrow product
     distribution and proceeded under mild conditions. The radical copolymn.
     of I with styrene, Me methacrylate, or vinylbenzyl chloride using AIBN as
     an initiator was investigated. Reactivity ratios and Alfrey-Price
     parameters for copolymers were obtained. Organophosphazene monomer I was
     less reactive than the other one. The thermal stability of the copolymer
     with phenoxy side group was greater than that with 2,2,2-trifluoroethoxy
     side group. The incorporation of organophosphazene units into an organic
     polymer backbone decreased Tg but increased the thermal stability of the
     copolymers. Young's modulus tended to decrease and the polymer became
     softer and ductile when the incorporated amount of I increases.
     reactivity ratio polymn allylphenoxyorganocyclotriphosphazene vinyl
     monomer; modulus thermal stability allylphenoxyorganocyclotriphosphazene
     copolymer; allylphenoxyorganocyclotriphosphazene polymn styrene
     methacrylate vinylbenzyl chloride
ΙT
     Polymerization catalysts
        (synthesis of 2-allylphenoxypentaphenoxycyclotriphosphazene copolymers
        with AIBN or BuLi catalysts)
ΙT
     Chains, chemical
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Glass temperature and transition

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Polymerization
    Q-e value in polymerization
    Reactivity ratio in polymerization
       (synthesis, characterization, and modification of 2-
       allylphenoxypentaphenoxycyclotriphosphazene copolymers)
    Phosphazene polymers
ΙT
    RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
       (synthesis, characterization, and modification of 2-
       allylphenoxypentaphenoxycyclotriphosphazene copolymers)
    78-67-1, AIBN 109-72-8, n-Butyllithium, uses
ΙT
    RL: CAT (Catalyst use); USES (Uses)
       (synthesis of 2-allylphenoxypentaphenoxycyclotriphosphazene copolymers
       with AIBN or BuLi catalysts)
                  180090-88-4P
ΙT
    137856-40-7P
    RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP
    (Preparation); RACT (Reactant or reagent)
       (synthesis, characterization, and modification of 2-
       allylphenoxypentaphenoxycyclotriphosphazene copolymers)
ΙT
    180090-89-5P 180090-90-8P 180090-91-9P
                                             180090-92-0P
    180090-93-1P
                  180090-94-2P
    RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
       (synthesis, characterization, and modification of 2-
       allylphenoxypentaphenoxycyclotriphosphazene copolymers)
ΙT
    75-89-8 80-62-6 100-42-5, reactions 108-95-2, Phenol, reactions
    940-71-6, Hexachlorocyclotriphosphazene 1745-81-9, 2-Allylphenol
    30030-25-2
    RL: RCT (Reactant); RACT (Reactant or reagent)
       (synthesis, characterization, and modification of 2-
       allylphenoxypentaphenoxycyclotriphosphazene copolymers)
ΙT
    5032-39-3P, Chloropentaphenoxycyclotriphosphazene
    RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
       (synthesis, characterization, and modification of 2-
       allylphenoxypentaphenoxycyclotriphosphazene copolymers)
L9
    ANSWER 27 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
    1995:905909 CAPLUS
ΑN
DN
    124:31116
ED
    Entered STN: 09 Nov 1995
    Temperature-sensitive hydrogels and their hydrogels for control of drug
TΙ
IN
    Ikeda, Koichi; Tamura, Kazumi; Murata, Kazuyuki; Okano, Mitsuo; Sakurai,
    Yasuhisa
PA
    Nippon Kayaku Kk, Japan
    Jpn. Kokai Tokkyo Koho, 12 pp.
    CODEN: JKXXAF
DT
    Patent
LA
    Japanese
IC
    ICM C08F220-56
    ICS A61K009-00; A61K009-70; C08F220-58
ICI
    C08F220-56, C08F230-08; C08F220-58, C08F230-08
    37-3 (Plastics Manufacture and Processing)
    Section cross-reference(s): 63
FAN.CNT 1
                             DATE APPLICATION NO.
    PATENT NO.
                       KIND
                                                              DATE
                      ____
                                         _____
    -----
    JP 07228639
                       A2 19950829
                                       JP 1994-43361
                                                               19940218
                       B2
    JP 3541894
                             20040714
PRAI JP 1994-43361
                             19940218
CLASS
              CLASS PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
 ----
               ____
 JP 07228639
               ICM
                      C08F220-56
                ICS
                      A61K009-00; A61K009-70; C08F220-58
                ICI
                      C08F220-56, C08F230-08; C08F220-58, C08F230-08
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[ICS, 6]; C08F0220-58 [ICS, 6]; C08F0220-56 [ICI, 6];
                        C08F0230-08 [ICI,6]; C08F0220-58 [ICI,6]; C08F0230-08
                        [ICI, 6]
                 ECLA
                        C08F220/54
AB
     Title hydrogels comprise composites having phase-separated structures of
     crosslinked temperature-sensitive polymers covalently bonded with inorg.
     polymers. Thus, N-isopropylacrylamide 5, N-vinyl-2-pyrrolidone 0.263, and
     eicosamethyl-1,19-divinyldecasiloxane 1.12 g were polymerized in 1,4-dioxane
     containing tert-Bu peroxy-2-ethylhexanoate to give a hydrogel showing phase
     transition temperature 36.4° and tensile strength 19.5 kg/cm2.
ST
     hydrogel temp sensitive drug delivery; acrylic siloxane hydrogel temp
     sensitive; eicosamethyldivinyldecasiloxane copolymer hydrogel temp
     sensitive; vinylpyrrolidone copolymer hydrogel temp sensitive;
     isopropylacrylamide copolymer hydrogel temp sensitive
ΙT
     Pharmaceutical dosage forms
        (phase-separated inorg. polymer-organic polymer composite
temperature-sensitive
        hydrogels useful for control of drug release)
     Phosphazene polymers
     Siloxanes and Silicones, uses
     RL: TEM (Technical or engineered material use); THU (Therapeutic use);
     BIOL (Biological study); USES (Uses)
        (acrylic, phase-separated inorg. polymer-organic polymer composite
        temperature-sensitive hydrogels useful for control of drug release)
ΙT
     171729-40-1P, Eicosamethyl-1,19-divinyldecasiloxane-N-isopropylacrylamide-
     N-vinyl-2-pyrrolidone copolymer
                                       171729-41-2P, Ethylene glycol
     dimethacrylate-N-isopropylacrylamide-3-methacryloxypropyltrimethoxysilane-
     N-vinyl-2-pyrrolidone copolymer
                                      171729-42-3P, Eicosamethyl-1,19-
     divinyldecasiloxane-N-isopropylacrylamide-3-methacryloxypropyltrimethoxysi
     lane-N-vinyl-2-pyrrolidone copolymer
                                            171729-44-5P, 1,3-Bis(3-
     methacryloxypropyl)-1,1,3,3-tetramethyldisiloxane-N-isopropylacrylamide-N-
     vinyl-2-pyrrolidone copolymer 171729-46-7P, Hexakis(2-
     methacryloxyethyl)cyclotriphosphazene-N-isopropylacrylamide-N-vinyl-2-
     pyrrolidone copolymer
     RL: IMF (Industrial manufacture); TEM (Technical or engineered material
     use); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation);
     USES (Uses)
        (phase-separated inorg. polymer-organic polymer composite
temperature-sensitive
        hydrogels useful for control of drug release)
L9
     ANSWER 28 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
AN
     1995:476651 CAPLUS
DN
     122:313980
ΕD
     Entered STN: 08 Apr 1995
     Reaction pathways in the thermolysis of (2-oxyethylmethacryl)pentachlorocy
     clotriphosphazene
ΑU
     Hayes, Robert F.; Allen, Christopher W.
     Dep. Chem., Univ. Vermont, Burlington, VT, 05405-0125, USA
CS
SO
     Phosphorus, Sulfur and Silicon and the Related Elements (1994),
     93-94(1-4), 269-72
     CODEN: PSSLEC; ISSN: 1042-6507
PB
     Gordon & Breach
DT
     Journal
LA
     English
CC
     22-6 (Physical Organic Chemistry)
GΙ
```

C08F0220-56 [ICM, 6]; A61K0009-00 [ICS, 6]; A61K0009-70

IPCI

The (2-oxyethyl methacryl) pentachlorocyclotriphosphazene, N3P3Cl5OCH2CH2OC(O)C(Me)=CH2 (I), has been shown to undergo a slow rearrangement to the corresponding phosphazane (II). Monitoring of the rate of rearrangement at 35° to 111° shows a first order process in I with a pos. enthalpy and neg. entropy of activation. A mechanism based on tightly bound ion pairs and transfer of the CH2CH2C(O)C(CH3)=CH2 moiety to the nitrogen center is proposed. A hydrolysis product, N3P3Cl5OH and significant amts. of the unique oxo-bridged dimer, N3P3Cl5OP(O)NCH2CH2OC(O)C(Me)=CH2(NPCl2)2 (III) have also been observed The identity of III was established by 31P COSY expts. and mass spectrometry. The formation of these products can be related to the proposed mechanistic pathway. The rearrangement in related oxyalkylalkylphosphazenes has been explored.

ST rearrangement oxyethylmethacrylpentachlorocyclotriphosphazene kinetics mechanism

IT Kinetics of rearrangement Rearrangement

(of (2-oxyethylmethacryl)pentachlorocyclotriphosphazene)

IT 163130-14-1 163130-17-4

RL: RCT (Reactant); RACT (Reactant or reagent)

(attempted. rearrangement of)

IT 125553-74-4, (2-Oxyethylmethacryl)pentachlorocyclotriphosphazene
RL: PEP (Physical, engineering or chemical process); PRP (Properties); RCT
(Reactant); PROC (Process); RACT (Reactant or reagent)
 (kinetics and mechanism of rearrangement of (2-

oxyethylmethacryl)pentachlorocyclotriphosphazene)

IT 163130-15-2P 163130-16-3P

RL: SPN (Synthetic preparation); PREP (Preparation) (kinetics and mechanism of rearrangement of (2oxyethylmethacryl)pentachlorocyclotriphosphazene)

IT 92832-53-6 163130-18-5 163130-19-6

RL: RCT (Reactant); RACT (Reactant or reagent)
 (rearrangement of)

- L9 ANSWER 29 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
- AN 1994:667930 CAPLUS
- DN 121:267930
- ED Entered STN: 26 Nov 1994
- TI holographic recording material

```
ΙN
     Tani, Motoaki
PΑ
     Fujitsu Ltd, Japan
SO
     Jpn. Kokai Tokkyo Koho, 4 pp.
     CODEN: JKXXAF
     Patent
DΤ
     Japanese
LA
     ICM G03H001-02
·IC
     ICS G03F007-004; G03F007-028; G03F007-032
     74-8 (Radiation Chemistry, Photochemistry, and Photographic and Other
CC
     Reprographic Processes)
FAN.CNT 1
     PATENT NO.
                      KIND DATE
                                        APPLICATION NO.
                                                               DATE
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                                          -----
                                                                _____
    JP 06035389
                       A2
                              19940210 JP 1992-192432
                                                                19920720
PRAI JP 1992-192432
                             19920720
CLASS
 PATENT NO.
              CLASS PATENT FAMILY CLASSIFICATION CODES
                ----
 _____
 JP 06035389
                ICM
                       G03H001-02
                ICS
                       G03F007-004; G03F007-028; G03F007-032
                IPCI
                       G03H0001-02 [ICM,5]; G03F0007-004 [ICS,5]; G03F0007-028
                       [ICS, 5]; G03F0007-032 [ICS, 5]
AΒ
     A highly sensitive holog, recording material comprises a binder resin
     containing a carbazole ring in the repeating unit, an acryl and a phosphaszene
     monomer having a refractive index lower than that of the binder resin, a
     sensitizer, and a biimidazole photoinitiator.
ST
     photopolymerizable holog recording material phosphazene; acryl
     photopolymerizable holog recording material; carbazole binder
     photopolymerizable holog material
IΤ
     Holography
        (photopolymerizable compns. containing carbazole resins and acryl and
        phosphazene monomers for)
     25067-59-8, Poly(N-vinylcarbazole)
TΤ
     RL: USES (Uses)
        (photopolymerizable compns. containing acryl and phosphazene monomers and,
        for holog.)
IT
     7189-82-4, 2,2'-Bis(o-chlorophenyl)-4,4',5,5'-tetraphenyl-1,2'-biimidazole
     38215-36-0, 3-(2'-Benzothiazoly1)-7-N, N-diethylaminocoumarin 63226-13-1
     79579-93-4, 4-Butoxyphenyl-2,6-diphenylthiopyrylium perchlorate
     96273-26-6, 4-tert-Butyl-2,6-bis(4'-dimethylaminobenzylidene)cyclohexanone
     RL: USES (Uses)
        (photopolymerizable compns. containing carbazole resins and acryl and
        phosphazene monomers and, for holog.)
     15625-89-5, Trimethylolpropanetriacrylate 56093-53-9, Pentaerythritol
IT
     acrylate
     RL: USES (Uses)
        (photopolymerizable compns. containing carbazole resins and phosphazene
        monomers and, for holog.)
ΙT
     92832-53-6, Hexakis (methacryloyloxyethoxy) cyclotriphospazene
     RL: USES (Uses)
        (photopolymerizable compns. containing carbazole resins and, for holog.)
L9
     ANSWER 30 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
AN
     1994:641701 CAPLUS
DN
     121:241701
ED
     Entered STN: 12 Nov 1994
TI
     electrophotog. photoreceptor
ΙN
     Nagahara, Susumu; Maruyama, Akio
PA
     Canon Kk, Japan
SO
     Jpn. Kokai Tokkyo Koho, 9 pp.
     CODEN: JKXXAF
DT
     Patent
LA
     Japanese
IC
     ICM G03G005-147
     ICS G03G005-147; G03G005-05; H04N001-29
```

CC 74-3 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

FAN.CNT 1

JP 06019176

| | PATENT NO. | KIND | DATE | APPLICATION NO. | DATE |
|------------|-------------------------------|------|----------------------|-----------------|----------|
| | | | | | |
| PI
PRAI | JP 06019176
JP 1992-192751 | A2 | 19940128
19920629 | JP 1992-192751 | 19920629 |

PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES

ICM G03G005-147 ICS G03G005-147; G03G005-05; H04N001-29

IPCI G03G0005-147 [ICM, 5]; G03G0005-147 [ICS, 5]; G03G0005-05

[ICS, 5]; H04N0001-29 [ICS, 5]

GI

$$\begin{array}{c|c}
R^{5} O & OR^{5} \\
R^{5} O & P \\
 & || & | OR^{5} \\
 & || & | OR^{5} \\
R^{5} O & P \\
 & || & | OR^{5}
\end{array}$$

AB An electrophotog. photoreceptor showing reduced contamination and improved surface sliding and cleaning properties comprises, on an electroconductive support, a photosensitive layer and a protective layer comprising dispersed metal oxide particles surface-treated with a compound represented by the formula CH2=CR1CO2XSi(OR2)(OR3)OR4 (R1 = H or methyl; R2-4 = H or C1-4 alkyl; X = C1-6 alkylene) and a resin prepared by polymerization of a compound

represented by the formula I (R6 = ZOCOCR7=CH2; R7 = H or methyl; Z = alkylene or arylene).

ST electrophotog photoreceptor protective layer metal oxide

IT Electrophotographic photoconductors and photoreceptors

(protective layers containing resin-coated metal oxide particles for)

IT 2530-85-0, γ -Methacryloxypropyltrimethoxysilane

RL: USES (Uses)

(antimony-containing tin oxide particles coated with, for protective layers for electrophotog. photoreceptors)

IT 1332-29-2, Tin oxide

RL: USES (Uses)

(antimony-containing, particles, coated with resins for protective layers for electrophotog. photoreceptors)

IT 92832-53-6 147075-51-2

RL: USES (Uses)

(protective layers containing resin-coated metal oxide particles and, for electrophotog. photoreceptors)

IT 7440-36-0, Antimony, uses

RL: USES (Uses)

(tin oxide particles containing, coated with resins for protective layers for electrophotog. photoreceptors)

- L9 ANSWER 31 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
- AN 1994:485900 CAPLUS
- DN 121:85900
- ED Entered STN: 20 Aug 1994
- TI Phosphazene-type polymer-coated parts with weatherability and abrasion resistance
- IN Mori, Shigeo; Ando, Hiroyuki
- PA Idemitsu Petrochemical Co, Japan
- SO Jpn. Kokai Tokkyo Koho, 22 pp. CODEN: JKXXAF

```
DT
    Patent
LA
    Japanese
IC
    ICM C08J007-04
    ICS B32B027-00; C08L085-02
ICA C08F299-02
ICI C08L085-02
    42-10 (Coatings, Inks, and Related Products)
CC
    Section cross-reference(s): 38
FAN.CNT 1
                                      APPLICATION NO.
    PATENT NO.
                      KIND DATE
                                                                DATE
                                          -----
     -----
                       ____
                                                                 _____
    JP 06001869
                       A2 19940111 JP 1992-162381
                                                                 19920622
PRAI JP 1992-162381
                              19920622
CLASS
             CLASS PATENT FAMILY CLASSIFICATION CODES
PATENT NO.
               ____
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JP 06001869
               ICM C08J007-04
                ICS
                       B32B027-00; C08L085-02
                ICA
                       C08F299-02
                ICI
                       C08L085-02
                       C08J0007-04 [ICM, 5]; C08J0007-00 [ICM, 5, C*];
                IPCI
                       B32B0027-00 [ICS,5]; C08L0085-02 [ICS,5]; C08F0299-02
                       [ICA, 5]; C08F0299-00 [ICA, 5, C*]; C08L0085-02 [ICI, 5];
                       C08L0085-00 [ICI,5,C*]
AΒ
    Title parts comprise substrates, interlayers comprising (1) curable resins
    having thermal expansion coefficient (E) \geq 5 + 10-5
    cm-cm-1-°C-1 at 40° or (2) thermoplastic resins having
    modulus of tensile elasticity ≤2000 MPa, and surface layers
    comprising curable polyphosphazenes. Thus, 55.0 g
    hexachlorocyclotriphosphazene was treated with 143 g 2-hydroxyethyl
    methacrylate in toluene in the presence of pyridine at 80° for 8 h
     to give 1,1,3,3,5,5-hexa(methacryloyloxyethoxy)cyclotriphosphazene, 40
    parts of which was mixed with UA 306H 30, Sartomer 9505 (bifunctional
     urethane acrylate) 30, colloidal silica 100, Irgacure 184 (I) 1, Tinuvin
     144 1, butanol 50, and Me iso-Bu ketone 80 parts, applied onto a
     transparent polycarbonate substrate with UV-cured precoating containing
     Sartomer 9640 (bifunctional urethane acrylate) 80, Kayarad PEG4000DA 12,
    Kayarad DPHA 3, I 3, a UV absorber 10, and solvents 200 parts showing E 6.0 + 10-5 cm-cm-l-°C-1, treated at room temperature for 1 min,
     dried at 80° for 10 min, and UV-irradiated to give a 8-μm
     surface layer showing retention of appearance, adhesion, and color after
     2000-h exposure to weatherometer.
    polyphosphazene coating primer thermal expansion; polycarbonate substrate
     abrasion resistant coating; acrylic polyphosphazene coating; photocurable
     phosphazene polymer coating; tensile elasticity primer polyphosphazene
     coating; weatherability polyphosphazene coating; wear resistance
     polyphosphazene coating; primer acrylic resin polyphosphazene topcoating
     Coating materials
ΙT
        (abrasion- and heat- and weather-resistant, acrylic polyphosphazenes,
        primers with controlled thermal expansion and tensile elasticity for)
ΙT
     Urethane polymers, uses
     RL: USES (Uses)
        (acrylic, primers, for polyphosphazene top coatings, with
        weatherability and wear resistance)
ΙT
     155880-83-4P, 1,1,3,3,5,5-Hexa(methacryloylethylenedioxy)cyclotrip
     hosphazene-Sartomer 9505-UA 306H copolymer
     RL: PREP (Preparation)
        (preparation of, coatings, with weatherability and wear resistance, primers
        for)
     92832-53-6P
IT
     RL: PREP (Preparation)
        (preparation of, for curable coatings, with wear resistance and
        weatherability)
                155880-84-5P 156738-01-1P
ΙT
     9011-87-4P
     RL: PREP (Preparation)
```

```
(preparation of, for interlayers for polyphosphazene top coatings, with
       weatherability and wear resistance)
IT
    868-77-9
    RL: RCT (Reactant); RACT (Reactant or reagent)
       (reaction of, with cyclotriphosphazenes, for curable coatings)
    940-71-6, Hexachlorocyclotriphosphazene
ΙT
    RL: RCT (Reactant); RACT (Reactant or reagent)
       (reaction of, with hydroxy-containing acrylates, for curable phosphazenes)
    ANSWER 32 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
AN
    1994:446715 CAPLUS
DN
    121:46715
ED
    Entered STN: 23 Jul 1994
ΤI
    holographic recording material
    Tani, Motoaki
IN
    Fujitsu Ltd, Japan
PΑ
    Jpn. Kokai Tokkyo Koho, 5 pp.
SO
    CODEN: JKXXAF
DT
    Patent
LA
    Japanese
IC
    ICM G03H001-02
    ICS G03C009-08; G03F007-004; G03F007-027; G03F007-028; G03F007-033
CC
    74-8 (Radiation Chemistry, Photochemistry, and Photographic and Other
    Reprographic Processes)
                   KIND DATE APPLICATION NO.
FAN.CNT 1
                                                             DATE
    PATENT NO.
     _____
                                       -----
                                                               _____
    JP 05341702
                       A2 19931224 JP 1992-151026
                                                              19920611
PRAI JP 1992-151026
                             19920611
CLASS
              CLASS PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
               ____
              ICM G03H001-02
JP 05341702
                      G03C009-08; G03F007-004; G03F007-027; G03F007-028;
                ICS
                      G03F007-033
                      G03H0001-02 [ICM, 5]; G03C0009-08 [ICS, 5]; G03C0009-00
                IPCI
                       [ICS,5,C*]; G03F0007-004 [ICS,5]; G03F0007-027 [ICS,5];
                       G03F0007-028 [ICS,5]; G03F0007-033 [ICS,5]
    A holog, recording material showing high diffraction efficiency and light
     transmittance and good resistance to moisture comprises a
    photopolymerizable composition containing a binder resin containing a
carbazole ring in
     the structural unit, polymerizable phosphazene derivative and acrylic monomers
     having a refractive index lower than that of the binder resin., a
    photopolymn. initiator, and a photosensitizer.
    holog recording material polycarbazole binder; acrylic monomer holog
    recording material; phosphazene monomer holog recording material
ΙT
    Holography
        (photopolymerizable compns. containing polycarbazole binders and
       phosphazene and acrylic monomers for)
ΙT
     25067-59-8, Poly(N-vinylcarbazole)
     RL: USES (Uses)
        (photopolymerizable compns. containing polycarbazole binders and
       phosphazene derivative and acrylic monomers for)
     76-54-0, 2',7'-Dichlorofluorescein 94-36-0, Benzoyl peroxide, uses
ΙT
     103-01-5, N-Phenylglycine 3524-68-3, Pentaerythritol triacrylate
     6542-67-2, 2,4,6-Tris(trichloromethyl)-1,3,5-triazine
     Isopropyl benzoin ether 7473-98-5, 2-Hydroxy-2-methylpropiophenone
     12176-31-7, (n6-Benzene) (n5-cyclopentadienyl) iron(II) hexafluoropho
     sphate 15625-89-5, Trimethylolpropane triacrylate 24650-42-8, Benzyl
     dimethyl ketal 29570-58-9, Dipentaerythritol hexaacrylate 38215-36-0
     51325-91-8, 4-(Dicyanomethylene)-2-methyl-6-(p-dimethylaminostyryl)-4H-
     pyran 77473-08-6, 3,3',4,4'-Tetrakis(tert-butylperoxycarbonyl)benzopheno
         79579-93-4 92832-53-6 96273-26-6, 4-tert-Butyl-2,6-bis(4'-
     dimethylaminobenzylidene)cyclohexanone 100844-79-9
```

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RL: USES (Uses)
       (photopolymerizable compns. containing, for holog. recording materials)
L9
    ANSWER 33 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
ΑN
    1994:311768 CAPLUS
DN
    120:311768
ED
    Entered STN: 11 Jun 1994
ΤI
    Holographic recording material
IN
    Tani, Motoaki
PΑ
    Fujitsu Ltd, Japan
SO
    Jpn. Kokai Tokkyo Koho, 5 pp.
    CODEN: JKXXAF
DΤ
    Patent
LA
    Japanese
    ICM G03H001-02
    ICS G03C001-675; G03C001-72; G03F007-004; G03F007-027; G03F007-038
    74-12 (Radiation Chemistry, Photochemistry, and Photographic and Other
    Reprographic Processes)
FAN.CNT 1
                      KIND DATE
    PATENT NO.
                                       APPLICATION NO.
                                                              DATE
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                                                               -----
                       A2 19930702 JP 1991-329361
    JP 05165389
                                                               19911213
PRAI JP 1991-329361
                             19911213
CLASS
PATENT NO.
             CLASS PATENT FAMILY CLASSIFICATION CODES
 _____
               ____
JP 05165389
              ICM G03H001-02
               ICS
                      G03C001-675; G03C001-72; G03F007-004; G03F007-027;
                IPCI
                      G03H0001-02 [ICM, 5]; G03C0001-675 [ICS, 5]; G03C0001-72
                      [ICS,5]; G03F0007-004 [ICS,5]; G03F0007-027 [ICS,5];
                      G03F0007-038 [ICS,5]
AB
    The title holog, recording material comprises a binder resin containing
    carbazole rings in its repeating units, a phosphazene type polymerizable
    monomer of n lower than that of the binder resin, a photopolymn.
    initiator, and a sensitizer and the phosphazene type polymerizable monomer
    will be polymerized when exposed to light. This recording material is chemical
    stable and shows good humidity resistance and high sensitivity.
    holog recording material carbazole resin; phosphazene monomer holog
    recording material
IT
    Recording materials
       (holog., carbazole type binder resin and phosphazene type polymerizable
       monomer containing)
ΙT
    Holography
       (recording materials, carbazole type binder resin and phosphazene type
       polymerizable monomer containing)
    25067-59-8, Poly(N-vinylcarbazole) 26710-15-6, N-Vinylcarbazole-styrene
ΙT
    copolymer
    RL: USES (Uses)
       (binder resin, holog. recording material containing)
ΙT
    92832-53-6
    RL: USES (Uses)
       (polymerizable monomer, for holog. recording material)
    ANSWER 34 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
    1994:194166 CAPLUS
ΑN
DN
    120:194166
    Entered STN: 16 Apr 1994
ED
    UV-curable resin composition for low-shrinkage, weather-resistant coatings
ΤI
IN
    Ando, Hiroyuki
    Idemitsu Petrochemical Co., Ltd., Japan
PA
SO
    Eur. Pat. Appl., 22 pp.
    CODEN: EPXXDW
DT
    Patent
LA
    English
```

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ICM C08G079-02
    ICS C08F283-00; C08F299-02
CC
    42-10 (Coatings, Inks, and Related Products)
FAN.CNT 1
                                       APPLICATION NO. DATE
    PATENT NO.
                      KIND DATE
                                       -----
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                                                               _____
    EP 557943 A2 19930901 EP 1993-102805
EP 557943 A3 19931208
                                                               19930224
ΡI
        R: BE, CH, DE, FR, GB, IT, LI, NL, SE
    JP 05230156 A2 19930907 JP 1992-37693
                                                              19920225
                       AA
                                       CA 1993-2090304
    CA 2090304
                             19930826
                                                              19930224
PRAI JP 1992-37693
                      A
                             19920225
CLASS
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
 ______
EP 557943 . ICM C08G079-02
               ICS
                     C08F283-00; C08F299-02
                IPCI
                      C08G0079-02 [ICM,5]; C08G0079-00 [ICM,5,C*];
                      C08F0283-00 [ICS,5]; C08F0299-02 [ICS,5]; C08F0299-00
                      [ICS, 5, C*]
                      C07F0009-00 [I,C*]; C07F0009-6581 [I,A]; C08F0283-00
                IPCR
                      [I,A]; C08F0283-00 [I,C*]; G03F0007-027 [I,A];
                      G03F0007-027 [I,C*]
JP 05230156
                IPCI
                      C08F0299-02 [ICM,5]; C08F0299-00 [ICM,5,C*]
                      C08F0220-10 [ICM,5]; C08F0220-00 [ICM,5,C*];
CA 2090304
                IPCI
                      C08F0230-02 [ICS,5]; C08F0230-00 [ICS,5,C*]
                      C07F0009-00 [I,C*]; C07F0009-6581 [I,A]; C08F0283-00
                IPCR
                      [I,A]; C08F0283-00 [I,C*]; G03F0007-027 [I,A];
                      G03F0007-027 [I,C*]
AB
    The title composition comprises a curable cyclic or chain phosphazene compound
    having repeating unit -[NP(A)a(B)b]-(A = polymerizable curable group; B =
    polymerizable non-curing group; a, b > 0; a + b = 2), a multifunctional
    (meth) acrylate monomer or oligomer or a multifunctional urethane
    (meth) acrylate, and a crosslinkable compound having a cure shrinkage
    \leq 6\%. A coating composition containing 1,1,3,3,5,5-
    hexa(methacryloylethylenedioxy)cyclotriphosphazene (preparation from
    hexachlorocyclotriphosphazene and 2-hydroxyethyl methacrylate given),
    dipentaerythritol hexaacrylate, hexamethylenediol diacrylate, and
    trimethylolpropane triacrylate had shrinkage factor 5.8.
ST
    phosphazene acrylic low shrinkage coating; shrinkage low coating
    hexamethacryloylethylenedioxycyclotriphosphazene;
    methacryloylethylenedioxycyclotriphosphazene prepn low shrinkage coating
IT
    Coating materials
       (UV-curable, acrylic phosphazene-polyol (meth)acrylate-(meth)acrylate-
       modified polymer blends, low-shrinkage)
ΙT
    Polyesters, uses
    Urethane polymers, compounds
    RL: TEM (Technical or engineered material use); USES (Uses)
        (acrylates, coatings, acrylic phosphazene-polyol (meth)acrylate blends,
       UV-curable, low-shrinkage)
ΙT
    Epoxy resins, uses
    RL: TEM (Technical or engineered material use); USES (Uses)
        (acrylic, coatings, acrylic phosphazene-polyol (meth)acrylate blends,
       UV-curable, low-shrinkage)
ΙT
    Phosphonitrile compounds
       (cyclophosphazenes, N-(meth)acryloyloxyethoxy, coatings, blends with
       polyol (meth) acrylates and (meth) acrylate-modified polymers,
       UV-curable, low-shrinkage)
IT - Phosphonitrile compounds
        (phosphazenes, N-(meth)acryloyloxyethoxy, coatings, blends with polyol
        (meth)acrylates and (meth)acrylate-modified polymers, UV-curable,
       low-shrinkage)
    13048-33-4 15625-89-5
                             29570-58-9, Dipentaerythritol hexaacrylate
ΙT
    76482-93-4, Ebecryl 584 77001-81-1, UA 306H 126904-04-9, Sartomer C
    9505
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IC

```
(coating composition containing acrylic phosphazene derivative and,
UV-curable,
       low-shrinkage)
    79-10-7D, Acrylic acid, esters with diols and polyols 79-41-4D, esters
IT
    with diols and polyols
    RL: TEM (Technical or engineered material use); USES (Uses)
       (coating compns. containing acrylic phosphazene derivative and, UV-curable,
       low-shrinkage)
ΙT
    92832-53-6P
    RL: SPN (Synthetic preparation); PREP (Preparation)
       (preparation of, as UV-curable, low-shrinkage coating component)
IT
    868-77-9
    RL: RCT (Reactant); RACT (Reactant or reagent)
       (reaction of, with hexachlorocyclotriphosphazene, in preparation of
       UV-curable, low-shrinkage curable coating component)
ΙT
    940-71-6, Hexachlorocyclotriphosphazene
    RL: RCT (Reactant); RACT (Reactant or reagent)
       (reaction of, with hydroxyethyl methacrylate, in preparation of UV-curable.
       low-shrinkage coating component)
    ANSWER 35 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
    1994:65829 CAPLUS
ΑN
DN
    120:65829
ED
    Entered STN: 05 Feb 1994
    Electrophotographic photoreceptor
ΤI
    Kurahashi, Akihiko
IN
    Idemitsu Petrochemical Co., Ltd., Japan
PΑ
    PCT Int. Appl., 39 pp.
    CODEN: PIXXD2
DT
    Patent
    Japanese
LΑ
IC
    ICM G03G005-147
    74-3 (Radiation Chemistry, Photochemistry, and Photographic and Other
    Reprographic Processes)
FAN.CNT 1
                                       APPLICATION NO.
                                                               DATE
    PATENT NO.
                      KIND DATE
                                        _____
                                                              -----
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                       A1 19930415 WO 1992-JP1241
                                                              19920929
    WO 9307540
        W: US
        RW: AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, SE
    JP 05088382 A2 19930409 JP 1991-250843 19910930
                             19931013
                                                              19920929
                                        EP 1992-920384
    EP 564655
                       Αl
        R: BE, CH, DE, FR, GB, IT, LI, NL, SE
PRAI JP 1991-250843 A 19910930
    WO 1992-JP1241
                       W
                             19920929
CLASS
 PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
               ____
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             ICM G03G005-147
WO 9307540
                      G03G0005-147 [ICM, 5]
               IPCI
                      G03G0005-147 [I,A]; G03G0005-147 [I,C*]
               IPCR
                      G03G0005-05 [ICM, 5]; G03G0005-147 [ICS, 5]
 JP 05088382
              IPCI
                      G03G0005-147 [ICM, 5]
EP 564655
               IPCI
                      G03G0005-147 [I,A]; G03G0005-147 [I,C*]
               IPCR
                      G03G005/147D2H; G03G005/147D2F; G03G005/147D2D12
                ECLA
    The title electrophotog. photoreceptor has a photoconductive layer formed
AB
    on a conductive substrate and coated with a protective layer prepared from a
    composition comprising a reactive (di)pentaerythritol compound, a reactive
    phosphazene compound and a reactive siloxane compound The photoreceptor is
    excellent in abrasion resistance, surface smoothness, heat resistance,
    moisture resistance and so forth.
ST
    electrophotog photoreceptor protective layer
    Electrophotographic photoconductors and photoreceptors
ΙT
```

(phosphazene- and urethane-containing silicone-containing acrylic protective

RL: TEM (Technical or engineered material use); USES (Uses)

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layer for)
IΤ
    Phosphonitrile compounds
    RL: USES (Uses)
       (phosphazenes, electrophoretic photoreceptor surface layer containing)
IT
    Phosphazene polymers
    RL: USES (Uses)
       (acrylic, and urethane-containing silicone-containing, electrophoretic
       photoreceptor surface layer containing)
    Siloxanes and Silicones, uses
ΙT
    RL: USES (Uses)
       (di-Me, alc.-terminated, reaction product with diisocyanates, surface
       layer for electrophotog. photoreceptors from)
TT
    Acrylic polymers, uses
    RL: USES (Uses)
       (phosphazene, and urethane-containing silicone-containing, electrophoretic
       photoreceptor surface layer containing)
ΙT
    92832-53-6D, polymer with isocyanate-containing silicones and hydroxyl
    group-containing acrylates
    RL: USES (Uses)
       (electrophoretic photoreceptor surface layer containing)
    3524-68-3D, reaction product with isocyanate-containing silicone and acryloyl
IT
    group-containing phosphazene 29570-58-9D, Dipentaerythritol hexaacrylate,
    reaction product with acryloyl-containing phosphazene, isocyanate-containing
    silicone, and hydroxy-containing acrylate 60506-81-2D, reaction product with
    acryloyl phosphazenes, hydroxy-containing acrylates, and isocyanate-containing
    silicone
    RL: USES (Uses)
       (electrophotog. photoreceptor surface layer containing)
ΙT
    92832-53-6P
    RL: SPN (Synthetic preparation); PREP (Preparation)
       (preparation of)
    ANSWER 36 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
ΑN
    1994:56854 CAPLUS
DN
    120:56854
    Entered STN: 05 Feb 1994
ΕD
    Emulsions of phosphazene compounds and their cured products
TI
ΙN
    Ando, Hiroyuki
PΑ
    Idemitsu Petrochemical Co, Japan
SO
    Jpn. Kokai Tokkyo Koho, 12 pp.
    CODEN: JKXXAF
DT
    Patent
LA
    Japanese
    ICM C08F299-02
IC
    ICS C08L085-02; C09D185-02
CC
    42-10 (Coatings, Inks, and Related Products)
    Section cross-reference(s): 29
FAN.CNT 1
                       KIND DATE APPLICATION NO. DATE
    PATENT NO.
                                         _____
    _____
                       ----
                             19930831 JP 1992-27967 . 19920214
                       A2
    JP 05222142
PRAI JP 1992-27967
                              19920214
CLASS
             CLASS PATENT FAMILY CLASSIFICATION CODES
PATENT NO.
               _____
-----
JP 05222142
               ICM
                      C08F299-02
                      C08L085-02; C09D185-02
                ICS
                      C08F0299-02 [ICM,5]; C08F0299-00 [ICM,5,C*];
                IPCI
                       C08L0085-02 [ICS,5]; C08L0085-00 [ICS,5,C*];
                       C09D0185-02 [ICS,5]; C09D0185-00 [ICS,5,C*]
ΑB
    Title emulsions with good storage stability containing curable phosphazene
    compds., emulsifying agents, and H2O are applied on substrates and cured
    to form coatings with good hardness and resistance to scratching, heat,
    and staining. Thus, an aqueous emulsion containing 1,1,3,3,5,5-
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hexa (methacryloylethylenedioxy) cyclotriphosphazene 30, Bz2O2 1, EtOH 10,

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and Na alkylbenzenesulfonate 1 g was sprayed on a steel plate and dried at
    150° for 20 min to form a coating showing cross-cut adhesion
    100/100, pencil hardness 9H, and good resistance to staining, boiling
    water, and salt spraying.
ST
    acrylic phosphazene polymer emulsion coating;
    methacryloyloxyethylenedioxycyclotriphosphazene polymer emulsion coating;
    heat resistance phosphazene polymer coating
ΙT
    Phosphazene polymers
    RL: TEM (Technical or engineered material use); USES (Uses)
       (acrylic, coatings, preparation of hard heat-resistant)
IT
    Coating materials
       (emulsion, acrylic phosphazene polymers, with good heat resistance and
       hardness)
    868-77-9DP, 2-Hydroxyethyl methacrylate, reaction products with
TΤ
    hexachlorocyclotriphosphazene and sodium phenolate 25231-98-5DP,
    Hexachlorocyclotriphosphazene polymer, reaction products with hydroxyethyl
    methacrylate and sodium phenolate 26085-02-9DP,
    Hexachlorocyclotriphosphazene polymer, sru, reaction products with
    hydroxyethyl methacrylate and sodium phenolate
                                                  93891-06-6P
    122564-38-9P
    RL: TEM (Technical or engineered material use); PREP (Preparation); USES
       (coatings, preparation of, heat-resistant, high-hardness)
ΙT
    12597-69-2, Steel, uses
    RL: USES (Uses)
       (plate, coatings for, acrylic phosphazene polymers as, with good heat
       resistance and hardness)
ΙT
    92832-53-6P
    RL: SPN (Synthetic preparation); PREP (Preparation)
       (preparation of)
ΙT
    868-77-9, 2-Hydroxyethyl methacrylate
    RL: RCT (Reactant); RACT (Reactant or reagent)
       (reaction of, with hexacyclotriphosphazene)
ΙT
    940-71-6, Hexachlorocyclotriphosphazene
    RL: RCT (Reactant); RACT (Reactant or reagent)
       (reaction of, with hydroxyethyl methacrylate)
L9
    ANSWER 37 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
    1993:497606 CAPLUS
AN
DN
    119:97606
ED
    Entered STN: 04 Sep 1993
    Hydrophilized phosphazene polymer articles
TΙ
IN
    Hosono, Hiroshi; Kurasaki, Shoichi; Taniguchi, Takashi
PA
    Toray Industries, Japan
    Jpn. Kokai Tokkyo Koho, 4 pp.
    CODEN: JKXXAF
DT
    Patent
    Japanese
LA
    ICM C08J007-12
IC
    ICS C09D005-00; C09K003-18
ICA C08G079-02; C09D185-02
    38-3 (Plastics Fabrication and Uses)
    Section cross-reference(s): 35
FAN.CNT 1
                       KIND DATE APPLICATION NO.
    PATENT NO.
                                                              DATE
                                         _____
    -----
                       ____
                              -----
                                                               _____
                       A2
    JP 05051472
                              19930302 JP 1991-217184
                                                               19910828
PRAI JP 1991-217184
                              19910828
            CLASS PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
 -----
               _____
 JP 05051472
              ICM
                      C08J007-12
                ICS
                      C09D005-00; C09K003-18
                ICA
                      C08G079-02; C09D185-02
                      CO8J0007-12 [ICM,5]; CO8J0007-00 [ICM,5,C*];
                IPCI
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C09D0005-00 [ICS,5]; C09K0003-18 [ICS,5]; C08G0079-02 [ICA,5]; C08G0079-00 [ICA,5,C*]; C09D0185-02 [ICA,5]; C09D0185-00 [ICA,5,C*]
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- AB Articles of polymers containing P:N units are hydrophilized by treatment with alkalis until they show water contact angle ≤40° for improved antifogging properties and scratch resistance. The polymers containing P:N units may be obtained by introducing polymerizable functional groups into polydichlorophosphazenes and polymerizing Thus, a polycarbonate sheet was coated with a mixture of hexamethacryloyloxyethylcyclotriphosphaze ne, p-chlorobenzophenone, and MIBK, UV-irradiated, and immersed in 20% aqueous NaOH to give a specimen with water contact angle 13° and good scratch resistance and antifogging properties.
- ST phosphazene polymer hydrophilization alkali treatment; scratch resistance hydrophilic phosphazene polymer; antifogging hydrophilic phosphazene polymer
- IT Polycarbonates, miscellaneous

RL: MSC (Miscellaneous)

(coated with phosphazene polymers, hydrophilization of, by alkali treatment, for improved antifogging properties and scratch resistance)

IT Phosphazene polymers

RL: PRP (Properties)

(hydrophilization of, with alkalis, for improved antifogging properties and scratch resistance)

IT Alkali metal hydroxides

RL: USES (Uses)

(hydrophilization with, of phosphazene polymers, for improved antifogging properties and scratch resistance)

IT Wettability

(improvement of, of phosphazene polymers, by alkali treatment)

IT Epoxy resins, compounds

RL: USES (Uses)

(acrylates, polymers with hexakis(methacryloyloxyethyl)cyclotriphosphaz ene and polyether (meth)acrylates, hydrophilization of, with alkali)

IT Polyethers, uses

RL: USES (Uses)

(methacrylate-containing, polymers, with hexamethacryloyloxyethylcyclotriph osphazene and epoxy di(meth)acrylates, hydrophilization of, with alkalis)

IT Epoxy resins, compounds.

RL: USES (Uses)

(methacrylates, polymers with hexamethacryloyloxyethylcyclotriphosphaze ne and polyether (meth)acrylates, hydrophilization of, with alkali)

IT 92832-53-6D, polymers with polyether (meth)acrylates and epoxy di(meth)acrylates 93891-06-6

RL: PRP (Properties)

(hydrophilization of, with alkali, for improved antifogging properties and scratch resistance)

IT 1310-73-2, Sodium hydroxide, uses

RL: USES (Uses)

(hydrophilization with, of phosphazene polymers, for improved antifogging properties and scratch resistance)

IT 7732-18-5

RL: USES (Uses)

(wettability, improvement of, of phosphazene polymers, by alkali treatment)

- L9 ANSWER 38 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
- AN 1993:497605 CAPLUS
- DN 119:97605
- ED Entered STN: 04 Sep 1993
- TI Coated phosphazene polymer articles
- IN Hosono, Hiroshi; Kurasaki, Shoichi; Taniguchi, Takashi
- PA Toray Industries, Japan
- SO Jpn. Kokai Tokkyo Koho, 10 pp. CODEN: JKXXAF

```
DT
    Patent
LA
    Japanese
    ICM C08J007-04
TC
    ICS C09D005-00; C09D129-04
    38-3 (Plastics Fabrication and Uses)
CC
    Section cross-reference(s): 40
FAN.CNT 1
                                     APPLICATION NO.
    PATENT NO.
                     KIND DATE
                                                              DATE
                                        -----
                                                              -----
    -----
                      ____
    JP 05051471 A2 19930302 JP 1991-215591
                                                               19910827
PRAI JP 1991-215591
                             19910827
CLASS
PATENT NO.
              CLASS PATENT FAMILY CLASSIFICATION CODES
_____
               _____
JP 05051471
              ICM C08J007-04
                ICS
                      C09D005-00; C09D129-04
                IPCI
                      CO8J0007-04 [ICM, 5]; CO8J0007-00 [ICM, 5, C*];
                      C09D0005-00 [ICS,5]; C09D0129-04 [ICS,5]; C09D0129-00
                      [ICS, 5, C*]
    Articles containing polymers with P:N units at least on their surface are
AB
    coated with a composition of poly(vinyl alc.) with saponification degree \geq 70
    mol% and average d.p. ≥300 and crosslinkers for durable antifogging
    properties. The polymers with P:N units may be obtained by introducing
    polymerizable functional groups to polydichlorophosphazenes and polymerizing
    Thus, hexamethacryloyloxyethylcyclotriphosphazene was polymerized under
    UV-irradiation to give a 2 mm-thick transparent sheet, which was immersed in
    aqueous NaOH, water-washed, coated with a composition of aqueous poly(vinyl
alc.)
    (saponification degree 91.0-94.0 mol%), hydrolyzate of \gamma-
    glycidoxypropyltrimethoxysilane, dioxane, MeOH, dimethylimidazolidinone,
    Al acetylacetonate, and silicone surfactant, cured at 130°, and
    immersed in aqueous NaOH. The specimen showed no peeling in a cross-cut
    adhesion test, pencil hardness 2H, and no fogging when breathed on.
    phosphazene polymer molding antifogging coating; polyvinyl alc crosslinker
    antifogging coating; siloxane polyvinyl alc antifogging coating
ΙT
    Phosphazene polymers
    RL: USES (Uses)
       (moldings, antifogging coatings for, poly(vinyl alc.) and silane
       crosslinkers as)
ΙT
    Polycarbonates, miscellaneous
    RL: MSC (Miscellaneous)
       (phosphazene polymer-coated, antifogging coatings for, poly(vinyl alc.)
       and silane crosslinkers as)
ΙT
    Antifogging agents
       (coatings, poly(vinyl alc.) and silane crosslinkers, for phosphazene
ΙT
    Siloxanes and Silicones, compounds
    RL: USES (Uses)
       (polymers, with poly(vinyl alc.), antifogging coatings for phosphazene
       polymers)
IT
    9002-89-5D, Poly(vinyl alcohol), polymers with silane hydrolyzates
    RL: USES (Uses)
        (coatings, antifogging, for phosphazene polymer moldings)
ΙT
    92832-53-6P
    RL: PREP (Preparation)
       (preparation of, crosslinked, antifogging coatings for, poly(vinyl alc.) and
       silane crosslinkers as)
    ANSWER 39 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
AN
    1993:40920 CAPLUS
DN
    118:40920
ED
    Entered STN: 03 Feb 1993
    Phosphorus polymer coatings
ΤI
ΙN
    Yaguchi, Atsunori
PA
    Idemitsu Petrochemical Co., Ltd., Japan
```

```
Jpn. Kokai Tokkyo Koho, 9 pp.
SO
    CODEN: JKXXAF
DT
    Patent
LA
    Japanese
    ICM C09D007-12
IC
    ICS C09D185-02; C09D201-00
CC
    42-10 (Coatings, Inks, and Related Products)
FAN.CNT 1
    PATENT NO.
                     KIND DATE
                                       APPLICATION NO.
    -----
                      ----
                                      -----
                       A2 19920507 JP 1990-255558 19900925
    JP 04132772
PRAI JP 1990-255558
                            19900925
CLASS
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
 ______
JP 04132772 ICM C09D007-12
               ICS . CO9D185-02; CO9D201-00
               IPCI C09D0007-12 [ICM,5]; C09D0185-02 [ICS,5]; C09D0185-00
                      [ICS,5,C*]; C09D0201-00 [ICS,5]
AB
    Title coatings, coatable on various substrates contain good solvents (e.g.
    ketones, aromatic hydrocarbons, glycol ethers or esters) and poor solvents
    (e.g. alcs., aliphatic hydrocarbons). Thus, a UV-curable composition
    3, PhMe 2, MeOH 3, benzophenone 0.50 and 1,1,3,3,5,5-hexa(methacryloyl
    ethylenedioxy)cyclotriphosphazene (from 2-hydroxyethyl methacrylate and
    hexachlorocyclotriphosphazene) 10 g showed good adhesion to polycarbonate,
    cellulose, urethane rubber, acrylic styrene polymer or acrylic polymer
    sheets and gave abrasion-resistant films.
ST
    acrylic phosphazene polymer coating solvent
    Polycarbonates, uses
    Rubber, urethane, uses
    RL: USES (Uses)
       (coatings for, UV-curable acrylic phosphazenes in good-poor solvents
    Coating materials
       (abrasion-resistant, UV-curable, phosphazene methacrylates in good-poor
       solvent blends, for plastics or rubbers)
    Phosphazene polymers
ΙT
    RL: TEM (Technical or engineered material use); USES (Uses)
       (cyclo-, methacrylic, coatings, in good-poor solvent blends, for
       plastics or rubbers)
    100-42-5D, Styrene, polymers with acrylic compds. 9004-36-8, CAB
ΙT
    9004-39-1, Cellulose acetate propionate
    RL: USES (Uses)
       (coatings for, UV-curable acrylic phosphazenes in good-poor solvents
       as)
ΙT
    93891-06-6
    RL: TEM (Technical or engineered material use); USES (Uses)
       (coatings, UV-cured, for plastics or rubbers)
ΙT
    16526-96-8, \alpha-Hydroxyethyl methacrylate
    RL: USES (Uses)
       (condensation of, with hexachlorotriphosphazene)
IT
    940-71-6, Hexachlorocyclotriphosphazene
    RL: USES (Uses)
       (condensation of, with hydroxyethyl methacrylate)
IT
    92832-53-6P
    RL: RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)
       (preparation and polymerization of, for coating)
    ANSWER 40 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
ΑN
    1992:614567 CAPLUS
DN
    117:214567
    Entered STN: 28 Nov 1992
ED
TI
    The characteristics of cyclic phosphazene and its applications for hard
```

coatings

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ΑU
     Yaguchi, A.; Mori, S.; Kitayama, M.; Onda, T.; Kurahashi, A.; Ando, H.
CS
     Planning Dev. Dep., Idemitsu Petrochem. Co. Ltd., Tokyo, 100, Japan
     Thin Solid Films (1992), 216(1), 123-5
SO
     CODEN: THSFAP; ISSN: 0040-6090
DT
     Journal
     English
LA
CC
     42-10 (Coatings, Inks, and Related Products)
     Hexachlorocyclotriphosphazene is condensed with 2-hydroxyethyl
AΒ
     methacrylate, and the adduct is then polymerized with UV light.
                                                                      The
     properties of the resulting coating are presented. The coating has
     extreme hardness as well as heat, chemical, and stain resistance and is
     suitable for use as a hard coating for wood or PVC floor tiles.
ST
     phosphazene polymethacrylate crosslinked coating property; PVC wood
     coating polyphosphazene polymethacrylate
IT
     Tiles
        (floor, PVC, coatings for, UV-curable polyphosphazene polymethacrylates
IT
     Coating materials
        (photocurable, phosphazene polymethacrylates, preparation and properties of,
        for wood and PVC floor tiles)
IT
     93891-06-6P
     RL: SPN (Synthetic preparation); TEM (Technical or engineered material
     use); PREP (Preparation); USES (Uses)
        (coatings, UV-curable, preparation and properties of)
     9002-86-2
IT
     RL: DEV (Device component use); USES (Uses)
        (floor tiles from, UV-curable coatings for, polyphosphazene
        polymethacrylates as)
IT
     92832-53-6P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (preparation and polymerization of)
     868-77-9
IT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with hexachlorocyclotriphosphazene)
ΙT
     940-71-6
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with hydroxyethyl methacrylate)
L9
     ANSWER 41 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
AN
     1992:501021 CAPLUS
DN
     117:101021
     Entered STN: 05 Sep 1992
ED
     Photosensitive composition for heat-resistant pattern formation
ΤI
     Tani, Motoaki; Horikoshi, Eiji; Watanabe, Isao
IN
PA
     Fujitsu Ltd., Japan
SO
     Eur. Pat. Appl., 23 pp.
     CODEN: EPXXDW
DT
     Patent
LA
     English
IC
     ICM G03F007-037
     74-4 (Radiation Chemistry, Photochemistry, and Photographic and Other
CC
     Reprographic Processes)
     Section cross-reference(s): 76
FAN.CNT 4
                         KIND
                                DATE
                                            APPLICATION NO.
                                                                   DATE
     PATENT NO.
                         ____
                                -----
                                            -----
                          A2
                                            EP 1991-303379
                                19911023
                                                                   19910416
PΙ
     EP 453237
                                19920930 .
     EP 453237
                         А3
                                19981209
     EP 453237
                         В1
         R: DE, FR, GB
                                            JP 1990-279088
                                                                   19901019
                         A2
                                19920122
     JP 04018450
                          B2
                                19991122
     JP 2980359
                         Α
PRAI JP 1990-99788
                                19900416
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Α

19901019

JP 1990-279088

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CLASS
                CLASS PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
                ____
 EP 453237
                ICM
                        G03F007-037
                        G03F0007-037 [ICM,5]; G03F0007-032 [ICM,5,C*]
                 IPCI
                        G03F0007-032 [I,C*]; G03F0007-037 [I,A]; H05K0003-00
                 IPCR
                        [I,A]; H05K0003-00 [I,C*]; H05K0003-38 [N,A];
                        H05K0003-38 [N,C*]; H05K0003-46 [N,A]; H05K0003-46
                        [N,C^*]
                        G03F007/037; H05K003/00K3F
                 ECLA
 JP 04018450
                 IPCI
                        C08L0079-08 [ICM,5]; C08L0079-00 [ICM,5,C*];
                        C08G0073-10 [ICS,5]; C08G0073-00 [ICS,5,C*];
                        C09D0004-02 [ICS,5]; C09D0179-08 [ICS,5]; C09D0179-00
                        [ICS, 5, C*]
    A photosensitive composition for forming heat-resistant resin patterns for use
AΒ
     in the production of elec. circuit boards and semiconductor devices comprises
     a varnish of a polyimide or polybismaleimide precursor, a polymerizable
    monomer or oligomer compatible with the varnish and capable of providing a
    heat-resistant resin upon polymerization, and a photosensitive polymerization
initiator,
     wherein the monomer or oligomer is selected from a group consisting of
     acrylic acid and methacrylic compds., compds. having ≥ 2 functional
     groups and isocyanurate structures, phosphazenic compds., and compds.
    having acryloyl and methacryloyl groups at terminals and/or side chains.
ST
    photosensitive compn heat resistant pattern
ΙT
     Semiconductor devices
        (photopolymerizable compns. containing polyimide precursors and acrylic
        monomers for production of heat-resistant resin patterns for production of)
ΙT
     Polyimides, uses
     RL: USES (Uses)
        (precursors, photopolymerizable compns. containing acrylic monomers and,
        for production of heat-resistant resin patterns for manufacture of
semiconductor
        devices)
IT
     Photoimaging compositions and processes
        (photopolymerizable, containing polyimide precursors, acrylic monomers, and
       photoinitiators for heat-resistant resin pattern production)
ΙT
     Electric circuits
        (printed, boards, photopolymerizable compns. containing polyimide
       precursors and acrylic monomers for production of heat-resistant resin
       patterns for production of)
ΙT
     9043-05-4
     RL: USES (Uses)
        (photopolymerizable compns. containing acrylic monomers and, for production
of
       heat-resistant resin patterns for manufacture of semiconductor devices)
ΙT
     94-36-0, Benzoyl peroxide, uses 103-01-5, N-Phenylglycine
                                                                   119-61-9,
                         6652-28-4, Isopropyl benzoin ether
     Benzophenone, uses
                                                               7473-98-5
     12176-31-7
                 15774-82-0, 2-Methylthioxanthone
                                                     24504-22-1
                                                                  24650-42-8
     77473-08-6, 3,3',4,4'-Tetra(tert-butylperoxycarbonyl)benzophenone
     RL: USES (Uses)
        (photopolymerizable compns. containing polyimide precursors and acrylic
        monomers and, for heat-resistant resin pattern formation in manufacture of
        semiconductor devices)
                                   108-80-5, Isocyanuric acid
ΙT
     106-90-1, Glycidyl acrylate
                                                                3524-68-3,
                                   15625-89-5, Trimethylolpropane triacrylate
     Pentaerythritol triacrylate
     29570-58-9, Dipentaerythritol hexaacrylate
                                                 40220-08-4 92832-53-6
     142875-50-1
                   142875-51-2
     RL: USES (Uses)
        (photopolymerizable compns. containing polyimide precursors and, for
        of heat-resistant resin patterns for production of semiconductor devices)
     ANSWER 42 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
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1992:500917 CAPLUS

ΑN

```
DN
    117:100917
    Entered STN: 05 Sep 1992
ED
TΙ
    Electrophotographic photosensitive member
    Kurahashi, Akihiko; Kageyama, Akira
ΙN
    Idemitsu Petrochemical Co., Ltd., Japan
PA
    Eur. Pat. Appl., 20 pp.
SO
    CODEN: EPXXDW
DT
    Patent
    English
LA
    ICM G03G005-147
IC
    ICS G03G005-05
    74-3 (Radiation Chemistry, Photochemistry, and Photographic and Other
CC
    Reprographic Processes)
    Section cross-reference(s): 28
    PATENT NO. KIND DATE APPLICATION NO. DATE
FAN.CNT 1
                                                             _____
    EP 443626 A2 19910828 EP 1991-102697
EP 443626 A3 19920429
                                                             19910223
PΙ
       R: BE, CH, DE, FR, GB, IT, LI, NL, SE
JP 03246553 A2 19911101 JP 1990-44007
PRAI JP 1990-44007 A 19900223
                                                             19900223
 PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
 _____
EP 443626 ICM G03G005-147
               ICS G03G005-05
               IPCI G03G0005-147 [ICM, 5]; G03G0005-05 [ICS, 5]
               IPCR C07F0009-00 [I,C*]; C07F0009-6593 [I,A]; G03G0005-05
                     [I,A]; G03G0005-05 [I,C*]; G03G0005-147 [I,A];
                     G03G0005-147 [I,C*]
An electrophotog. photosensitive member has an electroconductive support,
    a photoconductive layer, and a protective layer composed of a cured resin
    of a curable phosphazene compound as the outermost layer or the surface of
    the photoconductive layer is composed of the curable phosphazene compound
    The electrophotog. photosensitive member has a high surface hardness, is
    excellent in durability, and provides stable images without any
    electrophotog photoconductor phosphazene surface
    Electrophotographic photoconductors and photoreceptors
       (with surface layers containing cured phosphazene compds.)
ΙT
    92832-53-6P
                142996-07-4P
    RL: SPN (Synthetic preparation); PREP (Preparation)
       (preparation and use of, in surface layers of electrophotog.
       photoconductors)
    ANSWER 43 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
    1992:409964 CAPLUS
ΑN
    117:9964
DN
    Entered STN: 11 Jul 1992
ED
    Curable phosphazene compositions and photochromic products
TI
    Kurahashi, Akihiko
IN
    Idemitsu Petrochemical Co., Ltd., Japan
PΑ
    Jpn. Kokai Tokkyo Koho, 10 pp.
    CODEN: JKXXAF
DT
    Patent
LA
    Japanese
IC
    ICM C08L085-02
    ICS C09K009-02
    42-10 (Coatings, Inks, and Related Products)
    Section cross-reference(s): 74
FAN.CNT 1
                     KIND DATE
                                                         DATE
                                       APPLICATION NO.
    PATENT NO.
    -----
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JP 04025564
                       A2
                              19920129
                                         JP 1990-130396
                                                               19900522
PΙ
PRAI JP 1990-130396
                              19900522
CLASS
 PATENT NO.
              CLASS PATENT FAMILY CLASSIFICATION CODES
 _____
               _____
JP 04025564
                ICM
                     C08L085-02
                ICS
                     C09K009-02
                C09K0009-02 [ICS,5]
AB
    The title compns. forming photochromic coatings with good surface hardness
    and adhesion comprise curable phosphazenes [N:PAaBb]n (A = curable group;
    B = \text{noncurable group}; a + b = 2; n = 3-18) and organic photochromic compds.
    Thus, a polycarbonate plate was sprayed with a composition containing cyclic
    [N:P(OC2H4OCOCMe:CH2)2]3 25, 1-hydroxycyclohexyl Ph ketone 1,
    spironaphthoxazine 1, and solvents 50 g and irradiated by UV to form a
    photochromic coating with pencil hardness 7H and good cross-cut adhesion.
    phosphazene curable coating photochromic; adhesion coating phosphazene
ST
    polymer photochromic; hardness coating phosphazene polymer photochromic
ΙT
    Phosphazene polymers
    RL: USES (Uses)
       (acrylic, coatings, containing photochromic compds., with good adhesion and
       hardness)
ΙT
    Photochromic substances
       (spironaphthoxazines, phosphazene polymer coatings containing)
ΙT
    Coating materials
       (photochromic, curable phosphazene-based, with good adhesion and
       hardness)
ΙT
    92832-53-6P
    RL: RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)
       (preparation and polymerization of)
    93891-06-6P
    RL: PREP (Preparation)
       (preparation of, coatings, containing photochromic compds., with good
adhesion
       and hardness)
IT · 868-77-9, 2-Hydroxyethyl methacrylate
    RL: RCT (Reactant); RACT (Reactant or reagent)
       (reaction of, with hexachlorocyclotriphosphazene)
     940-71-6, Hexachlorocyclotriphosphazene
TΤ
     RL: RCT (Reactant); RACT (Reactant or reagent)
       (reaction of, with hydroxyethyl methacrylate)
    ANSWER 44 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
AN
    1992:49025 CAPLUS
DN
    116:49025
    Entered STN: 08 Feb 1992
ED
    Reversible thermal recording media with protective layer containing
TΤ
    triazatriphosphorines
    Suzuki, Masayasu; Iida, Tsutomu; Aoyama, Koichi
IN
    Tomoegawa Paper Co., Ltd., Japan
PA
    Jpn. Kokai Tokkyo Koho, 6 pp.
SO
    CODEN: JKXXAF
DT
    Patent
    Japanese
LA
IC
     ICM B41M005-36
CC
     74-12 (Radiation Chemistry, Photochemistry, and Photographic and Other
     Reprographic Processes)
FAN.CNT 1
                              DATE
                                       APPLICATION NO.
                                                              DATE
     PATENT NO.
                      KIND
                              -----
                                          ______
    JP 03180390
                       A2
                              19910806
                                        JP 1989-318787
                                                               19891211
                              19920806
     JP 04048352
                       B4
PRAI JP 1989-318787
                              19891211
CLASS
```

CLASS PATENT FAMILY CLASSIFICATION CODES

PATENT NO.

```
JP 03180390 ICM B41M005-36
               IPCI B41M0005-36 [ICM, 5]
AB
     The title media are prepared by forming a heat-sensitive layer containing a low
     mol. weight organic compound dispersed in a high mol. weight organic compound,
of which
     the transparency is reversibly changed by heating, on a colored support
     and subsequently forming a protective layer containing mainly a thermally
     hardenable triazatriphosphorine. The media provide high contrast images
     which can be erased by heating, and show good durability in repeated use.
     Thus, a carbon black-colored polyester film was coated with a recording
     layer containing behenic acid and VYHH (vinyl acetate-vinyl chloride
     copolymer) and with a protective layer containing cyclic compound
     [NP(OC2H4OCOCMe:CH2)2]3 and Al(OH)3 and irradiated with an electron beam
     to give a thermal recording film.
ST
     reversible thermal recording medium; phosphagene protective layer thermal
     recording
     Printing, nonimpact
ΙT
        (thermal, materials for, reversible, with triazatriphosphorine
        protective layer)
    112-85-6, Behenic acid 9003-22-9, VYHH
IT
     RL: USES (Uses)
        (reversible thermal recording material using)
     7251-15-2 21645-51-2, Aluminum hydroxide, uses 92832-53-6
IΤ
     138495-70-2
     RL: USES (Uses)
        (reversible thermal recording medium protective layer using)
L9
     ANSWER 45 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
AN
    1991:208825 CAPLUS
DN
    114:208825
    Entered STN: 31 May 1991
ED
    Manufacture and uses of phosphazene-coated styrene-based resin composites
ΤI
    Yaquchi, Atsunori; Funaki, Keisuke
IN
    Idemitsu Petrochemical Co., Ltd., Japan
PΑ
SO
     Eur. Pat. Appl., 26 pp.
     CODEN: EPXXDW
DT
     Patent
LA
     English
     ICM C08J007-04
IC
     ICS C09D185-02; B41J031-05; G11B005-702; G11B005-71
ICI C08L025-06
     38-3 (Plastics Fabrication and Uses)
     Section cross-reference(s): 42, 75, 77
FAN.CNT 1
                    KIND DATE APPLICATION NO.
     PATENT NO.
                                                                DATE
                                           -----
    EP 376021 A2 19900704 EP 1989-122644
EP 376021 A3 19911106
EP 376021 B1 19960424
                                                                   19891208
PΤ
        R: BE, CH, DE, FR, GB, IT, LI, NL, SE
JP 02162038

JP 07077790

B4 19950823

JP 02162083

A2 19900621

JP 1988-316168

JP 2733078

US 5082717

CA 2005675

PRAI JP 1988-316168

A 19881216

JP 1989-19497

A 19890131

CLASS
                                           JP 1988-316168
                                                                   19881216
                                                                   19881216
                                                                   19891130
                                          CA 1989-2005675
                                                                   19891215
CLASS
 PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
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                ____
               ICM C08J007-04
 EP 376021
```

C09D185-02; B41J031-05; G11B005-702; G11B005-71

ICS

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ICI
                        C08L025-06
                        C08J0007-04 [ICM,5]; C08J0007-00 [ICM,5,C*];
                 IPCI
                        C09D0185-02 [ICS,5]; C09D0185-00 [ICS,5,C*];
                        B41J0031-05 [ICS,5]; G11B0005-702 [ICS,5]; G11B0005-71
                        [ICS,5]; C08L0025-06 [ICI,5]; C08L0025-00 [ICI,5,C*]
                 IPCR
                        C07F0009-00 [I,C*]; C07F0009-6581 [I,A]; C08J0007-00
                        [I,C*]; C08J0007-04 [I,A]; G11B0005-62 [I,C*];
                        G11B0005-702 [I,A]; G11B0005-702 [I,C*]; G11B0005-72
                        [I,C*]; G11B0005-725 [I,A]; G11B0005-73 [I,A];
                        G11B0005-735 [I,A]
                        C07F009/6581F2; C08J007/04L85+L25/06; G11B005/702;
                 ECLA
                        G11B005/702C; G11B005/702E; G11B005/725; G11B005/73B;
                        G11B005/735
 JP 02162038
                        B32B0027-30 [ICM,5]; B05D0007-02 [ICS,5]; B05D0007-24
                 IPCI
                        [ICS, 5]; B32B0027-16 [ICS, 5]; B32B0027-28 [ICS, 5];
                        C08F0112-00 [ICS,5]
 JP 02162083
                 IPCI
                        B41J0031-00 [ICM, 5]; C08F0012-08 [ICS, 5]; C08F0012-00
                        [ICS, 5, C*]
                 ECLA
                        CO7F009/6581F2; C08J007/04L85+L25/06; G11B005/702;
                        G11B005/702C; G11B005/702E; G11B005/725; G11B005/73B;
                        G11B005/735
                 IPCI
 US 5082717
                        B32B0027-28 [ICM,5]; C08J0007-04 [ICS,5]; C08J0007-00
                        [ICS,5,C*]; B41J0031-05 [ICS,5]; C09D0185-02 [ICS,5];
                        C09D0185-00 [ICS,5,C*]
                 IPCR
                        C07F0009-00 [I,C*]; C07F0009-6581 [I,A]; C08J0007-00
                        [I,C*]; C08J0007-04 [I,A]; G11B0005-62 [I,C*];
                        G11B0005-702 [I,A]; G11B0005-702 [I,C*]; G11B0005-72
                        [I,C*]; G11B0005-725 [I,A]; G11B0005-73 [I,A];
                        G11B0005-735 [I,A]
                 NCL
                        428/207.000; 428/412.000; 428/419.000; 428/421.000;
                        428/447.000; 428/480.000; 428/494.000; 428/495.000;
                        428/516.000; 428/518.000; 428/521.000; 428/843.000;
                        428/900.000; 526/160.000; 528/168.000
CA 2005675
                        C08J0007-04; C08J0007-00 [C*]
                 IPCI
                 IPCR
                        C07F0009-00 [I,C*]; C07F0009-6581 [I,A]; C08J0007-00
                        [I,C*]; C08J0007-04 [I,A]; G11B0005-62 [I,C*];
                        G11B0005-702 [I,A]; G11B0005-702 [I,C*]; G11B0005-72
                        [I,C*]; G11B0005-725 [I,A]; G11B0005-73 [I,A];
                        G11B0005-735 [I,A]
AΒ
    A styrene (I)-based resin composite, which comprises a I-based polymer
    having a syndiotactic configuration or its composition coated with a curable
    heat-resistant phosphazene resin is prepared The composite material, in
     which a hard coating having high hardness and adhesiveness is formed on
     the surface of the I-based resin molding material, is useful in magnetic
     recordings, electronics and other applications. Thus, polystyrene (II)
     having a syndiotactic configuration, was prepared by polymn of I in the
     presence of cyclopentadienyltitanium trichloride and methylaluminoxane
     catalysts, and subsequent solvent extraction II was mixed with a phosphite and
     a phenolic antioxidants, extruded, pelletized, extruded into a sheet,
     stretched to a d. 1.04 and crystallinity 12%, biaxially drawn at
     120° and annealed at 250° to give a film 200-mm thick.
     Then, 1,1,3,3,5,5-hexa(methacryloyl ethylenedioxyl)cyclotriphosphazene was
     mixed with a 1:1 iso-PrOH-MeCOEt blend to a viscosity 10 cP, mixed with
     10-hydrroxycyclohexyl Ph ketone photoinitiator, coated onto II molding
     material and cured by UV radiation to give a coated film having a
     thickness 6-µm, Taber abrasion 8%, and sand abrasion 12%, compared with
     Taber abrasion 38% and sand abrasion 46% for a similar II without the
     curable phosphazene coating.
ST
     polystyrene composite polyphosphazene coating manuf; syndiotactic
     polystyrene curable phosphazene coating; magnetic recording film
     polystyrene composite; UV curable phosphazene coating composite
ΙT
     Coating materials
        (phosphazene polymers, UV-curable, for styrene-based resin composites,
        abrasion-resistant)
```

ΙT

Heat-resistant materials

```
TΤ
    Recording materials
        (magnetic, styrene-based resin-phosphazene coating composites as,
       abrasion-resistant)
IT
     92832-53-6
    RL: USES (Uses)
        (UV-curable, coatings, for styrene-based resin composites,
       abrasion-resistant)
    24937-78-8, Ethylene-vinylacetate copolymer
ΙT
    RL: USES (Uses)
        (binders, for inks, for styrene-based resin composites)
ΙT
    28325-75-9P, Syndiotactic polystyrene
    RL: PREP (Preparation)
        (composites with curable phosphazene polymer coatings, preparation and uses
    ANSWER 46 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
AN
    1991:187671 CAPLUS
    114:187671
DN
    Entered STN: 17 May 1991
ED
    Curable acrylic resin compositions and coated products
TΙ
IN
    Ando, Hiroyuki; Kurahashi, Akihiko
PΑ
    Idemitsu Petrochemical Co., Ltd., Japan
SO
    Jpn. Kokai Tokkyo Koho, 10 pp.
    CODEN: JKXXAF
\mathsf{DT}
    Patent
LA
    Japanese
IC
    ICM C08F020-36
    ICS B32B027-30; C08G018-67; C09D004-00; C09D175-14
ICA C08F299-06
CC
     42-7 (Coatings, Inks, and Related Products)
FAN.CNT 1
     PATENT NO.
                                          APPLICATION NO.
                                                                  DATE
                        KIND
                               DATE
                                           ______
     _____
                        ____
                               _____
                                                                  -----
    JP 02284904
                               19901122
                                           JP 1989-107983
                        A2
                                                                  19890426
PI
PRAI JP 1989-107983
                               19890426
CLASS
             CLASS PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
 _____
                _____
                ICM
JP 02284904
                       C08F020-36
                ICS
                       B32B027-30; C08G018-67; C09D004-00; C09D175-14
                ICA
                       C08F299-06
                       C08F0020-36 [ICM,5]; C08F0020-00 [ICM,5,C*];
                IPCI
                       B32B0027-30 [ICS,5]; C08G0018-67 [ICS,5]; C08G0018-00
                       [ICS, 5, C*]; C09D0004-00 [ICS, 5]; C09D0175-14 [ICS, 5];
                       C08F0299-06 [ICA,5]; C08F0299-00 [ICA,5,C*]
    The title compns., useful in ink ribbons for magnetic recording,
AΒ
     thermal-transfer recording, etc., contain prepolymers of compds. containing
     ≥3 (meth)acryloyl groups and ≥1 active H, isocyanates, and
     optionally curable phosphazenes. Substrates are coated with the compns.
     and cured to prepare coated products with reduced tack and good chalking and
     scratching resistance. Thus, 50 parts 4:6 dipentaerythritol
     pentaacrylate-dipentaerythritol hexaacrylate mixture was treated with 4.5
     parts PhNCO and mixed with 50 parts hexakis(methacryloylethylenedioxy)cycl
    otriphosphazene to give title composition Then, a polyester sheet was coated with a mixture containing the composition 30, MIBK 30, BuOH 20, Me2CHOH 20, and
     1-hydroxycyclohexyl Ph ketone 1 g and irradiated by UV to form a tack-free
     coating with no chalking.
ST
     curable acrylic resin coating tackfree; photocurable coating acrylic
     polymer; phosphazene curable blend acrylic coating; scratch resistance
     coating acrylic polymer; chalking resistance coating acrylic polymer
ΙT
    Coating materials
        (curable, acrylic, with reduced tack and good chalking resistance)
ΙT
    Coating materials
        (photocurable, acrylic, with reduced tack and good chalking resistance)
```

(styrene-based resin-phosphazene polymer composites as)

103-71-9D, Phenyl isocyanate, reaction products with (di)pentaerythritol TΤ acrylates 3524-68-3D, Pentaerythritol triacrylate, reaction products 4074-88-8, Diethylene glycol diacrylate with isocyanates TDI, reaction products with (di)pentaerythritol acrylates 29570-58-9, Dipentaerythritol hexaacrylate 60506-81-2D, Dipentaerythritol pentaacrylate, reaction products with isocyanates 92832-53-6 RL: USES (Uses) (curable acrylic resin compns. containing, for tack-free coatings) L9 ANSWER 47 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN AN 1990:613404 CAPLUS DN 113:213404 ED Entered STN: 08 Dec 1990 TI Fiber-reinforced curable phosphazene molding materials and cured products IN Kurahashi, Akihiko Idemitsu Petrochemical Co., Ltd., Japan PΑ SO Jpn. Kokai Tokkyo Koho, 8 pp. CODEN: JKXXAF DTPatent LA Japanese ICM C08L085-02 IC ICS C08J005-24; C08K007-02; C08L101-00 C08L085-00 37-6 (Plastics Manufacture and Processing) Section cross-reference(s): 38 FAN.CNT 1 KIND DATE APPLICATION NO. PATENT NO. DATE ____ -----_____ A2 19900807 JP 1989-18526 JP 02199174 19890127 PRAI JP 1989-18526 19890127 CLASS PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES ______ JP 02199174 ICM C08L085-02 ICS C08J005-24; C08K007-02; C08L101-00 ICI C08L085-00 C08L0085-02 [ICM,5]; C08J0005-24 [ICS,5]; C08K0007-02 [ICS,5]; C08K0007-00 [ICS,5,C*]; C08L0101-00 [ICS,5]; C08L0085-00 [ICI,5] AΒ The title materials and products with good mech. strength, surface appearance, hardness, dimensional stability, chemical resistance, etc., comprise curable phosphazene compds. and fibrous materials. Thus, treating 116 g hexachlorocyclotriphosphazene with 400 g 2-hydroxyethyl methacrylate in pyridine-toluene mixture containing hydroquinone at 60° for 6 h gave 300 g 1,1,3,3,5,5-hexakis(methacryloyloxyethyloxy)cyclotripho sphazene (I). Then, 12 glass cloths were impregnated with I containing 3 parts Bz202, stacked, and hot pressed at 120° for 2 h to give a plate with tensile strength 29 kg/mm2, bending strength 35 kg/mm2, Rockwell hardness M 121, and good resistance to aqueous HCl, aqueous NaOH, Me2CO, and PhCl, vs. 28, 8.4, M 121, and good, resp., for the plate manufactured without glass cloth. glass fiber reinforced phosphazene molding; curable phosphazene molding ST mech strength; chem resistance curable phosphazene molding ΙT Carbon fibers, uses and miscellaneous RL: USES (Uses) (cloth, composites with phosphazene polymers, with good mech. strength and chemical resistance) ΙT Phosphazene polymers RL: PEP (Physical, engineering or chemical process); PROC (Process) (moldings, fiber-reinforced, with good mech. strength and chemical

(phosphazene polymers, fiber-reinforced, manufacture of)

resistance)

Chemically resistant materials

Polyamide fibers, uses and miscellaneous

ΙT

IT

```
(aramid, composites with phosphazene polymers, Kevlar 49, with good
       mech. strength and chemical resistance)
ΙT
    Glass fibers, uses and miscellaneous
    RL: USES (Uses)
       (textiles, composites with phosphazene polymers, with good mech.
       strength and chemical resistance)
IT
    7440-44-0
    RL: USES (Uses) .
       (carbon fibers, cloth, composites with phosphazene polymers, with good
       mech. strength and chemical resistance)
    24938-64-5, p-Phenylenediamine-terephthalic acid copolymer, sru
ΙT
    25035-37-4, p-Phenylenediamine-terephthalic acid copolymer
    RL: USES (Uses)
       (fibers, composites with phosphazene polymers, with good mech. strength
       and chemical resistance)
IT
    92832-53-6P
    RL: PEP (Physical, engineering or chemical process); PREP (Preparation);
    PROC (Process)
       (preparation and polymerization of)
    93891-06-6P
IT
    RL: PREP (Preparation)
       (preparation of, fiber-reinforced, with good mech. strength and chemical
       resistance)
IT
    868-77-9, 2-Hydroxyethyl methacrylate
    RL: RCT (Reactant); RACT (Reactant or reagent)
       (reaction of, with hexachlorocyclotriphosphazene)
ΙT
    940-71-6
    RL: RCT (Reactant); RACT (Reactant or reagent)
       (reaction of, with hydroxyethyl methacrylate)
    ANSWER 48 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
    1990:554438 CAPLUS
ΑN
    113:154438
DN
    Entered STN: 27 Oct 1990
ED
    Phosphazene compound compositions for curable coatings
TΙ
ΙN
    Yaquchi, Atsunori
    Idemitsu Kosan Co., Ltd., Japan
PA
    Jpn. Kokai Tokkyo Koho, 11 pp.
SO
    CODEN: JKXXAF
DΤ
    Patent
LA
    Japanese
IC
    ICM C08L085-02
    42-10 (Coatings, Inks, and Related Products)
CC
    Section cross-reference(s): 38
FAN.CNT 1
                      KIND DATE APPLICATION NO.
    PATENT NO.
                                                             DATE
                                         -----
                      ----
                       A2 19900517 JP 1988-283059
    JP 02129263
                                                               19881109
PRAI JP 1988-283059
                             19881109
CLASS
            CLASS PATENT FAMILY CLASSIFICATION CODES
PATENT NO.
 ______
JP 02129263
              ICM C08L085-02
               IPCI C08L0085-02 [ICM,5]; C08L0085-00 [ICM,5,C*]
    Title compns., useful for 10-1000 µm coating layers, comprise curable
AB
    phosphazene compound, monomer and/or prepolymer. A composition, prepared from
а
    mixture of A compound prepared from 0.167 mol hexachlorocyclotriphosphazene and
    1.1 mol 2-hydroxyethyl methacrylate (I) 20, I 20, MIBK 20, BuOH 10, PhMe
    20, 1-hydroxycyclohexylphenyl ketone 2 g, was coated onto polyester fibers
    and irradiated with UV to give a 70 µm layer having pencil hardness 9 H
    with good flexibility.
    hexachlorocyclotriphosphazene deriv copolymer coating flexible;
    hydroxyethyl methacrylate phosphazene copolymer coating flexible
```

RL: USES (Uses)

```
IT
     Coating materials
         (phosphazene derivative copolymers, hard and flexible)
     75-89-8DP, 2,2,2-Trifluoroethanol, reaction product with
 ΙT
     hexachlorocyclotriphosphazene and hydroxyethyl methacrylate 868-77-9DP,
      2-Hydroxyethyl methacrylate, reaction product with
     hexachlorocyclotriphosphazine 924-42-5DP, N-Methylol acrylamide,
      reaction product with hexachlorocyclotriphosphazene, hydroxyethyl
     methacrylate, and trifluoroethanol 940-71-6DP, reaction product with
     hydroxyethyl methacrylate 15625-89-5DP, Trimethylolpropane triacrylate,
      reaction product with hexachlorocyclotriphosphazene, hydroxyethyl
     methacrylate, and trifluoroethanol 92832-53-6P
     RL: RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)
         (preparation and polymerization of, for hard and flexible coating)
 ΙT
     129844-86-6P
                   129869-78-9P 129869-79-0P
      RL: PREP (Preparation)
         (preparation of, for hard and flexible coating)
 L9
     ANSWER 49 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
 AN
     1990:498757 CAPLUS
 DN
     113:98757
 ED
     Entered STN: 16 Sep 1990
 ΤI
     Phosphazene compositions as antioxidants for magnetic powders
 IN
     Yaguchi, Atsunori
 PΑ
     Idemitsu Petrochemical Co., Ltd., Japan
 SO
     Jpn. Kokai Tokkyo Koho, 9 pp.
     CODEN: JKXXAF
 DT
     Patent
 LA
     Japanese
 IC
     ICM C23F011-167
     ICS C01B021-072
 ICA B22F001-02
     37-6 (Plastics Manufacture and Processing)
     Section cross-reference(s): 55, 56, 57, 77
 FAN.CNT 1
     PATENT NO.
                  · KIND DATE
                                        APPLICATION NO.
     ______
                      ----
                                          -----
     JP 02047281.
                       A2 19900216 JP 1988-197172
                                                               19880809
 PRAI JP 1988-197172
                             19880809
 CLASS
  PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
  ______
  JP 02047281 ICM C23F011-167
                ICS C01B021-072
                ICA B22F001-02
                 IPCI C23F0011-167 [ICM,5]; C23F0011-10 [ICM,5,C*];
                       C01B0021-072 [ICS,5]; C01B0021-00 [ICS,5,C*];
                       B22F0001-02 [ICA,5]
 AΒ
     Title compns., i.e. (NP)-bearing crosslinkable groups and optional
      uncrosslinkable groups and having d.p. ≥3 are useful for
      formulation with powdered transition meals and/or their compds. for prolonged
      stability. Thus, 1,1,3,3,5,5-hexa(methacryloylethylenedioxy)cyclotriphosp
     hazene was prepared in a dispersion with powdered Fe for magnetic coating.
 ST
     magnetic coating stabilizer acrylic phosphazene; antioxidant acrylic
     phosphazene magnetic powder
 ΙT
     Antioxidants
        (for magnetic powders, preparation of acrylic phosphazene compns. as)
 ΙT
     Magnetic substances
        (transition metals, preparation of crosslinkable phosphazene compns. as
        antioxidants for)
      1309-37-1P, Iron(III) oxide, preparation 7439-89-6P, Iron, preparation
 ΙT
      24304-00-5P, Aluminum nitride
      RL: PREP (Preparation)
        (preparation of acrylic phosphazene compns. as antioxidants for)
 ΙT
      92832-53-6P
                 93891-06-6P
      RL: PREP (Preparation)
```

```
(preparation of, as stabilizers for magnetic powders)
ΙT
    940-71-6, Hexachlorocyclotriphosphazene
    RL: RCT (Reactant); RACT (Reactant or reagent)
       (reaction of, with hydroxyethyl methacrylate)
    ANSWER 50 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
    1990:236079 CAPLUS
AN
DN
    112:236079
    Entered STN: 23 Jun 1990
ED
ΤI
    Manufacture of hard transparent resins
ΙN
    Yaquchi, Atsunori
    Idemitsu Petrochemical Co., Ltd., Japan
PΑ
    Jpn. Kokai Tokkyo Koho, 7 pp.
    CODEN: JKXXAF
DT
    Patent
    Japanese
LA
IC
    ICM C08G079-02
    35-7 (Chemistry of Synthetic High Polymers)
FAN.CNT 1
    PATENT NO.
                      KIND
                           DATE
                                       APPLICATION NO.
                                                            DATE
    -----
                      ____
                            _____
                                       ______
                                                            _____
                                       JP 1988-146810
    JP 01315423
                      A2
                             19891220
                                                            19880616
    JP 06053803
                      B4
                             19940720
PRAI JP 1988-146810
                             19880616
CLASS
PATENT NO.
             CLASS PATENT FAMILY CLASSIFICATION CODES
 _____
              _____
               ICM
                     C08G079-02
JP 01315423
                     C08G0079-02 [ICM, 4]; C08G0079-00 [ICM, 4, C*]
               IPCI
GΙ
```

$$R^{1}S \longrightarrow \begin{bmatrix} 0 & R^{2} \\ || & | \\ C - C - N \\ || & R^{3} \end{bmatrix}$$

ΙT

940-71-6

RL: USES (Uses)

The resins, useful for optical materials, are manufactured by irradiating AΒ curable phosphazene compds. with actinic radiation in the presence of a morpholine derivative as photoinitiator. The morpholine derivative may be I (R1-R3 = H, C1-8 alkyl, C6-10 aryl, allyl, C2-8 alkenyl, C1-8hydroxyalkyl, C1-8 mercaptoalkoxyalkyl). Thus, a mixture of 30 g 1,1,3,3,5,5-hexakis(methacryloyloxyethoxy)cyclotriphosphazene and 0.9 g Irgacure 907 was UV-irradiated to give a product with pencil hardness ≥9H, vs. 2B for an injection-molded specimen from a com. polycarbonate. acrylic phosphazene polymer transparency hardness; morpholine photoinitiator acryloyl phosphazene polymn Transparent materials ΙT (acrylic phosphazene polymers, hard) Phosphazene polymers ΙT RL: USES (Uses) (acrylic, manufacture of, by radiation-induced polymerization) ΙT Polymerization catalysts (photochem., morpholine derivs., for acrylic phosphazenes) IT Polymerization (photochem., of hexakis(methacryloyloxyethoxy)cyclotriphosphazene) ΙT 868-77-9

(condensation of, with hexachlorocyclotriphosphazene)

(condensation of, with hydroxyethyl methacrylate)

```
IT
    71867-90-8
               71868-10-5, 2-Methyl-1-[4-(methylthio)phenyl]-2-morpholino-1-
    propanone 88324-57-6 88324-59-8 104856-52-2 127303-97-3
    127408-91-7 127408-92-8 127408-93-9
    RL: USES (Uses)
       (photoinitiator, for polymerization of acrylic phosphazenes)
    92832-53-6P
IΤ
    RL: PREP (Preparation)
       (preparation of)
    ANSWER 51 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
    1990:181607 CAPLUS
AN
DN
    112:181607
ED
    Entered STN: 12 May 1990
    Abrasion- and weather-resistant labels or signs
ΤI
    Yaguchi, Atsunori
IN
PA
    Idemitsu Petrochemical Co., Ltd., Japan
SO
    Jpn. Kokai Tokkyo Koho, 10 pp.
    CODEN: JKXXAF
DT
    Patent
    Japanese
LA
    ICM E01F009-00
IC
    ICS G09F007-00
ICA C09D003-49
    42-10 (Coatings, Inks, and Related Products)
    Section cross-reference(s): 43, 55
FAN.CNT 1
                                       APPLICATION NO. DATE
    PATENT NO.
                      KIND DATE
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                                         -----
                                                              _____
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                              19891013 JP 1988-84879
    JP 01256609
                       A2
                                                              19880406
PRAI JP 1988-84879
                              19880406
CLASS
 PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
               ____
 -----
 JP 01256609
               ICM E01F009-00
               ICS
                      G09F007-00
                ICA
                      C09D003-49
                      E01F0009-00 [ICM, 4]; G09F0007-00 [ICS, 4]; C09D0003-49
                IPCI
                      [ICA, 4]
    The title labels or signs are coated with cured phosphazenes. Spraying
AΒ
    plastic, steel, or plywood with an iso-BuCOMe solution of initiator and
    1,1,3,3,5,5-hexakis[(methacryloyloxy)ethoxy]cyclotriphosphazene and
    irradiating with UV gave products resisting abrasion, heat, acids,
    alkalies, hot water, and weathering.
    abrasion resistance coating cyclophosphazene; plastic coating abrasion
ST
    resistance; metal coating abrasion resistance; label coating abrasion
    resistance; wood coating abrasion resistance
ΙT
    Labels
    Signs
    Polycarbonates, uses and miscellaneous
    RL: USES (Uses)
        (abrasion resistant cyclotriphosphazene methacrylate coatings for)
    Acrylic polymers, uses and miscellaneous
TΤ
    Polyesters, uses and miscellaneous
    RL: USES (Uses)
        (abrasion-resistant cyclotriphosphazene methacrylate coatings for)
ΙT
    Coating materials
       (abrasion-resistant, photocurable, cyclophosphazene methacrylates, for
       labels or signs)
     9011-14-7, PMMA 25038-59-9, PET (polymer), uses and miscellaneous
ΙT
    RL: USES (Uses)
        (abrasion resistant cyclotriphosphazene methacrylate coatings for)
ΙT
    92832-53-6
     RL: TEM (Technical or engineered material use); USES (Uses)
        (coatings, abrasion-and weather-resistant, for signs or labels)
```

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L9
     ANSWER 52 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
AN
    1990:159133 CAPLUS
DN
    112:159133
     Entered STN: 28 Apr 1990
ED
     Manufacture of curable phosphazene compounds containing (meth)acrylate
TI
     Kitayama, Masahiro; Tsubokawa, Masaya; Yaguchi, Atsunori
IN
PA
     Idemitsu Petrochemical Co., Ltd., Japan
     Jpn. Kokai Tokkyo Koho, 4 pp.
SO
     CODEN: JKXXAF
\mathsf{DT}
     Patent
LA
     Japanese
     ICM C08G079-02
IC
     ICS C08G079-02
PATENT NO. KIND DATE APPLICATION NO. DATE

PI JP 01236241 A2 19890921 JP 1989-6050

PRAI JP 1988-60586

CLASS
                        A2 19890921 JP 1988-60586 19880316
 PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
 _____
 JP 01236241
               ICM C08G079-02
                ICS C08G079-02
                IPCI C08G0079-02 [ICM, 4]; C08G0079-02 [ICS, 4]; C08G0079-00
                       [ICS, 4, C*]
     Cl-free title compds., giving cured products with excellent transparency,
AΒ
     hardness, and heat resistance and useful for coatings, lenses, magnetic
     recording materials, etc., are prepared by the reaction of Cl-containing
     phosphazenes with polyol monoalkali metal salts and treatment of the
     resulting OH-containing phosphazene compds. with (meth)acrylic acid (or acid
     halide). A solution of hexachlorocyclotriphosphazene in toluene was added
     slowly to a solution of NaOCH2CH2OH in EtOH and refluxed to give 97%
     hexakis(2-hydroxyethoxy)cyclotriphosphazene (I) which (83.6 g) was treated
     with 116 g H2C:CMeCOCl in Et2O containing PhNMe2 at reflux for 2 h to give 129
     g Cl-free hexakis(2-methacryloyloxyethoxy)cyclotriphosphazene.
     phosphazene methacryloyloxyethyl cyclic prepn;
     methacryloyloxyethylphosphazene cyclic prepn; acryloyloxyethylphosphazene
     cyclic prepn; hydroxyethylphosphazene methacrylate cyclic prepn
IT
     Monomers
     Vinyl compounds, preparation
     RL: PREP (Preparation)
        ((methacryloyloxyethoxy) (methacryloyloxytrifluoroethoxy) phosphazenes,
        preparation of)
IT
     Phosphonitrile compounds
     RL: PREP (Preparation)
        (phosphazenes, (meth)acryloyloxyethoxy and
        (meth)acryloyloxytrifluoroethoxy derivs., preparation of polymerizable)
     920-46-7, Methacryloyl chloride
ΙT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (esterification of, with (hydroxyethyl)phosphazenes)
     96435-30-2P
IT
     RL: PREP (Preparation)
        (preparation and esterification with (meth)acryloyl chloride)
     75-89-8DP, 2,2,2-Trifluoroethanol, etherification products with
     hexachlorocyclotriphosphazene and ethanediol, methacrylate esters
     79-41-4DP, esters with (hydroxyethoxy) (hydroxytrifluoroethoxy) cyclotriphos
     phazenes 940-71-6DP, etherification products with trifluoroethanol and
     ethanediol, methacrylate esters 7388-28-5DP, Ethylene glycol monosodium
     salt, etherification products with hexachlorocyclotriphosphazene and
     trifluoroethanol, methacrylate esters 92832-53-6P
     RL: PREP (Preparation)
        (preparation of polymerizable)
```

```
L9
     ANSWER 53 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
     1990:158642 CAPLUS
 AN
 DN
     112:158642
 ED
     Entered STN: 28 Apr 1990
 TΙ
     Preparation of hardenable chlorine-free phosphazene compounds
 IN
     Kurahashi, Akihiko; Mori, Shigeo
 PΑ
     Idemitsu Petrochemical Co., Ltd., Japan
     Jpn. Kokai Tokkyo Koho, 6 pp.
 SO
     CODEN: JKXXAF
 DT
     Patent
 LA Japanese
 IC
     ICM C07F009-02
     ICS A61K006-08; C07F009-22
 ICA C08F030-02
     29-7 (Organometallic and Organometalloidal Compounds)
      Section cross-reference(s): 35
 FAN.CNT 1
                      KIND DATE APPLICATION NO.
                                                               DATE
     PATENT NO.
                        ----
      _____
     JP 01246292
                      A2 19891002 JP 1988-70886 19880326
 PI
 PRAI JP 1988-70886
                              19880326
 CLASS
  PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
  ______
  JP 01246292 ICM C07F009-02
                ICS A61K006-08; C07F009-22
                 ICA C08F030-02
                 IPCI C07F0009-02 [ICM, 4]; A61K0006-08 [ICS, 4]; A61K0006-02
                       [ICS, 4, C*]; C07F0009-22 [ICS, 4]; C07F0009-00
                       [ICS, 4, C*]; C08F0030-02 [ICA, 4]; C08F0030-00 [ICA, 4, C*]
 AΒ
     The title compds., containing practically no Cl, are prepared by treatment of
     phosphazene compds. having P-Cl bonds with organic halides having an OH or
      NH2 group, followed by treatment with (meth)acryloyl compds. C1CH2CH2OH
      (254 g) was added dropwise to a solution of 174 g
     hexachlorocyclotriphosphazene in pyridine/1,4-dioxane and the reaction
     mixture was stirred at 60° for 6 h to give 288 g hexa(2-
      chloroethoxy)cyclotriphosphazene (I). A dioxane solution of 102 g I was
      added dropwise to a mixture of 137 g CH2: CMeCO2K, Bu4NBr, and dioxane and
      the reaction mixture was stirred at 80° for 4 h to give 141 g
     hexa(2-methacryloyloxyethoxy)cyclotriphosphazene.
 ST
     phosphazene compd prepn monomer; chlorocyclotriphosphazene condensation
      ethylene chlorohydrin; chloroethanol condensation
      chlorocyclotriphosphazene; chloroethoxycyclotriphosphazene acyloxylation
     methacrylate; cyclotriphosphazene methacryloyloxyethoxy prepn monomer
 ΙT
     Phosphazene polymers
      RL: RCT (Reactant); RACT (Reactant or reagent)
         ((meth)acryloyloxy-containing phosphazenes as monomers for)
      940-71-6, Hexachlorocyclotriphosphazene
IT
      RL: RCT (Reactant); RACT (Reactant or reagent)
         (condensation of, with ethylene chlorohydrin,
         (chloroethoxy)cyclotriphosphazene from)
      107-07-3, Ethylene chlorohydrin, reactions
 ΙT
      RL: RCT (Reactant); RACT (Reactant or reagent)
         (condensation of, with hexachlorocyclotriphosphazene,
         (chloroethoxy)cyclotriphosphazene from)
      79-41-4, Methacrylic acid, reactions 6900-35-2, Potassium methacrylate
 ΙT
      RL: RCT (Reactant); RACT (Reactant or reagent)
         (esterification with, of (chloroethoxy)cyclotriphosphazene,
         (methacryloyloxyethoxy)cyclotriphosphazene from)
 ΙT
      29364-58-7P
      RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
      (Reactant or reagent)
         (preparation and esterification of, with methacrylic acid (salt),
         (methacryloyloxyethoxy)cyclotriphosphazene from)
```

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ΙT
    92832-53-6P
    RL: SPN (Synthetic preparation); PREP (Preparation)
       (preparation of, as monomer)
    ANSWER 54 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
    1990:120262 CAPLUS
AN
DN
    112:120262
    Entered STN: 31 Mar 1990
ED
    Phosphazene-containing polymer composites for printed circuit boards
TI
IN
    Mori, Shigeo; Yaguchi, Atsunori
    Idemitsu Petrochemical Co., Ltd., Japan
PΑ
SO
    Jpn. Kokai Tokkyo Koho, 8 pp.
    CODEN: JKXXAF
DT
    Patent
LA
    Japanese
IC
    ICM H05K001-03
ICA B32B015-08; C08G079-02; C08J005-24
    38-3 (Plastics Fabrication and Uses)
    Section cross-reference(s): 76
FAN.CNT 1
                     KIND DATE APPLICATION NO.
    PATENT NO.
                                                            DATE
                      ____
    -----
                      A2 19890821 JP 1988-33375 19880216
    JP 01207986
PRAI JP 1988-33375
                            19880216
 PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
 _____
JP 01207986
              ICM H05K001-03
               ICA B32B015-08; C08G079-02; C08J005-24
               IPCI H05K0001-03 [ICM, 4]; B32B0015-08 [ICA, 4]; C08G0079-02
                      [ICA, 4]; C08G0079-00 [ICA, 4, C*]; C08J0005-24 [ICA, 4]
AB
    Strong but flexible title boards are manufactured by coating and/or
    impregnating substrates with curable phosphazene compds. (NPR1R2)n,
    (≥1 of R1-2 = polymerizable group; n \ge 3), then curing.
    Thus, hexachlorocyclotriphosphazene and 2-hydroxyethyl methacrylate were
    dissolved in benzene containing hydroquinone and pyridine and heated to give
    1,1,3,3,5,5-hexa(methacryloyloxyethoxy)cyclotriphosphazene, which was
    dissolved in MEK with benzophenone (photoinitiator). Kraft paper 0.207 mm
    thick was impregnated with the solution, dried, and irradiated with UV lamps
    on both sides to give a composite showing tensile strength 21.7 kg/15 mm,
    vs. 4.7 for the untreated paper.
    crosslinked phosphazene polymer circuit board;
    methacryloyloxyethoxycyclotriphosphazene composite printed circuit board;
    printed circuit board vinylphosphazene polymer; paper curable phosphazene
    polymer impregnated; tensile strength printed circuit board;
    polyphosphazene crosslinked reinforced circuit board
    Phosphazene polymers
    RL: USES (Uses)
       (crosslinked, reinforced, for strong flexible printed circuit boards)
    Polyamide fibers, uses and miscellaneous
IT
    RL: USES (Uses)
       (polymerized phosphazene compds. reinforced with, for printed circuit
      · boards)
ΙT
    Paper
       (ceramic, alumina, polymerized phosphazene compds. reinforced with, for
       printed circuit boards)
IT
       (kraft, polymerized phosphazene compds. reinforced with, for printed
       circuit boards)
ΙT
    Polymers, uses and miscellaneous
    RL: USES (Uses)
       (phosphazene group-containing, crosslinked, reinforced, for strong flexible
       printed circuit boards)
ΙT
    Electric circuits
        (printed, boards, flexible, reinforced polymers of phosphazene
```

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compds,., with high strength)
    Glass fibers, uses and miscellaneous
ΙT
    RL: USES (Uses)
       (textiles, polymerized phosphazene compds. reinforced with, for printed
       circuit boards)
ΙT
    868-77-9, 2-Hydroxyethyl methacrylate
    RL: RCT (Reactant); RACT (Reactant or reagent)
       (dehydrochlorination of, with hexachlorocyclotriphosphazene)
    940-71-6, Hexachlorocyclotriphosphazene
ΙT
    RL: RCT (Reactant); RACT (Reactant or reagent)
       (dehydrochlorination of, with hydroxyethyl methacrylate)
    1344-28-1, Aluminum oxide (Al2O3), uses and miscellaneous
ΙT
    RL: USES (Uses) .
       (paper, polymerized phosphazene compds. reinforced with, for printed
       circuit boards)
ΙT
    92832-53-6P
    RL: RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)
       (preparation and polymerization of)
ΙT
    93891-06-6P
    RL: PREP (Preparation)
       (preparation of, reinforced, for strong flexible printed circuit boards)
    ANSWER 55 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
AN
    1990:45697 CAPLUS
DN
    112:45697
    Entered STN: 04 Feb 1990
ED
    Durable patterning member useful in printed circuit fabrication
ΤI
IN
    Mori, Shigeo; Yaguchi, Atsunori
PA
    Idemitsu Petrochemical Co., Ltd., Japan
SO
    PCT Int. Appl., 39 pp.
    CODEN: PIXXD2
DT
    Patent
LA
    Japanese
IC
    ICM G03F001-00
    ICS H01L021-30
CC
    74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other
    Reprographic Processes)
    Section cross-reference(s): 76
FAN.CNT 1
                      KIND DATE APPLICATION NO.
    PATENT NO.
                                                              DATE
                       ----
    _____
                                         _____
                                                                _____
                    · Al 19890223 WO 1988-JP546
PΙ
    WO 8901650
                                                                19880606
        W: KR, US
        RW: AT, BE, CH, DE, FR, GB, IT, LU, NL, SE
                    A2 19890228 JP 1987-199636
    JP 01052155
                                                                19870810
    JP 07007207 B4 19950130
EP 328648 A1 19890823 EP 1988-904670
                                                                19880606
        R: BE, CH, DE, FR, GB, IT, LI, NL, SE
    US 5051295 A 19910924
JP 1987-199636 A 19870810
JP 1987-119926 A1 19870516
WO 1988-JP546 W 19880606
                                         US 1989-360935
                                                                19890317
PRAI JP 1987-199636
CLASS
             CLASS PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
 -----
               _____
                      G03F001-00
 WO 8901650
                ICM
                ICS
                      H01L021-30
                       G03F0001-00 [ICM, 4]; H01L0021-30 [ICS, 4]; H01L0021-02
                IPCI
                       G03F0001-14 [I,A]; G03F0001-14 [I,C*]
                IPCR
 JP 01052155
                       G03F0001-00 [ICM,4]; H01L0021-30 [ICS,4]; H01L0021-02
                IPCI
                       [ICS, 4, C*]; H05K0003-00 [ICS, 4]
                ECLA
                       G03F001/00
 EP 328648
                IPCI
                       G03F0001-00 [ICM,4]; H01L0021-30 [ICS,4]; H01L0021-02
                       [ICS, 4, C*]
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IPCR
                         G03F0001-14 [I,A]; G03F0001-14 [I,C*]
 US 5051295
                  IPCI
                         B32B0009-04 [ICM, 5]; H01L0021-00 [ICS, 5]
                  IPCR
                         G03F0001-14 [I,A]; G03F0001-14 [I,C*]
                  NCL
                         428/195.100; 428/064.100; 428/076.000; 428/201.000;
                         428/203.000; 428/204.000; 428/209.000; 428/411.100;
                         428/520.000; 428/704.000; 428/835.600; 428/901.000; 428/913.000; 430/003.000; 430/004.000; 430/005.000;
                         430/006.000; 430/007.000; 430/320.000; 430/321.000
AΒ
     The invention provides a durable patterning member (photomask, lith film)
     having a protective film consisting of a cured product of a curable compound
     having excellent mech., optical and chemical properties and the member is
     used for patterning in the fabrication of ICs, printed circuits, hybrid
     ICs, etc.
     photomask protective film; lith film protective layer; protective film
ST
     photomask
ΤT
     Semiconductor devices
         (photomasks for fabrication of)
ΙT
     Photomasks
         (protective films for)
ΙT
     Electric circuits
         (integrated, photomasks for fabrication of)
ΙT
     92832-53-6
                  124365-46-4, Tosquard 520 124365-49-7, Uni-DIC
     17-827
     RL: USES (Uses)
         (protective coating, for photomasks)
L9
     ANSWER 56 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
     1989:633778 CAPLUS
AN
DN
     111:233778
     Entered STN: 23 Dec 1989
ED
     Heat-resistant polymers prepared from [(4'-(2-vinyl)-4-
TΙ
     biphenylyl)oxy]pentachlorocyclotriphosphazene
ΑU
     Inoue, Kenzo; Nakamura, Hajime; Ariyoshi, Satoru; Takagi, Masataka;
     Tanigaki, Teiichi
CS
     Fac. Eng., Ehime Univ., Matsuyama, 790, Japan
     Macromolecules (1989), 22(12), 4466-9
SO
     CODEN: MAMOBX; ISSN: 0024-9297
\mathsf{DT}
     Journal
     English
LA
CC
     35-4 (Chemistry of Synthetic High Polymers)
     The radical-initiated copolymn. of [(4-(2-vinyl)-4-
AΒ
     biphenylyl)oxy]pentachlorocyclotriphosphazene (I) with styrene or Me
     methacrylate in 1,2-dichloroethane and the thermal behavior of their
     copolymers were investigated. The copolymers were enriched in I with
     respect to the monomer feed. The comparison of Alfrey-Price parameters of
     I with those of 4-hydroxy-4'-vinylbiphenyl suggested that the phosphazene
     ring in I acts only as an electron-withdrawing group. The
     thermogravimetric anal. indicated that incorporation of I in the
     copolymers leads to increased thermal stability. The copolymers containing
     >50 mol % I afforded 56-65% char yields at 800^{\circ} in air or N.
.ST
     vinylbiphenylyloxypentachlorocyclotriphosphazene copolymer
IT
     Heat-resistant materials
         ([((vinyl)biphenylyl)oxy]pentachlorocyclotriphosphazene copolymers,
        preparation of)
     Glass temperature and transition
ΙT
         (of [((vinyl)biphenylyl)oxy]pentachlorocyclotriphosphazene copolymers)
ΙT
     Q-e value in polymerization
     Reactivity ratio in polymerization
         (of [((vinyl)biphenylyl)oxy]pentachlorocyclotriphosphazene with styrene
        or Me methacrylate)
ΙT
     115529-67-4
     RL: RCT (Reactant); RACT (Reactant or reagent)
         (polymerization of, with Me methacrylate or styrene, reactivity ratio and
Q-e
        values in)
```

80-62-6, Methyl methacrylate 100-42-5, Styrene, reactions IΤ RL: RCT (Reactant); RACT (Reactant or reagent) (polymerization of, with [((vinyl)biphenylyl)oxy]pentachlorocyclotriphosphazen e, reactivity ratio and Q-e values in) 122847-48-7P 122847-49-8P ΙT RL: SPN (Synthetic preparation); PREP (Preparation) (preparation and heat resistance of) L9 ANSWER 57 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN ΑN 1989:596099 CAPLUS DN 111:196099 ED Entered STN: 25 Nov 1989 ΤI Cyclic polyphosphazene group-containing crosslinked acrylic plastics for IN Kitayama, Masahiro; Mori, Shigeo PA Idemitsu Petrochemical Co., Ltd., Japan SO Jpn. Kokai Tokkyo Koho, 9 pp. CODEN: JKXXAF DTPatent Japanese LA IC ICM C08G079-02 ICS C08G079-02 CC 37-6 (Plastics Manufacture and Processing) Section cross-reference(s): 38 FAN.CNT 1 APPLICATION NO. PATENT NO. KIND DATE --------------JP 01074230 A2 19890320 JP 1987-230951 19870917 PRAI JP 1987-230951 19870917 PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES JP 01074230 ICM C08G079-02

GΙ

$$Q = -M - R^{1} R^{2}$$

$$R^{5} R^{4}$$

ICS IPCI

[ICS, 4, C*]

AB Hard, lightwt., solvent-resistant title plastics with high refractive index (n) are manufactured by curing compns. of (preferably cyclic) phosphazenes having ≥3 repeating units NPAaBb [A = polymerizable/crosslinkable group; B = Q; M = O, S, NH; R1-5 = H, halogen, C1-4 (halo)alkyl; a + b = 2], optionally other monomers, and inorg. particles including some ≤200 mµm in size, with n ≥1.6.

Thus, 290 g hexachlorocyclotriphosphazene was treated with 716 g 2-hydroxyethyl methacrylate in C6H6 in the presence of pyridine to give 640 g [NP(OC2H4O2CCMeCH2)2]3 with n 1.50, 90 g of which was mixed with 10 g TiO2 (particle size 8-15 mµm) and 0.27 g 2,2-diethoxyacetophenone, degassed, then cured with UV irradiation to give a transparent plastic with good resistance to Me2CO, MeOH, and C6H6, n 1.60, transparency 90.2%, pencil hardness 6H, and heat distortion temperature 170-190°, vs. poor solvent resistance, 1.78, 3H, and 75-100° for poly(Me methacrylate) containing TiO2.

C08G0079-02 [ICM, 4]; C08G0079-02 [ICS, 4]; C08G0079-00

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ST
     phosphazene contg acrylic plastic lens; glass substitute polyphosphazene
     crosslinked plastic; solvent resistant polyphosphazene crosslinked
     plastic; scratch resistant polyphosphazene crosslinked plastic
IT
     Lenses
        (cyclic phosphazene group-containing acrylic polymers for, containing inorg.
        particles, with high refractive index and hardness)
ΙT
     Heat-resistant materials
        (abrasion- and chemical resistant, transparent, cyclic phosphazene
        group-containing acrylic polymers, containing inorg. particles, with high
        refractive index)
IT
     Transparent materials
        (abrasion- and chemical- and heat-resistant, cyclic phosphazene
        group-containing acrylic polymers, containing inorg, particles, with high
        refractive index)
TT
     Chemically resistant materials
        (abrasion- and heat-resistant, transparent, cyclic phosphazene
        group-containing acrylic polymers, containing inorg. particles, with high
        refractive index)
IT
     Abrasion-resistant materials
        (chemical- and heat-resistant, transparent, cyclic phosphazene
        group-containing acrylic polymers, containing inorg. particles, with high
        refractive index)
IT
     Polymers, preparation
     RL: PREP (Preparation)
        (phosphazene group-containing, manufacture of, containing inorg. particles,
with
        high refractive index, for lenses)
IT
     Phosphazene polymers
     RL: USES (Uses)
        (unsatd., vinyl-crosslinked, containing inorg. particles, with high
        refractive index, for lenses)
ΙT
     13463-67-7P, Titanium oxide (TiO2), preparation
                                                      1312-43-2, Indium oxide
              1314-36-9, Yttrium oxide, uses and miscellaneous
     (In2O3)
     RL: PREP (Preparation)
        (cyclic triphosphazene group-containing acrylic polymers containing, with
high
        refractive index)
     93891-06-6P
                  123416-41-1P
                                  123416-42-2P
IΤ
     RL: PREP (Preparation)
        (manufacture of, containing inorg. particles with high refractive index, for
        lenses)
TΤ
     92832-53-6P
                   123416-30-8P
                                  123416-31-9P
     RL: PEP (Physical, engineering or chemical process); PREP (Preparation);
     PROC (Process)
        (preparation and polymerization of)
IT
     868-77-9
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with hexachlorocyclotriphosphazene)
     7003-65-8
                 35535-81-0
ΙT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with hexachlorocyclotriphosphazene and hydroxyethyl
        methacrylate)
     940-71-6, Hexachlorocyclotriphosphazene
TΤ
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with hydroxyethyl methacrylate)
     ANSWER 58 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
     1989:596098 CAPLUS
ΑN
DN
     111:196098
     Entered STN: 25 Nov 1989
ED
ΤI
     Polymerizable phosphazene compound compositions for lenses
     Kitayama, Masahiro; Mori, Shigeo
IN
PA
     Idemitsu Petrochemical Co., Ltd., Japan
SO
     Jpn. Kokai Tokkyo Koho, 9 pp.
```

CODEN: JKXXAF

DT Patent

LA Japanese

ICM C08G079-02 IC

ICS C08G079-02

37-6 (Plastics Manufacture and Processing) CC

Section cross-reference(s): 38

FAN.CNT 1

| PATENT NO. | KIND | DATE | APPLICATION NO. | DATE |
|---------------------------------------|------|----------------------|-----------------|----------|
| PI JP 01074229
PRAI JP 1987-230950 | A2 | 19890320
19870917 | JP 1987-230950 | 19870917 |
| OT 7 OO | | | | |

| CLASS PATENT NO. | CLASS | PATENT FAMILY CLASSIFICATION CODES |
|------------------|-------|------------------------------------|
| JP 01074229 | | C08G079-02
C08G079-02 |

C08G0079-02 [ICM, 4]; C08G0079-02 [ICS, 4]; C08G0079-00 IPCI

[ICS, 4, C*]

GΙ

$$R^1$$
 R^2 R^3 R^5 R^4 I

Molding compns. which can be cured with radical initiators to form hard, AΒ heat- and solvent-resistant transparent plastics with high refractive index (n) useful as lightwt. glass substitutes, comprise (preferably cyclic) polyphosphazenes having ≥3 repeating units NPAaBb [A = polymerizable/crosslinkable group; B = Q; R1-5 = H, halogen, C1-4 (halo)alkyl; M = O, S, NH; a + b = 2) and inorg. particles including some \leq 200 mµm in size, with n \geq 1.6. Thus, 290 g hexachlorocyclotriphosphazene and 716 g 2-hydroxyethyl methacrylate were combined in C6H6 containing pyridine to give 640 g [NP(OC2H4O2CCMeCH2)2]3 with n 1.50, 90 g of which was mixed with 10 g TiO2 (particle size $8\text{--}15~\text{m}\mu\text{m}$, n 2.5). The composition was mixed with 0.27 g 2,2-diethoxyacetophenone, degassed, then cured by UV irradiation to give a transparent plastic with good resistance to Me2CO, MeOH, and C6H6, n 1.60, transparency 90.2%, pencil hardness 6H, and heat distortion temperature 170-190°, vs. poor solvent resistance, 1.77, 87.2%, 3H, and 75-100° for poly(Me methacrylate) containing TiO2.

phosphazene compd unsatd polymerizable molding; lens crosslinked STpolyphosphazene molding compn; scratch resistant polyphosphazene molding compn; solvent resistant polyphosphazene molding compn

ΙT

high

(phosphazene group-containing polymers, containing inorg. particles, with

refractive index, molding compns. for)

IT Heat-resistant materials

> (abrasion- and chemical resistant, transparent, phosphazene group-containing polymers, containing inorg. particles, with high refractive index, molding compns. for)

ΙT Transparent materials

(abrasion- and chemical- and heat-resistant, phosphazene group-containing polymers, containing inorg. particles, with high refractive index, molding compns. for)

ΙT Chemically resistant materials

(abrasion- and heat-resistant, transparent, phosphazene group-containing polymers, containing inorg. particles, with high refractive index, molding

```
compns. for)
ΙT
    Abrasion-resistant materials
       (chemical- and heat-resistant, transparent, phosphazene group-containing
       polymers, containing inorg. particles, with high refractive index, molding
       compns. for)
IT
    Polymers, preparation
    RL: PREP (Preparation)
       (phosphazene group-containing, manufacture of, containing inorg. particles,
for
       lenses, molding compns. for)
ΙT
    Phosphazene polymers
    RL: USES (Uses)
       (unsatd., crosslinked, containing inorg. particles, for lenses, molding
       compns. for)
IT
    1312-43-2, Indium oxide (In2O3)
                                    1314-36-9, Yttrium oxide, uses and
    miscellaneous 13463-67-7, Titanium oxide (TiO2), uses and miscellaneous
    RL: USES (Uses)
       (cyclotriphosphazene-containing crosslinked acrylic polymers containing, for
       lenses, molding compns. for)
                123416-41-1P 123416-42-2P
IT
    93891-06-6P
    RL: PREP (Preparation)
       (manufacture of, containing inorg, particles, for lenses, molding compns.
for)
    92832-53-6P 123416-30-8P
                                123416-31-9P
    RL: PEP (Physical, engineering or chemical process); PREP (Preparation);
    PROC (Process)
       (preparation and polymerization of)
ΙT
    868-77-9
             7003-65-8 35535-81-0
    RL: RCT (Reactant); RACT (Reactant or reagent)
       (reaction of, with hexachlorocyclotriphosphazene)
ΙT
    940-71-6, Hexachlorocyclotriphosphazene
    RL: RCT (Reactant); RACT (Reactant or reagent)
       (reaction of, with hydroxyethyl methacrylate)
L9
    ANSWER 59 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
ΑN
    1989:574827 CAPLUS
    111:174827
DN
    Entered STN: 10 Nov 1989
ED
    Manufacture of polymerizable phosphazenes
TΙ
    Mori, Shigeo; Kitayama, Masahiro
IN
PA
    Idemitsu Petrochemical Co., Ltd., Japan
    Jpn. Kokai Tokkyo Koho, 6 pp.
SO
    CODEN: JKXXAF
    Patent
DT
    Japanese
LA
    ICM C08G079-02
IC
    ICS C08G079-02
    35-2 (Chemistry of Synthetic High Polymers)
    Section cross-reference(s): 28
FAN.CNT 1
                             DATE APPLICATION NO.
    PATENT NO.
                       KIND
                                                              DATE
                                         -----
                                                               _____
    -----
                       ----
                       A2
                              19890331 JP 1987-243003
                                                               19870928
    JP 01087636
PRAI JP 1987-243003
                              19870928
CLASS
             CLASS PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
 _____
               _____
 JP 01087636
              ICM
                      C08G079-02
                ICS
                      C08G079-02
                      C08G0079-02 [ICM, 4]; C08G0079-02 [ICS, 4]; C08G0079-00
                IPCI
                      [ICS, 4, C^*]
AB
    Phosphazenes giving cured products with good heat resistance and mech.
    properties are manufactured by treating chlorophosphazene polymers unsatd.
    alcs. and, optionally, unpolymerizable alcs. in the presence of
```

bicycloheterocyclic bases as dehydrochlorinating agents. A solution of 477 g

```
193 g hexachlorocyclotriphosphazene and stirred at 60° for 8 h to
    give hexakis[(methacryloyloxy)ethyl]cyclotriphosphazene.
    phosphazene monomer prepn; methacrylate cyclotriphosphazene deriv;
ST
    hydroxyethyl methacrylate reaction chlorophosphazene;
    hexachlorocyclotriphosphazene reaction alc
ΙT
    Substitution reaction catalysts
       (bicycloheterocyclic amines, for chlorophosphazenes by unsatd. alcs.)
    6674-22-2, DBU
ΙT
    RL: CAT (Catalyst use); USES (Uses)
       (catalyst, for etherification of chlorophosphazenes by unsatd. alcs.)
    75-89-8DP, 2,2,2-Trifluoroethanol, reaction products with hydroxyethyl
ΙT
    methacrylate and hexachlorocyclotriphosphazene 7251-15-2P
    92832-53-6P
    RL: PREP (Preparation)
       (preparation of)
ΙT
    868-77-9, 2-Hydroxyethyl methacrylate
    RL: RCT (Reactant); RACT (Reactant or reagent)
       (reaction of, with cyclotriphosphazenes)
    107-18-6, Allyl alcohol, reactions
ΙT
    RL: RCT (Reactant); RACT (Reactant or reagent)
       (reaction of, with hexachlorocyclotriphosphazene)
    940-71-6, Hexachlorocyclotriphosphazene
ΙT
    RL: RCT (Reactant); RACT (Reactant or reagent)
       (reaction of, with unsatd. alcs.)
    ANSWER 60 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
    1989:574826 CAPLUS
AN
DN
    111:174826
ED
    Entered STN: 10 Nov 1989
    Manufacture of polymerizable phosphazenes
ΤI
IN
    Mori, Shigeo; Kitayama, Masahiro
PA
    Idemitsu Petrochemical Co., Ltd., Japan
SO
    Jpn. Kokai Tokkyo Koho, 7 pp.
    CODEN: JKXXAF
DT
    Patent
LA
    Japanese
    ICM C08G079-02
IC
    ICS C08G079-02
    35-2 (Chemistry of Synthetic High Polymers)
    Section cross-reference(s): 28
FAN.CNT 1
                     KIND DATE APPLICATION NO. DATE
    PATENT NO.
                                         -----
                                                              -----
    _____
                      ____
    JP 01087635
                       A2 19890331 JP 1987-243001
                                                              19870928
PRAI JP 1987-243001
                             19870928
 PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
 ______
              ICM C08G079-02
 JP 01087635
                ICS C08G079-02
                IPCI C08G0079-02 [ICM, 4]; C08G0079-02 [ICS, 4]; C08G0079-00
                      [ICS, 4, C*]
AB
    Phosphazenes giving cured products with good heat resistance and mech.
    properties are manufactured by treating chlorophosphazene polymers with unsatd.
    alcs. and, optionally, unpolymerizable alcs. in the presence of metal
    halides. A solution of 477 g 2-hydroxyethyl methacrylate and 20.4 g ZnCl2 in
    DMF was treated dropwise with 193 g hexachlorocyclotriphosphazene and
    stirred at 60° for 8 h to give hexakis[(methacryloyloxy)ethoxy]cycl
    otriphosphazene.
ST
    phosphazene monomer prepn; methacrylate cyclotriphosphazene deriv;
    hydroxyethyl methacrylate reaction chlorophosphazene;
    hexachlorocyclotriphosphazene reaction alc unsatd; catalyst substitution
    chlorophosphazene; zinc chloride catalyst substitution
IT
    Chlorides, uses and miscellaneous
```

2-hydroxyethyl methacrylate and 559 g DBU in DMF was treated dropwise with

```
RL: CAT (Catalyst use); USES (Uses)
        (catalyst, for reaction of chlorophosphazenes with unsatd. alcs.)
     Substitution reaction catalysts
ΙT
        (metal halides, for chlorophosphazenes with unsatd. alcs.)
     1344-13-4, Tin chloride (unspecified) 7646-85-7, Zinc chloride, uses and
ΙT
     miscellaneous 7758-89-6, Cuprous chloride
     RL: CAT (Catalyst use); USES (Uses)
        (catalyst, for reaction of chlorophosphazenes with unsatd. alcs.)
     75-89-8DP, 2,2,2-Trifluoroethanol, reaction products with hydroxyethyl
ΙT
     methacrylate and hexachlorocyclotriphosphazene 7251-15-2P
     92832-53-6P
     RL: PREP (Preparation)
        (preparation of, catalysts for)
     107-18-6, Allyl alcohol, reactions 868-77-9, 2-Hydroxyethyl methacrylate
ΙT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with hexachlorocyclotriphosphazene, catalysts for)
     940-71-6
TT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with unsatd. alcs., catalysts for)
     ANSWER 61 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
     1989:478839 CAPLUS
ΑN
DN
     111:78839
ED
     Entered STN: 03 Sep 1989
     Polymers of aryl vinyl phosphazenes for optical equipment
ΤI
IN
     Mori, Shigeo; Kitayama, Masahiro
PA
     Idemitsu Petrochemical Co., Ltd., Japan
SO
     Jpn. Kokai Tokkyo Koho, 8 pp.
     CODEN: JKXXAF
DT
     Patent
LA
     Japanese
     ICM C08G079-02
IC
     35-4 (Chemistry of Synthetic High Polymers)
CC
FAN.CNT 1
                                          APPLICATION NO.
     PATENT NO.
                        KIND
                               DATE
                                                                DATE
                                          _____
                                                                -----,
                              _____
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                        ____
     JP 01014240
                        A2
                              19890118
                                          JP 1987-168381
                                                                19870706
PΙ
PRAI JP 1987-168381
                              19870706
CLASS
 PATENT NO.
             CLASS PATENT FAMILY CLASSIFICATION CODES
                _____
 ______
              ICM
                       C08G079-02
 JP 01014240
                       C08G0079-02 [ICM, 4]; C08G0079-00 [ICM, 4, C*]
                IPCI
GI
```

AB Hard, transparent title polymers with high refractive index, useful for lenses, etc., are prepared from (preferably cyclic) phosphazenes having ≥3 repeating units NPAB (A = polymerizable group; B = aromatic group Q; R1-5 = H, Cl, Br, C1-4 haloalkyl; M = O, S, imino), and optionally other monomers. Thus, 121.2 g thiophenol was treated with 23 g Na, then mixed with 115.9 g hexachlorocyclotriphosphazene and 143 g 2-hydroxyethyl methacrylate in the presence of pyridine and hydroquinone at 50°

```
for 27 h to give a product, which was mixed with 0.1% benzoin iso-Pr
    ether, poured into a mold, and exposed to UV radiation to give a polymer
    plate having light transmission 90%, n 1.62, and pencil hardness 4H.
    phosphazene aryl vinyl deriv polymer; acrylic polymer phosphazene contg
ST
    transparent; cyclic phosphazene contg vinyl polymer; cyclotriphosphazene
    contg vinyl polymer lens; glass substitute polyvinyl aryl phosphazene;
    refractive index polyvinyl aryl phosphazene
TΤ
    Transparent materials
       (phosphazene-containing vinyl polymers, with high refractive index and
       hardness)
IT
    Phosphonitrile compounds
    RL: RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)
       (cyclic, phosphazenes, preparation and polymerization of, for optical
applications)
    Polymers, preparation
    RL: PREP (Preparation)
       (phosphazene group-containing, preparation of, with high refractive index
and
       hardness, for optical apparatus)
    940-71-6, Hexachlorocyclotriphosphazene
IT
    RL: USES (Uses)
       (condensation of, with (thio)phenolates and unsatd. alcs. or amines)
    868-77-9
ΙT
    RL: USES (Uses)
       (condensation of, with hexachlorocyclotriphosphazene and
       (thio)phenolates)
    930-69-8, Sodium thiophenolate 2666-53-7
TΤ
    RL: USES (Uses)
       (condensation of, with hexachlorocyclotriphosphazene and hydroxyethyl
       methacrylate)
    92832-53-6P
                 122108-26-3P
                               122108-27-4P
IT
    RL: RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)
       (preparation and polymerization of)
ΙT
    93891-06-6P
                122108-70-7P
                               122141-87-1P
    RL: PREP (Preparation)
       (preparation of, transparent, for lenses)
    ANSWER 62 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
    1989:441464 CAPLUS
ΑN
DN
    111:41464
    Entered STN: 05 Aug 1989
F.D
    Curable phosphazene polymers for coatings with good abrasion, alkali, heat
TΙ
    and weather resistance
    Mori, Shigeo; Yaguchi, Atsunori; Kitayama, Masahiro
IN
    Idemitsu Petrochemical Co., Ltd., Japan
PΑ
SO
    Jpn. Kokai Tokkyo Koho, 16 pp.
    CODEN: JKXXAF
DT
    Patent
LA
    Japanese
IC
    ICM C09D003-49
ICA C08G079-02
CC
    42-10 (Coatings, Inks, and Related Products) ·
FAN.CNT 1
                                         APPLICATION NO. DATE
    PATENT NO.
                      KIND
                              DATE
                      ----
                              _____
                                          ______
                                                                _____
    -----
    JP 63241075
                             19881006
                       A2
                                        JP 1987-152772
                                                               19870618
PΙ
PRAI JP 1986-282784
                       A1
                              19861126
CLASS
             CLASS PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
 -----
               ____
 JP 63241075
                ICM
                      C09D003-49
                ICA
                      C08G079-02
                IPCI
                      C09D0003-49 [ICM, 4]; C08G0079-02 [ICA, 4]; C08G0079-00
                       [ICA, 4, C*]
                      C08G0079-00 [I,C*]; C08G0079-02 [I,A]
                IPCR
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The coatings contain phosphazene compds. (NPRlaR2b)n (R1 = double
ΑB
    bond-containing group; R2 = double bond-containing group, residues of OH- or
    NH2-containing compds.; a > 0; b \ge 0; a + b = 2; n \ge 3) and/or
    fillers. A solution from 1,1,3,3,5,5-hexakis(methacryloylethylenedioxy)cyclo
    triphosphazene (from 2-hydroxyethyl methacrylate and
    hexachlorocyclotriphosphazene) 4.00, MIBK 2.75, iso-BuOH 2.75, and
    1-hydroxycyclohexyl Ph ketone 0.12 q was coated on a polycarbonate plate
    and UV-irradiatedh to give a film showing good abrasion, weather, heat,
    and alkali resistance.
    phosphazene resin coating alkali resistance; abrasion resistance curable
ST
    phosphazene coating; heat resistance curable phosphazene coating; weather
    resistance curable phosphazene coating
IT
    Coating materials
        (abrasion- and alkali- and heat- and weather-resistant, curable
       phosphazene resins)
    7631-86-9, Silica, uses and miscellaneous
ΙT
     RL: USES (Uses)
        (colloidal, curable phosphazene resins containing, for coatings with
        abrasion, alkali, heat and weather resistance)
     139-02-6, Sodium phenolate 2923-18-4 33374-41-3, Potassium allyl
ΙT
     alcoholate 121462-09-7
     RL: USES (Uses)
        (condensation of, with halophosphazenes, for curable oligomers)
IΤ
     868-77-9
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (dehydrohalogenation of, with halophosphazenes, for curable oligomers,
        for coatings)
     940-71-6, Hexachlorocyclotriphosphazene 2950-45-0,
ΙT
     Octachlorocyclotetraphosphazene
     RL: USES (Uses)
        (dehydrohalogenation or condensation of, for curable oligomers, for
        coatings)
     291-37-2DP, 1,3,5,2,4,6-Triazatriphosphorine, allylphenoxy derivs.
IT
     7251-15-2P 92832-53-6P 115524-07-7P
     RL: RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)
        (preparation and polymerization of)
    ANSWER 63 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
    1989:425033 CAPLUS
ΑN
DN
    111:25033
    Entered STN: 21 Jul 1989
ED
    Cured phosphazines for optical decorations
ΤI
     Hosoya, Ikuo
IN
     Idemitsu Petrochemical Co., Ltd., Japan
PA
     Jpn. Kokai Tokkyo Koho, 6 pp.
SO
     CODEN: JKXXAF
DT
     Patent
LA
     Japanese
     ICM F21V005-06
IC
     ICS B44C005-08; C08J007-18; F21S001-02; F21V003-04
     42-10 (Coatings, Inks, and Related Products)
CC
FAN.CNT 1
                                                               DATE
                                         APPLICATION NO.
     PATENT NO.
                        KIND
                               DATE
                                          ______
     ______
                        ----
                               _____
     JP 63266703
                                        JP 1987-101328
                                                                19870424
                        A2
                              19881102
                     . B4
     JP 07031924
                              19950410
PRAI JP 1987-101328
                               19870424
                CLASS PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
                _____
• JP 63266703
                ICM
                       F21V005-06
                       B44C005-08; C08J007-18; F21S001-02; F21V003-04
                ICS
                       F21V0005-06 [ICM, 4]; F21V0005-00 [ICM, 4, C*];
                 IPCI
                       B44C0005-08 [ICS, 4]; B44C0005-00 [ICS, 4, C*];
                       C08J0007-18 [ICS,4]; C08J0007-00 [ICS,4,C*];
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F21S0001-02 [ICS, 4]; F21V0003-04 [ICS, 4]; F21V0003-00
                        [ICS, 4, C*]
                 IPCR
                        B44C0005-00 [I,C*]; B44C0005-08 [I,A]; C08J0007-00
                        [I,C*]; C08J0007-18 [I,A]; F21V0003-00 [I,C*];
                        F21V0003-04 [I,A]; F21V0005-00 [I,C*]; F21V0005-06
                        [I,A]
    Optical decorations such as chandeliers and stained glass contain hardened
    phosphazine surfaces. Soaking polycarbonate-made chandelier parts in a
     solution of 1,1,3,3,5,5-hexa(methacryloylethylenedioxy)cyclotriphosphazine
     (from hexachlorocyclotriphosphazene and 2-hydroxyethyl methacrylate) 4.00,
     iso-BuCOMe 2.75, iso-BuOH 2.75, and an initiator 0.12 part, heating to
     80° to remove the solvents, and irradiating with 370-nm UV gave
    products, which when assembled give a chandelier having an appearance
     comparable to a glass chandelier but having 1/5 the weight of a glass
     chandelier.
    phosphazine acrylic coating polycarbonate chandelier; stained glass
     acrylic phosphazine coating
    Lamps
        (chandeliers, polycarbonate moldings, acrylic phosphazine coatings for)
    Glass, oxide
     RL: USES (Uses)
        (stained, acrylic phosphazine coatings for)
     93891-06-6
     RL: TEM (Technical or engineered material use); USES (Uses)
        (coatings, on polycarbonate moldings, for chandeliers)
     92832-53-6P
     RL: RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)
        (preparation and polymerization of, as coatings for optical decorations)
     868-77-9, 2-Hydroxyethyl methacrylate
     RL: USES (Uses)
        (reaction with chlorocyclotriphosphazine, for coatings)
     940-71-6
     RL: USES (Uses)
        (reaction with hydroxyethyl methacrylate, for coatings)
     ANSWER 64 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
     1987:412858 CAPLUS
ΑN
DN
     107:12858
ED
     Entered STN: 11 Jul 1987
     A study on crown and bridge resin using PNC-EMA monomer
ΤI
    Anzai, Misaki; Hoya, Ken; Toriyama, Fumito; Ide, Kozo; Kikuchi, Hisaji;
ΑU
     Hirose, Hideharu; Yuda, Masashi; Ohashi, Masayoshi
     Sch. Dent., Nihon Univ., Japan
     Journal of Nihon University School of Dentistry (1986), 28(4), 240-8
SO
     CODEN: JNUDAT; ISSN: 0029-0432
DT
     Journal
     English
LA
     63-7 (Pharmaceuticals)
CC
     The properties of polymer resins for crowns and bridges using a new
AΒ
     PNC-EMA monomer [P3N3(OCH2CH2O2CCCH3:CH2)6], mixed with monomers currently
     available on the market (14-70% weight) and poly(Me methacrylate) (PMMA) as
     the polymer base, were studied. Both the compressive strength and
     hardness increased with an increasing amts. of PNC-EMA monomer. Water
     sorption reached 1.86 mg/cm2 when the amount of PNC-EMA mixed was 28%.
     degree of abrasion decreased in accordance with an increase in the amount of
     PNC-EMA showing a value one-third that of PMMA when mixed in a proportion .
     of 70%.
     PNC EMA dental bridge crown; methacryloxyethylenedioxycyclotriphosphazene
     dental polymer; cyclotriphosphazene methacryloxyethylenedioxy dental
     Dental materials and appliances
        (bridges, methacryloxyethylenedioxycyclotriphosphazene-containing polymers
        for, properties of)
ΙT
     Dental materials and appliances
        (composites, methacryloxyethylenedioxycyclotriphosphazene
        monomer-containing, properties of, for bridges and crowns)
```

AΒ

ST

ΙT

IT

IT

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TT

ΙT

CS

ST

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ΙT
     Dental materials and appliances
        (crowns, methacryloxyethylenedioxycyclotriphosphazene-containing polymers
        for, properties of)
                          3253-39-2, 2,2-Bis(4-methacryloxy)phenylpropane
ΙT
     80-62-6
              109-16-0
     RL: BIOL (Biological study)
        (dental resin for crowns and bridges containing acrylic cyclotriphosphazene
        and, properties of)
ΙT
     92832-53-6
     RL: BIOL (Biological study)
        (dental resin for crowns and bridges containing, properties of)
     36936-74-0P 92832-54-7P
ΙT
                              108704-94-5P
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (preparation and properties of, for dental bridges and crowns)
ΙT
     7732-18-5, properties
     RL: PEP (Physical, engineering or chemical process); PROC (Process)
        (sorption of, by acrylic cyclotriphosphazene monomer-containing dental
        polymers)
    ANSWER 65 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
     1986:110237 CAPLUS
AN
DN
     104:110237
     Entered STN: 05 Apr 1986
ED
    Organophosphazenes. 19. Copolymerization of 2-(\alpha-
TΙ
     ethoxyvinyl)pentafluorocyclotriphosphazene with styrene and methyl
    methacrylate
    Allen, Christopher W.; Bright, Randall P.
ΑU
     Dep. Chem., Univ. Vermont, Burlington, VT, 05405, USA
CS
    Macromolecules (1986), 19(3), 571-4
SO
    CODEN: MAMOBX; ISSN: 0024-9297
DT
     Journal
LA
     English
     35-3 (Chemistry of Synthetic High Polymers)
CC
     (\alpha-\text{Ethoxyvinyl}) pentafluorocyclotriphosphazene (I) [80297-67-2]
AB
     underwent radical copolymn. with styrene [100-42-5] and Me methacrylate.
     The styrene system was examined in detail, with flame retardant copolymers
     [80297-68-3] having ≤43.7% I content being obtained. Reactivity
     ratios for the styrene-I copolymn. were calculated by several methods. An
     examination of the Alfrey-Price parameters for I indicated that the major
     perturbation of the olefinic center was through the \alpha-electron-
     withdrawing effect of I. The copolymer mol.wts. decreased with increasing
     I content. The thermal decomposition of the copolymers was a two-step process,
     with I being involved in the first step.
ST
     vinylfluorocyclotriphosphazene styrene copolymn; methacrylate
     vinylfluorocyclotriphosphazene copolymer; reactivity ratio
     vinylfluorocyclotriphosphazene polymn; thermal degrdn
     vinylfluorocyclotriphosphazene polymer; phosphazene vinyl polymer
IT
     Fire-resistant materials
        ((ethoxyvinyl)pentafluorocyclotriphosphazene-styrene copolymer)
ΙT
     Phosphazene polymers
     RL: USES (Uses)
        (ethoxyvinylpentafluorocyclotriphosphazene-vinyl compound copolymers)
IT
     Q-e value in polymerization
     Reactivity ratio in polymerization
        (of (ethoxyvinyl)pentafluorocyclotriphosphazene with styrene)
IT
     Kinetics of polymerization
        (of (ethoxyvinyl)pentafluorocyclotriphosphazene with styrene,
        reactivity ratio in relation to)
ΙT
     Polymer degradation
        (thermal, of (ethoxyvinyl)pentafluorocyclotriphosphazene-styrene
        copolymers, mechanism of)
ΙT
     100-42-5, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (polymerization of, with (ethoxyvinyl)pentafluorocyclotriphosphazene,
        reactivity ratio and Q-e values in)
ΙT
     80297-67-2
```

```
RL: RCT (Reactant); RACT (Reactant or reagent)
        (polymerization of, with styrene, reactivity ratio and Q-e values in)
ΙT
     80297-68-3P
     RL: PEP (Physical, engineering or chemical process); SPN (Synthetic
     preparation); PREP (Preparation); PROC (Process)
        (preparation and thermal degradation of)
     99798-90-0P
IT
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (preparation of)
L9
     ANSWER 66 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
     1985:191109 CAPLUS
ΑN
       Correction of: 1984:598144
     102:191109
DN
       Correction of: 101:198144
     Entered STN: 02 Jun 1985
ED
     Synthesis of hexakis(methacryloyloxyethoxy)cyclotriphosphazene and its
TΤ
     properties and the use as a composite resin
ΑU
     Anzai, Misaki; Ohashi, Masayoshi
     Sch. Dent., Nihon Univ., Tokyo, 101, Japan
CS
     Shika Zairyo, Kikai (1984), 3(3), 401-8
SO
     CODEN: SZKIDA; ISSN: 0286-5858
DT
     Journal
     Japanese
LA
CC
     63-7 (Pharmaceuticals)
     Section cross-reference(s): 29
GI
```

TΤ

1832-07-1

```
The title compound (I) [92832-53-6] was prepared as new dental resin
AB
     monomer. Phosphonitrile chloride (II) [1832-07-1] and 2-hydroxyethyl
     methacrylate (HEMA) [868-77-9] were allowed to react at 40° to give
     I. A mixture of approx. 21% (weight) of I, 8.5% triethyleneglycol
     dimethacrylate and 70% Si3N4 were polymerized, and the polymer had a
     compressive strength, tensile strength and transverse strength of 473.6,
     33.8 and 80.4 MPa, resp., and hardness HK 96, coefficient of thermal expansion
     24.2 X 106/^{\circ} and its water sorption was 1.2 mg/cm2 at 1 wk.
     methacryloyloxyethoxy cyclotriphosphazene dental
ST
     Dental materials and fillings
ΙT
        (hexakis (methacryloyloxyethoxy) cyclotriphosphazene for)
     12033-89-5, biological studies
IT
     RL: BIOL (Biological study)
        (dental composite resin containing acrylic cyclotriphosphazene-triethylene
        glycol dimethacrylate copolymer and)
ΙT
     92832-54-7P
     RL: SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological
     study); PREP (Preparation); USES (Uses)
        (preparation of, as dental material)
TΤ
     92832-53-6P
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (preparation of, for dental resins)
TT
     868-77-9
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with hexachlorotriphosphazene)
```

RL: RCT (Reactant); RACT (Reactant or reagent)

(reaction of, with hydroxyethyl methacrylate)

L9 ANSWER 67 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN AN 1985:137742 CAPLUS

DN 102:137742

ED Entered STN: 20 Apr 1985

Studies on the physical properties of dental resin. 2. Studies on composite resins with hexa(methacryloylethylenedioxy)cyclotriphosphazene used as a monomer. Fillers and physical properties

AU Anzai, Misaki; Ohashi, Masayoshi

CS Sch. Dent., Nihon Univ., Tokyo, Japan

SO Journal of Nihon University School of Dentistry (1984), 26(3), 238-42 CODEN: JNUDAT; ISSN: 0029-0432

DT Journal

LA English

CC 63-7 (Pharmaceuticals)

GI

$$[H_2C = C (Me) CO_2CH_2CH_2O]_2P \qquad P[OCH_2CH_2O_2CC (Me) = CH_2]_2$$

$$|| \qquad \qquad || \qquad || \qquad \qquad ||$$

AB Composite resins of hexa(methacryloylethylenedioxy)cyclotriphosphazene (I) [92832-53-6] containing 70% inorg. fillers (SiO2, Si3N4, Al2O3, and their binary mixts.) are suitable as dental fillings, especially for resoration of molars. These composites set in 1.5-2 min. Si3N4 as the filler gave maximum compressive strength (473.6 MPa) and Knoop hardness (HK96) and min. coefficient of thermal expansion (24.2 + 10-6/1°). The tensile strength was maximum with SiO2 (43.4 MPa), and the transverse strength was maximum (88.3 MPa) with SiO2-Al2O3 (1:1). Water absorption was 1.2-1.4 mg/cm2 in 7 days for the composites.

ST dental composite hexamethacryloylethylenedioxycyclotriphosphazene; cyclophosphazene methacryloyl dental composite; silica filler dental composite; alumina filler dental composite; silicon nitride filler dental composite

IT Dental materials and fillings

(composites, hexa(methacryloylethylenedioxy)cyclotriphosphazene and inorg. fillers of, mech. properties and strength of)

IT 1344-28-1, biological studies 7631-86-9; biological studies 12033-89-5, biological studies

RL: BIOL (Biological study)

(dental composites containing hexa(methacryloylethylenedioxy)cyclotriphosph azene and, mech. and phys. properties of)

IT 92832-53-6

RL: BIOL (Biological study)

(dental composites containing inorg. fillers and, mech. and phys. properties of)

L9 ANSWER 68 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN

AN 1985:32326 CAPLUS

DN 102:32326

ED Entered STN: 26 Jan 1985

TI Dental sealants containing phosphazenes

PA Nihon University, Japan

SO Belg., 12 pp. CODEN: BEXXAL

DT Patent

LA French

ICI A61

```
Section cross-reference(s): 35
FAN.CNT 1
                     KIND DATE APPLICATION NO. DATE
     PATENT NO.
                     A1 19840831 BE 1984-212932
A2 19850227 JP 1983-147690
B4 19910802
A 19860401 US 1984-603648
A1 19870324 CA 1984-452920
A1 19850531 FR 1984-7420
B1 19880916
A1 19850313 GB 1984-12441
B2 19861217
A1 19850228 DE 1984-3421060
C2 19920820
A 19850301 NI 1984-1961
                                          _____
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                                                                    -----
     BE 899654
PΙ
                                                                    19840514
     JP 60038307
                                                                    19830811
     JP 03050726
     US 4579880
                                                                    19840424
     CA 1219403
                                                                    19840426
     FR 2555440
                                                                    19840514
     FR 2555440
     GB 2144754
                                                                    19840516
     GB 2144754
     DE 3421060
                                                                    19840606
     DE 3421060
NL 8401961
                        Α
                               19850301 NL 1984-1961
                                                                    19840620
PRAI JP 1983-147690 A
                               19830811
              CLASS PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
 _______
               ICI
 BE 899654
                       A61
                        A61K; C07F
                 IPCI
                 IPCR
                        C08G0079-00 [I,C*]; C08G0079-02 [I,A]
                        A61K0006-08 [ICM, 4]; A61K0006-02 [ICM, 4, C*];
 JP 60038307
                IPCI
                        C08K0005-49 [ICS,4]; C08K0005-00 [ICS,4,C*]
                        A61K0006-08 [ICM, 4]; A61K0006-02 [ICM, 4, C*]
 US 4579880
               IPCI
                        C08G0079-00 [I,C*]; C08G0079-02 [I,A]
                 IPCR
                 NCL
                        523/116.000; 260/998.110; 433/228.100; 522/142.000;
                        522/162.000; 522/171.000; 523/117.000; 526/276.000;
                        528/399.000
                        A61K0006-08 [ICM, 4]; A61K0006-02 [ICM, 4, C*]
 CA 1219403
              IPCI
                 IPCR
                        C08G0079-00 [I,C*]; C08G0079-02 [I,A]
                        A61K0006-08 [ICM, 4]; A61K0006-02 [ICM, 4, C*];
 FR 2555440
                IPCI
                        C08F0230-02 [ICS, 4]; C08F0230-00 [ICS, 4, C*]
                        C08G0079-00 [I,C*]; C08G0079-02 [I,A]
                IPCR
                        C08F0030-02 [ICM, 4]; C08F0030-00 [ICM, 4, C*];
 GB 2144754
               IPCI
                        A61K0006-08 [ICS, 4]; A61K0006-02 [ICS, 4, C*];
                        C07F0009-65 [ICA, 4]
                 IPCR
                        C08G0079-00 [I,C*]; C08G0079-02 [I,A]
 DE 3421060
               IPCI
                        C08G0079-02 [ICM, 3]; C08G0079-00 [ICM, 3, C*];
                        C08L0085-02 [ICS, 3]; C08L0085-00 [ICS, 3, C*];
                        C08K0003-00 [ICS, 3]; C08K0005-14 [ICS, 3]; C08K0005-17
                        [ICS, 3]; C08K0005-00 [ICS, 3, C*]; C08J0003-24 [ICS, 3];
                        A61K0006-08 [ICS, 3]; A61K0006-02 [ICS, 3, C*];
                        C07F0009-65 [ICS, 3]
                        C08G0079-00 [I,C*]; C08G0079-02 [I,A]
                 IPCR
                        A61K0006-08 [ICM, 4]; A61K0006-02 [ICM, 4, C*];
 NL 8401961
                 IPCI
                        C08L0043-02 [ICS, 4]; C08L0043-00 [ICS, 4, C*];
                        C08L0085-02 [ICS,4]; C08L0085-00 [ICS,4,C*]
                 IPCR
                        C08G0079-00 [I,C*]; C08G0079-02 [I,A]
     MARPAT 102:32326
OS
     Dental sealant compns. contain 10-99% polymers prepared from phosphazenes
AΒ
     containing acrylic groups and alkylene glycol dimethacrylates and silica
     fillers. Thus, 1,1,3,3,5,5-hexakis(methacryloyloxyethoxy)cyclotriphosphaz
     ene (I) [92832-53-6] was prepared by the reaction of hydroxyethyl
     methacrylate [868-77-9] with hexachlorocyclotriphosphazene [940-71-6] in
     anhydrous benzene followed by the addition of pyridine. This compound was
polymerized
     in the presence of Bz202 for 6 h at 60^{\circ} and for a further 2 h at
     120°. The properties of the resulting polyphosphazene
     [93891-06-6] are tabulated. A polymer for dental sealants was prepared from
     70 parts I, 30 parts triethylene glycol dimethacrylate and 0.3 parts Bz202
     and the properties of the polymer determined The properties of the phosphazene
     polymer such as resistance to compression stability, coefficient of thermal
```

CC

63-7 (Pharmaceuticals)

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dilation, etc., were better than those of the polymers not containing
     phosphazene groups.
     cyclotriphosphazene acrylic dental sealant
ST
IT
     Phosphazene polymers
     RL: BIOL (Biological study)
        (cyclo-, methacrylic, dental sealants containing)
IT
     Dental materials and fillings
        (sealants, methacryloylcyclotriphosphazenes)
     7631-86-9, biological studies
TΤ
     RL: BIOL (Biological study)
        (dental sealant compns. containing methacryloylcyclotriphosphazenes and)
ΙT
     92832-53-6P
     RL: RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)
        (preparation and polymerization of, for dental sealant compns.)
IT
     92832-54-7P
                   93891-06-6P
     RL: PREP (Preparation)
        (preparation of, for dental sealant compns.)
IΤ
     868-77-9
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with hexachlorocyclotriphosphazene)
IT
     940-71-6
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with hydroxyethyl methacrylate)
     ANSWER 69 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
L9
AN
     1984:598144 CAPLUS
DN
     101:198144
ED
     Entered STN: 25 Nov 1984
     Synthesis of hexakis(methacryloyloxyethoxy)cyclotriphosphazene and its
TΤ
     properties and the use as a composite resin
ΑU
     Anzai, Misaki; Ohashi, Masayoshi
CS
     Sch. Dent., Nihon Univ., Tokyo, 101, Japan
     Shika Zairyo, Kikai (1984), 59(5), 401-8
SO
     CODEN: SZKIDA; ISSN: 0286-5858
DT
     Journal
LA
     Japanese
CC
     63-7 (Pharmaceuticals)
     Section cross-reference(s): 29
GΙ
            I, R=CH2 = CMeCO2CH2CH2O
            II, R=Cl
     The title compound (I) [92832-53-6] was prepared as new dental
AΒ
     resin monomer. Phosphonitrile chloride (II) [1832-07-1] and
     2-hydroxyethyl methacrylate (HEMA) [868-77-9] were allowed to react at
     40° to give I. A mixture of approx. 21% (weight) of I, 8.5%
     triethyleneglycol dimethacrylate and 70% Si3N4 were polymerized, and the
     polymer had a compressive strength, tensile strength and transverse
     strength of 473.6, 33.8 and 80.4 MPa, resp., and hardness HK 96, coefficient of
     thermal expansion 24.2 + 106/° and its water sorption was 1.2
     mg/cm2 at 1 wk.
ST
     methacryloyloxyethoxy cyclotriphosphazene dental
TΤ
     Dental materials and fillings
        (hexakis(methacryloyloxyethoxy)cyclotriphosphazene for)
```

(dental composite resin containing acrylic cyclotriphosphazene-triethylene

IT

12033-89-5, biological studies RL: BIOL (Biological study)

```
glycol dimethacrylate copolymer and)
IΤ
    92832-54-7P
    RL: SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological
    study); PREP (Preparation); USES (Uses)
       (preparation of, as dental material)
IT
    92832-53-6P
    RL: SPN (Synthetic preparation); PREP (Preparation)
       (preparation of, for dental resins)
IT
    868-77-9
    RL: RCT (Reactant); RACT (Reactant or reagent)
       (reaction of, with hexachlorocyclotriphosphazene)
ΙT
    1832-07-1
    RL: RCT (Reactant); RACT (Reactant or reagent)
       (reaction of, with hydroxyethyl methacrylate)
L9
    ANSWER 70 OF 70 CAPLUS COPYRIGHT 2006 ACS on STN
    1973:137383 CAPLUS
AN
DN
    78:137383
    Entered STN: 12 May 1984
ED
ΤI
    Treating polymeric substrates with high-energy radiation
    Hook, Edwin O.; Nichols, Larry D.
ΙN
    Moleculon Research Corp.
PA
SO
    U.S., 8 pp.
    CODEN: USXXAM
DT
    Patent
LA
    English
IC
    B01J; C07C
INCL 204159120
    36-6 (Plastics Manufacture and Processing)
    Section cross-reference(s): 28, 43
FAN.CNT 1
                     KIND DATE APPLICATION NO. DATE
    PATENT NO.
                                       -----
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                                                               -----
    US 3711389
                      A 19730116 US 1969-841981
                                                               19690715
                      A
PRAI US 1969-841981
                            19690715
CLASS
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
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US 3711389 IC B01J; C07C
               INCL
                      204159120
              IPCI
                      B01J0001-10; B01J0001-12; C07C0125-00
              NCL
                      522/089.000; 427/513.000; 522/115.000; 522/136.000;
                      522/137.000; 522/138.000; 522/155.000; 522/157.000;
                      522/160.000; 522/161.000; 522/162.000; 522/164.000;
                      522/165.000; 525/287.000; 525/404.000; 525/445.000;
                      558/080.000; 564/013.000
    Flame, solvent, heat, and deformation resistant plastic, textile, and wood
AΒ
    products were obtained by treating the substrates with unsatd. cyclotri-or
    cyclotetraphosphazene stabilizers, e.g. tetrakis(dimethylamino)cyclotripho
    sphazene derivative, and 1-10 Mrad ionizing radation. Thus, a mixture of
    hexakis(allylamino)cyclotriphosphazene [986-11-8] 4,
    hexakis (methallyloxy) cyclotriphosphazene [39033-43-7] 2, and nylon 12
    [25038-74-8] 100 parts was extruded into a 1/16 in. rod, and irradiated in
    N to 5 Mrads with a 60Co source. The stabilized sample had a higher
    elastic modulus that phosphazene-free nylon, and self-extinguished in
    horizontal and 45.deg.(from horizontal) flame tests after consuming less
    than 1/2 in. of 3-in. samples, while phosphazene-free samples burned
    completely into an angular flame test. Cellophane samples were similarly
    stabilized using bis(allyloxy)tetrakis(dimetylamino)cyclotriphosphazene
    [39050-76-5] prepared from phosphonitrilic chloride, Me2NH, and Na allylate
    through tetrakis(dimethylamino)dichlorocyclotriphosphazene.
ST
    cyclotriphosphazene fire retardant plastic; cyclotetraphosphazene fire
    retardant plastic; radiation crosslinks stabilized plastic; nylon 12 fire
    resistance; cellophane fire resistance; birch fire resistance; wood fire
    resistance; paper fire resistance; cotton textile fire resistance;
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cyclic fire resistance
ΙT
     Fireproofing
        (agents for, cyclotri- and cyclotetraphosphazene unsatd. derivs. as)
     Gamma ray, chemical and physical effects
TΤ
        (cyclotri- and cyclotetraphosphazene unsatd. derivative- impregnated
        product treatment with, for fireproofing)
IT.
     Wood
        (fireproofing of birch by treatment with cyclotri- or
        cyclotriphosphazene unsatd. derivs. and \gamma-irradiation)
ΙT
     Textiles
        (fireproofing of cotton, by treatment with cyclotri- or
        cyclotetraphosphazene unsatd. derivs. and \gamma- irradiation)
ΤT
     Polyamides, uses and miscellaneous
     RL: USES (Uses)
        (fireproofing of, by treatment with cyclotri- or cyclotetraphosphazene
        unsatd. derivs. and \gamma-irradiation)
IT
     Cellophane
        (fireproofing of, by treatment with cyclotri- or cyclotetraphosphazene
        usatd. derivs. and \gamma- irradiation)
                39033-43-7 39050-76-5
                                          41407-14-1 41407-15-2
ΙT
     986-11-8
     41407-17-4
     RL: USES (Uses)
        (fireproofing agents)
     24937-16-4
IT
                  25038-74-8
     RL: PROC (Process)
        (fireproofing of, by treatment with cyclotri- or cyclotetraphosphazene
        unsatd. derivs. and γ-irradiation)
IT
     24937-16-4
     RL: PROC (Process)
        (fireproofing of, by treatment with cyclotri- or cyclotetraphosphiazene
        unsatd. derivs. and \gamma-irradiation)
ΙT
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (preparation and reactions of)
IT
     41407-16-3
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with acryloyl chloride)
ΙT
     107-11-9
                20907-32-8
                             41159-49-3
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with dichlorotetrakis(dimethylamino)cyclotriphosphazene)
IT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with dimethylamine)
ΙT
     814-68-6
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, with tetrakis(dimethylamino)bis[(2-
        hydroxyethyl)amino]cyclotriphosphazene)
ΙT
     141-43-5, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (with dichlorotetrakis(dimethylamino)cyclotriphosphazene)
IT
     124-40-3, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (with phosphonitrilic chloride)
=> D HIS
     (FILE 'HOME' ENTERED AT 08:07:37 ON 04 SEP 2006)
     FILE 'REGISTRY' ENTERED AT 08:07:48 ON 04 SEP 2006
L1
            492 S PHOSPHAZENE
L2
             31 S L1 AND ACRYL?
              6 S L2 AND ENE
L3
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polypropylene fire resistance; PVC radiation crosslinking; phosphazene

| L4 31 S L2 OR L3
L5 1216 S BIURET
L6 1216 S BIURET?
L7 0 S L6 AND (ACRYL ENE ALLYL) | | |
|--|---------------------|------------------|
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L8 31 S L4 | EP 2006 | |
| FILE 'CAPLUS' ENTERED AT 08:10:13 ON 04 SEP
19 70 S L8 | 2006 | |
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COST IN U.S. DOLLARS | SINCE FILE
ENTRY | TOTAL
SESSION |
| FULL ESTIMATED COST | 215.36 | 346.07 |
| DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS) | SINCE FILE
ENTRY | TOTAL
SESSION |
| CA SUBSCRIBER PRICE | -52.50 | -56.76 |

STN INTERNATIONAL LOGOFF AT 08:10:41 ON 04 SEP 2006

EAST Search History

| Ref
| Hits | Search Query | DBs | Default
Operator | Plurals | Time Stamp |
|----------|-------|---|--|---------------------|---------|------------------|
| L1 | 34 | triazatriphosphorine | US-PGPUB;
USPAT | OR | OFF | 2006/09/04 08:17 |
| L2 | 40 | \$triazatriphosphorine\$ | US-PGPUB;
USPAT | OR | OFF | 2006/09/04 08:18 |
| L3 | 58 | \$triazatriphosphorine\$ | US-PGPUB;
USPAT;
USOCR;
FPRS;
EPO; JPO;
DERWENT;
IBM_TDB | OR | OFF | 2006/09/04 08:18 |
| L4 | 10200 | \$PHOSPHAZEN\$ | US-PGPUB;
USPAT;
USOCR;
FPRS;
EPO; JPO;
DERWENT;
IBM_TDB | OR | OFF | 2006/09/04 08:20 |
| L5 | 3070 | \$PHOSPHAZEN\$ AND (photoa\$ photob\$ photoc\$ photod\$ photoe\$ photof\$ photog\$ photoh\$ photoi\$ photoo\$ photoop\$ photom\$ photon\$ photoo\$ photop\$ photoo\$ photov\$ photox\$ photox\$ photox\$ photox\$ photox\$ photoX\$ photoX\$ photoX\$ photoX\$ | US-PGPUB;
USPAT;
USOCR;
FPRS;
EPO; JPO;
DERWENT;
IBM_TDB | OR . | OFF | 2006/09/04 08:21 |
| L6 | 238 | \$PHOSPHAZEN\$ AND (photoa\$ photob\$ photoc\$ photod\$ photoe\$ photof\$ photog\$ photoh\$ photoi\$ photoi\$ photoi\$ photoo\$ photop\$ photom\$ photon\$ photoo\$ photop\$ photoq\$ photor\$ photos\$ photot\$ photov\$ photov\$ photoX\$ photoX\$ photoZ\$) AND ("430"/\$ "522"/\$).CCLS. | US-PGPUB;
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USOCR;
FPRS;
EPO; JPO;
DERWENT;
IBM_TDB | OR | OFF | 2006/09/04 08:22 |
| L7 | 11 | \$PHOSPHAZEN\$ AND (photoa\$ photob\$ photoc\$ photod\$ photoe\$ photof\$ photog\$ photoh\$ photoi\$ photoo\$ photop\$ photom\$ photon\$ photoo\$ photop\$ photov\$ photos\$ photov\$ photov\$ photox\$ phot | US-PGPUB;
USPAT;
USOCR;
FPRS;
EPO; JPO;
DERWENT;
IBM_TDB | OR | OFF | 2006/09/04 10:30 |
| L8 | 210 | 6 NOT 7 | US-PGPUB;
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| L9 | 2 | (("4634602") or ("4424252")).PN. | US-PGPUB;
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| L10 | 2 | ("20030171449").PN. | US-PGPUB;
USPAT;
DERWENT | OR | OFF | 2006/09/04 09:10 |
| L11 | 1 | 2002-549189.NRAN. | DERWENT | OR | OFF | 2006/09/04 09:14 |
| L12 | 0 | EP-19860041-\$.DID. | US-PGPUB;
USPAT;
USOCR;
FPRS;
EPO; JPO;
DERWENT;
IBM_TDB | OR | OFF | 2006/09/04 09:14 |